INTERNATIONAL **STANDARD**

First edition 1999-08-01

Imaging materials — Thermally processed silver microfilm — Specifications for stability

Matériaux pour image — Microfilm à l'argent traité thermiquement — Spécifications pour la stabilité

ISO 18919:1999(E)

Contents

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Foreword

ISO (the International Organization for Standardization) is a worldwide federation of national standards bodies (ISO member bodies). The work of preparing International Standards is normally carried out through ISO technical committees. Each member body interested in a subject for which a technical committee has been established has the right to be represented on that committee. International organizations, governmental and non-governmental, in liaison with ISO, also take part in the work. ISO collaborates closely with the International Electrotechnical Commission (IEC) on all matters of electrotechnical standardization.

Draft International Standards adopted by the technical committees are circulated to the member bodies for voting. Publication as an International Standard requires approval by at least 75 % of the member bodies casting a vote.

International Standard ISO 18919 was prepared by Technical Committee ISO/TC 42, Photography.

This International Standard is one of a series of standards dealing with the physical properties and stability of imaging materials. To facilitate identification of these International Standards, they are assigned a number within the block from 18900 to 18999 (see annex B). Copyright International Organization Copyright Internation for Standardization Provided by IHS under a material Organization or networking permitted by IHS under a material or networking permitted by IHS III is internation

Annex A forms an integral part of this International Standard. Annexes B, C, D and E are for information only.

Introduction

Thermally processed silver (TPS) films are used widely for computer-output microfilming (COM) and for document recording. This International Standard is intended to provide the desired information on the stability of IPS images as well as other relevant properties of TPS microfilms. The basic elements of the TPS imaging process are also reviewed.

The first commercial TPS imaging product for the micrographic market was a photothermographic paper, called dry silver paper. It was introduced in 1964. This paper was designed for exposure by projection and for processing with a heated drum in a combination reader-printer processor. A TPS film based on this technology, but adopted for COM recording, followed in 1968. Since then, several other TPS-type films for computer-output and sourcedocument microfilming have been introduced by several manufacturers. Special TPS products for other imaging applications have also been developed, including films for graphic arts and for duplication of aerial photographs, radiographic applications, as well as for line recording and remote sensing systems using laser beam and cathoderay tube (CRT) imaging devices. However, these special products are not covered by this International Standard. It covers only the currently available TPS microfilms based on the present state of photothermographic technology.

The unique feature of TPS microfilm and its major advantage over conventional silver-gelatin products is its onestep, dry processing method. Another notable difference is that the image-forming components and, therefore, also the final silver image are dispersed in a non-gelatin binder, primarily [poly(vinylbutyral)]. This renders them inert to moisture and its deleterious effects. The support of TPS films is normal, photographic grade PET [poly(ethylene terephthalate)] safety film ([1], [2], [3], [4], [5], [6]).

In most contemporary TPS films, the metallic silver that forms the image is contributed by light-insensitive silver behenate salts that react with an incorporated reducing agent during heat development. This reaction is catalyzed by latent image silver formed during light exposure of silver halide crystals that are also incorporated in the imaging layer. Accordingly, the reaction occurs at a much higher rate in exposed than in unexposed areas, akin to the different rate of reduction of exposed and unexposed silver halide crystals by a chemical developer in a conventional photographic system.

Two important advantages offered by the TPS process include rapid, relatively simple and convenient dry processing and inertness to oxidation of silver images. These images are relatively stable, based on behaviour under normal user and storage conditions as well as on accelerated ageing studies. ([7], [8], [9]). Since TPS films are heat-processed by raising the temperature to between 119 °C and 125 °C, which is well above any expected use and recommended storage temperatures, no chemical fixation is required. Hence, TPS films do not fall within the provisions of ISO 10602 that apply to chemical fixation.

These attractive features should be weighed against the disadvantage that, in the TPS process, the residual imageforming components are not removed during processing. Therefore, the potential for formation of excessive fog exists throughout the life of the record; such fog may render the image unusable. This may occur during dark storage at elevated temperatures, or on prolonged exposure to ambient illumination, or especially on excessive exposure to light and heat in a reader-printer or to heat generated by a nearby fire. In the case of fire, the temperature inside a "fireproof" vault or safe can also rise to cause image degradation. Concerns with these possible causes of degradation have led to the adoption of considerably lower life expectancy ratings of TPS films in these specifications than indicated by accelerated ageing studies.

This International Standard includes all the requirements for the stability of wet-processed silver-gelatin films on safety bases, set forth in ISO 10602. They also include special thermal requirements applicable to TPS films and the requirement of at least ten duplications with a high-intensity mercury vapour lamp, stipulated for diazo and vesicular films. A few other relevant requirements for thermally processed vesicular films (ISO 9718) and ammonia processed diazo films (ISO 8225) are also included. This: International Standard includes all the requirements for the safety bases, set forth in ISO 10602. They also include special the requirement of at least ten duplications with a high-intensity vesicular films. A few o

Imaging materials — Thermally processed silver microfilm — Specifications for stability

1 Scope

This International Standard establishes specifications for the stability of photographic films intended for storage of records; specifically, microfilms with a base of safety polyester [poly(ethylene terephthalate)] having predominantly silver behenate salts dispersed in nongelatinous emulsions, and thermally processed to produce a black-and-white silver image.

This International Standard applies to thermally processed silver (TPS) microfilms having ultrasonic or dielectric (induction-heated) splices. It does not cover films with splices made by means of adhesive tape.

This International Standard does not cover other types of black-and-white TPS films, black-and-white paper, colour images and colour prints that are produced with thermally processed silver behenate systems.

It does not apply to films to which lacquers have been applied.

It also does not apply to conventional black-and-white silver images that are produced by wet processing of silvergelatin films (see ISO 10602).

2 Normative references

The following standards contain provisions which, through reference in this text, constitute provisions of this International Standard. At the time of publication, the editions indicated were valid. All standards are subject to revision, and parties to agreements based on this International Standard are encouraged to investigate the possibility of applying the most recent editions of the standards indicated below. Members of IEC and ISO maintain registers of currently valid International Standards.

ISO 5-2:1991, Photography — Density measurements — Part 2: Geometric conditions for transmission density.

ISO 5-3:1995, Photography — Density measurements — Part 3: Spectral conditions.

ISO 527-3:1995, Plastics — Determination of tensile properties — Part 3: Test conditions for films and sheets.

- ISO 543:1990, Photography Photographic films Specifications for safety film.
- ISO 6077:1993, Photography Photographic films and papers Wedge test for brittleness.

ISO 7565:1993, Micrographics — Readers for transparent microforms — Measurement of characteristics.

ISO 8225:1995, Photography — Ammonia-processed diazo photographic film — Specifications for stability.

ISO 9718:1995, Photography — Processed vesicular photographic film — Specifications for stability.

ISO 10602:1995, Photography — Processed silver-gelatin type black-and-white film — Specifications for stability.

3 Terms and definitions

For the purposes of this International Standard, the following definitions apply. Copyright International Organization for Standardization **Provident International Standard**, the following definitions apply.

For the purposes of this International Standard, the following definitions apply.

Copyright In

3.1

archival medium

recording material that can be expected to retain information for ever so that it can be retrieved without significant loss when properly stored

NOTE There is, however, no such material and it is not a term to be used in International Standards or system specifications.

3.2

life expectancy

LE

length of time that information is predicted to be retrievable in a system under extended-term storage conditions

NOTE However, the actual useful life of film is very dependent upon the existing storage conditions (for example, see ISO 5466 [19] and ISO 10214) [21].

3.3

LE designation

rating for the "life expectancy" of recording materials and associated retrieval systems; the number following the LE symbol is a prediction of the minimum life expectancy, in years, for which information can be retrieved without significant loss when stored under extended-term storage conditions

NOTE For example, LE-100 indicates that information can be retrieved for at least 100 years storage.

3.4

extended-term storage conditions

storage conditions suitable for the preservation of recorded information having a permanent value

3.5

medium-term storage conditions

storage conditions suitable for the preservation of recorded information for a minimum of ten years

3.6

film base

plastic support for the emulsion and backing layers

3.7

emulsion layer(s)

image or image-forming layer(s) of photographic films, papers and plates

3.8

non-curl backing layer

layer, usually made of gelatin, applied to the side of the film base opposite to that of the emulsion layer, for the purpose of preventing curl 3.4

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NOTE 1 It is comparable to the emulsion layer in thickness and is not removed in processing.

NOTE 2 Antihalation or other layers removed are excluded from this definition.

3.9

safety photographic film

photographic film which passes the ignition time test and burning time test as specified in ISO 543

3.10

safety poly(ethylene terephthalate) base

polyester film base composed mainly of a polymer of ethylene glycol and terephthalic acid

4 Safety and hazards

4.1 Hazard warnings

Some of the chemicals specified in the test procedures are caustic, toxic or otherwise hazardous. Safe laboratory practice for the handling of chemicals requires the use of safety glasses or goggles, rubber gloves and other protective apparel such as face-masks or aprons where appropriate. Specific danger notices are given in the text for particularly dangerous materials, but normal precautions are required during the performance of any chemical procedures at all times. The first time that a hazardous material is noted in the test procedure section, the hazard is indicated by the word "DANGER" followed by a symbol consisting of angle brackets "〈〉" containing a letter which designates the specific hazard. A double bracket "〈〈〉〉" is used for particularly perilous situations. In subsequent statements involving handling of these hazardous materials, only the hazard symbol consisting of the brackets and letter(s) is displayed. Furthermore, for a given material, the hazard symbol is used only once in a single paragraph.

Detailed warnings for handling chemicals and their diluted solutions are beyond the scope of this International Standard.

Employers shall provide training and health and safety information in conformance with legal requirements.

The hazard symbol system used in this International Standard is intended to provide information to the users and is not meant for compliance with any legal requirements for labelling as these vary from country to country.

It is strongly recommended that anyone using these chemicals obtain from the manufacturer pertinent information about the hazards, handling and disposal of these chemicals.

4.2 Hazard information code system

- 〈B〉 Harmful if inhaled. Avoid breathing dust, vapour, mist or gas. Use only with adequate ventilation.
- 〈C〉 Harmful if contact occurs. Avoid contact with eyes, skin or clothing. Wash thoroughly after handling.
- 〈S〉 Harmful if swallowed. Wash thoroughly after handling. If swallowed, obtain medical attention immediately.
- 〈〈S〉〉 May be fatal if swallowed. If swallowed, obtain medical attention immediately.
- 〈F〉 Will burn. Keep away from heat, sparks and open flame. Use with adequate ventilation.

The flammable warning signal $\langle F \rangle$ shall not be used for quantities of common solvents under 1 litre.

4.3 Safety precautions

All pipette operations shall be performed with a pipette bulb or plunger pipette. This is a critical safety warning. Safety glasses shall be worn for all laboratory work.

5 Requirements for the film base

The base used for record films, as specified in this International Standard, shall be a safety polyester [i.e. poly(ethylene terephthalate)] type and can be identified by the method described in 9.1.

6 Requirements for the thermally processed silver microfilm

6.1 Safety film

The film shall meet the requirements specified in ISO 543.

6.2 Amount of free acid

The polyester base shall not have an amount of free acid greater than the equivalent of 1,0 ml of 0,1 mol/l sodium hydroxide solution per gram of film. The amount of free acid shall be measured in accordance with the procedure described in 9.3.

The volume of 0,1 mol/l sodium hydroxide equivalent to the amount of free acid of the processed film shall not increase by more than 0,5 ml over its original value after the accelerated ageing described in 9.2.

6.3 Tensile properties and loss in tensile properties

The film samples shall be processed and dried under the conditions used for the film records. Processed films shall be tested for tensile properties as described in 9.4 and shall have a tensile stress and elongation at break as specified in Table 1 (unheated film). The loss in tensile properties after accelerated ageing, as described in 9.2, shall not exceed the percentages specified in Table 1 (heated film).

Table 1 — Limits for tensile properties and loss in tensile properties on accelerated ageing of polyester-base film

7 Requirements for the emulsion and backing layers of thermally processed silver microfilm

7.1 Layer adhesion

7.1.1 Tape-stripping adhesion

The processed film shall not show any removal of the emulsion layer or backing layer, when tested as described in 9.5.

7.1.2 Humidity-cycling adhesion

The emulsion layer or backing layer of the processed film shall not show separation or cracking that can impair its intended use, when tested as described in 9.6.

7.2 Binder stability

The processed film shall not exceed a 1 mm increase in brittleness after testing as described in 9.7.

7.3 Blocking

Processed film shall show no evidence of blocking (sticking), delamination or surface damage when tested as described in 9.8. A slight sticking of the film samples that does not result in physical damage or a change in the gloss of the surface shall be acceptable. **Computer International Organization**
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The processed film shall not show any removal of the emulsion layer or backing layer, when tested as

8 Requirements for image stability

8.1 General requirements

International Standard (ISO) visual diffuse density or Status A blue diffuse transmission density shall be measured on a densitometer that has spectral conformance to ISO 5-3, and geometric conformance to ISO 5-2. In order to be classified as an LE-100 film, the processed microfilm shall meet the requirements of both the microfilm-reader test and the dark-ageing test described in 8.2 and 8.3 respectively.

8.2 Image stability: Microfilm-reader test

Status A blue diffuse transmission density patches with high-density and low-density levels (see Table 2) of the processed film shall be tested in a microfilm reader as described in 9.9. After testing for 1 h at 70 ℃, the low-density patch shall not gain more than 0,4 density units. The difference between the high-density and the low-density patches, before and after testing shall be 0,8 or greater. This test simulates the simultaneous effects of heat and light on thermally processed silver films that are discussed in annex C.

Table 2 — Limits for changes in image density and contrast retention for the microfilm reader test

1) The original density levels suggested for the low-density and high-density patches reflect a minimum contrast of at least 0,8 density units to perform this test. A Status A blue diffuse transmission density contrast of thermally processed microfilm is generally 1,6 \pm 0,1 units with a D_{low} of approximately 0,2 \pm 0,05 units.

2) The final density levels shall be determined with the same densitometer after exposure for 1 h at 70 \degree C in the microfilm reader.

8.3 Image stability: Dark-ageing test

Four microfilm specimens from the processed film samples shall be tested for dark ageing as described in 9.10, after ten additional exposures in a vesicular or diazo duplicator. After ten exposures in the duplicator, each specimen shall have an area of minimum density of ≤ 0.4 and an area of high density of ≥ 1.2 . The following criteria shall apply.

LE-100 film: Neither the minimum-density nor the high-density area shall change by more than \pm 0,1 in Status A blue diffuse transmission density units after incubation.

9 Test methods

9.1 Identification of film base

Remove all emulsion and backing layers from a sample of the unknown film by scraping. Then remove all sublayers by scraping. Prepare a sample of the base material by scuffing the surface with a suitable tool such as a razor blade. The general procedure is to move the scuffing device back and forth over the sample manually while exerting a very slight pressure. This removes the top layer of the base as a very fine dust. Carefully brush this into a mortar.

Mix the sample with about 100 times its mass of potassium bromide, previously ground to about 75 µm. Prepare a strip or pellet as described in reference [10].

Obtain an infrared absorption curve from the prepared strip or pellet by means of an infrared absorption spectrometer. By comparing the infrared absorption curve for the unknown with curves for known polymers, the identity of the unknown can be established (see reference [11]).

9.2 Accelerated ageing conditions

Processed film shall be subjected to accelerated ageing conditions to meet the requirements for increase in the amount of free acid, loss in tensile properties, and binder stability.

The test specimens shall be conditioned at 23 °C \pm 2 °C and (50 \pm 2) % relative humidity for at least 15 h. After conditioning, place the specimens in a moisture-proof envelope and heat-seal the envelope¹⁾. To prevent sticking between adjacent specimens, it may be necessary to interleave them with aluminum foil. Ensure a high ratio of film to air volume by squeezing out excess air prior to heat-sealing. Use a separate envelope for each film sample. Heat the envelopes in an oven for 72 h at 100 °C \pm 2 °C²).

An alternative method of incubating the specimens in a closed environment is by placing them in 25 mm borosilicate glass tubes. Each tube shall have two flanged sections separated by a gasket to provide a moisture seal³⁾, and shall be held together by a metal clamp. Sufficient film specimens shall be used to provide a high ratio of film-to-air volume.

NOTE In the subsequent text, samples subjected to these accelerated ageing conditions are designated "heated film". Comparison samples kept under room conditions are designated "unheated film".

Since these are thermally processed silver films, significant differences in appearance due to increase in image density will be noticed between unheated film and heated film specimens.

9.3 Determination of the amount of free acid

9.3.1 Specimen preparation

Measurements shall be made on two unheated and two heated specimens of imaged film that weigh approximately 1 g to 2 g each. Weigh the specimens to the nearest 0,01 g. Heat the films in accordance with 9.2. Remove all coatings from the film base by scraping. Cut each specimen into small pieces and accurately weigh the specimen prior to dissolving it in the appropriate solvents.

9.3.2 Solution preparation

Immerse each specimen in 30 ml of a 70/30 (m/m) mixture of purified o-cresol/chloroform (DANGER: $\langle B \rangle \langle C \rangle \langle S \rangle$).

WARNING — Chloroform is harmful if inhaled. Avoid breathing vapour, mist or gas. Use with adequate ventilation. If inhaled, move to fresh air. Contact should be avoided between chloroform and the eyes, skin or clothing. If contact occurs, obtain medical attention immediately.

^o-cresol is toxic if swallowed. Contact should be avoided between o-cresol and the eyes, skin or clothing. Wash after handling. In case of contact, flush eyes and skin thoroughly with water and obtain medical attention immediately.

The use of chloroform and o-cresol shall conform to all applicable national and local regulations.

Take care to dispose of chloroform and o-cresol in accordance with national and local regulations for hazardous waste disposal.

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¹⁾ A suitable moisture-proof envelope is a metal foil bag that is coated on the inside with polyethylene for heat-sealing.

²⁾ Incubation is accomplished in a closed environment to prevent escape of any decomposition products that may be produced during incubation. Such products may catalyse further degradation of the film base.

³⁾ A suitable inert gasket may be made from polytetrafluoroethylene.

Dissolve the polyester support by heating it at $93 \text{ °C} \pm 2 \text{ °C}$ for 30 min or until the specimen has dissolved. Precautions shall be taken to prevent excessive evaporation of the solvent. Cool the dissolved specimen to room temperature.

9.3.3 Titration

Titrate the polyester solution potentiometrically with standardized 0,1 mol/l tetrabutylammonium hydroxide using an automatic recording titrimeter and a glass/calomel electrode system. The electrodes shall have been preconditioned for 24 h in the o-cresol/chloroform solvent mixture $(\langle B \rangle \langle C \rangle \langle S \rangle)$ to prevent excessive instrumentation noise.

During titration, the burette tip shall be immersed in the solution as far as possible, and shall also be as far from the electrodes as practical. The stirring rate shall be as rapid as can be maintained without causing bubbles.

Also titrate 30 ml of a blank solution which has been heated for the same length of time as the polyester solution. Details of preparation of the standardized tetrabutylammonium hydroxide are given in annex A.

9.3.4 Calculation

The amount of free acid, *A*, expressed in equivalent millilitres of 0,1 mol/l sodium hydroxide per gram of film base, is calculated as follows for each specimen:

$$
A = \frac{c_{\mathsf{T}}(V_{\mathsf{S}} - V_{\mathsf{B}})}{0,1m}
$$

where

- $V_{\rm S}$ is the volume, in millilitres, of titrant used for the specimen;
- $V_{\rm B}$ is the volume, in millilitres, of titrant used for the blank;
- c_T is the concentration, in moles per litre, of the titrant;
- *m* is the mass, in grams, of the specimen.

Carry out the titration in duplicate on separately prepared solutions. The average amount of free acid for the unheated and heated film specimens shall be calculated and reported separately.

9.4 Tensile property strength for processed film

9.4.1 Specimen preparation

Processed film already in 16 mm format may be tested in this width. In the case of perforated 16 mm film, specimens shall be cut from the area between the perforations. Film in other sizes shall be cut into sections 15 mm to 16 mm wide and at least 150 mm long, using a sharp tool that does not nick the edges of the specimen. Five specimens are required for the unheated film and five specimens for the heated film.

The specimens to be heated and the control specimens shall be cut alternately and contiguously from a single piece of film. The thickness of each specimen shall be measured with a suitable gauge to the nearest 0,002 mm, and the width to the nearest 0,1 mm.

9.4.2 Accelerated ageing

Five specimens shall be subjected to accelerated ageing as described in 9.2.

9.4.3 Conditioning

All specimens, both unheated and heated, shall be conditioned at 23 °C \pm 2 °C and at (50 \pm 2) % relative humidity for at least 15 h. This can be accomplished by means of an air-conditioned room or an air-conditioned cabinet. The specimens shall be supported in such a way as to permit free circulation of the air around the film and the linear air velocity shall be at least 150 mm/s. Speciments International Organization for Standardization From Internation From Internation Internation Internation Internation Internation Internation Internation Internation Provided and the control Specimens shall be of

9.4.4 Procedure

The film specimens shall not be removed from the conditioning atmosphere for testing. The tensile stress and percent elongation at break shall be tested alternately as specified in ISO 527-3.

The initial grip separation shall be 100 mm and the rate of grip separation shall be 50 mm/min. The tensile stress and elongation at break shall be calculated separately for the unheated and heated film.

9.5 Tape-stripping adhesion test

9.5.1 Specimen preparation

Although the dimensions of the processed film specimen are not critical, one dimension shall be at least 150 mm. Four specimens shall be used for the emulsion surface and four specimens for the backing layer, if present.

9.5.2 Conditioning

The specimens shall be conditioned as described in 9.4.3.

9.5.3 Procedure

The film specimens shall not be removed from the conditioning atmosphere for testing. Apply a strip of pressuresensitive, plastic-base adhesive tape about 150 mm long to the surface of the processed film. Press the tape down with thumb pressure to ensure adequate contact, leaving enough tape at one end to grasp. No portion of the tape shall extend to the edges of the film specimens or extend to film perforations. Hold the specimen firmly on a flat surface and remove the tape rapidly from the film surface. This shall be accomplished by peeling the tape back on itself and pulling the end so that it is removed from the film at an angle of approximately 180∞. Removal by the tape of any portion of the surface layer on any of the specimens shall be considered failure.

The results of the tape-stripping test may be very dependent upon the adhesive tape used if the bonding force between it and the particular film surface under test is not sufficiently high. For this reason, a minimum bonding force is specified for this test. This bonding force shall be determined by applying the adhesive tape to the film surface in the same manner as described in the tape-stripping test. The tape shall be rapidly peeled back from the film surface at an angle of approximately 180∞. The peelback force required to separate the tape from the film shall be measured by a suitable device such as a strain gauge or spring scale capable of reading the maximum force used. A bonding force of at least 0,9 N per millimetre of tape width is required.

9.6 Humidity-cycling adhesion test

9.6.1 Specimen preparation

Two specimens of processed film shall be selected from an area of high density. The preferred specimen size is 50 mm \times 50 mm, or 50 mm \times the film width where the size of the film permits, but the dimensions are not critical, provided all specimens are of uniform size.

9.6.2 Procedure

Mount the test specimens in a specimen rack and place the rack in a glass laboratory desiccator jar in such a way that the specimens are freely exposed to the required conditioning atmosphere. Place the jar in a forced-air circulating oven for 8 h at 50 °C \pm 2 °C. The atmosphere within the jar shall be maintained at 96 % relative humidity, which can be obtained by keeping a saturated solution of potassium sulfate in water (see references [12], [13]) in the bottom of the jar⁴⁾. Ensure that the saturated solution contains an excess of undissolved crystals at 50 °C. The undissolved crystals shall be completely covered by a layer of saturated salt solution and the surface area of the solution should be as large as practical. The jar and salt solution shall be kept at 50 °C \pm 2 °C for at least 20 h prior to use to ensure attainment of equilibrium.

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⁴⁾ The relative humidity is based on the normal vapour pressure of the salt solution but the relative humidity tolerance cannot be specified.

After 8 h, place the specimens and specimen rack for 16 h in a second desiccator jar that is also in the same oven. The atmosphere within the second jar shall be maintained at 11 % relative humidity, which can be obtained by keeping a saturated solution of lithium chloride in water (see references [12], [13]) in the bottom of the jar4).

Time periods of 8 h at the high humidity and 16 h at the low humidity shall constitute one cycle⁵⁾. Each film specimen shall be subjected to 12 humidity cycles. After this, remove the film specimens from the specimen rack and examine the emulsion and any backing layer for any evidence of peeling, flaking, or cracking produced as a result of the humidity-cycling treatment.

NOTE Films may sometimes exhibit what appear to be small pinholes in the image after processing. These can be caused by dirt or dust particles on the emulsion surface at the time the raw film is exposed and should not be confused with holes or cracks in the emulsion layer. The existence of such clear spots in the image prior to humidity cycling should be noted so that their presence does not lead to a false interpretation of adhesion weakness.

The film shall be examined under the magnification and lighting conditions that are normal for the end use of the product. During an interruption in the cycling procedure, the film specimens shall be kept at 50 °C \pm 2 °C and 11 % relative humidity.

9.7 Binder stability test

Wedge brittleness measurements shall be made on five unheated and five heated specimens of processed film. Each specimen shall preferably contain a low-density area. Although the dimensions of the processed film specimen are not critical, one dimension shall be at least 150 mm. Five film specimens shall be subjected to accelerated ageing as described in 9.2.

9.7.1 Conditioning

Both the heated and unheated specimens shall be conditioned as described in 9.4.3.

9.7.2 Procedure

The film specimens shall not be removed from the conditioning atmosphere for testing. The wedge brittleness of the unheated and heated specimens shall be measured as specified in ISO 6077.

9.8 Blocking test

At least five specimens of processed film shall be conditioned at 62 % relative humidity and 40 °C ± 2 °C. The preferred specimen size is 50 mm \times 50 mm but the dimensions are not critical provided all specimens are of uniform size. Place the specimens in a glass laboratory desiccator jar so that they are freely exposed to the required conditioning atmosphere for at least 15 h. Place the jar containing the specimens in a forced-air circulating oven at 40 °C ± 2 °C. A relative humidity of approximately 62 % can be obtained by keeping a saturated solution of sodium nitrite (see references [12], [13]) in water at the bottom of the jar⁴⁾. Ensure that the saturated solution contains an excess of undissolved crystals at 40 ℃. The undissolved crystals shall be completely covered by a layer of saturated salt solution and the surface area of the solution should be as large as is practical. The jar and salt solution shall be kept at 40 $\mathrm{°C}$ ± 2 $\mathrm{°C}$ for at least 20 h prior to use to ensure adequate equilibrium.

After moisture equilibrium is attained, remove the jar from the oven. Without removing the film specimens from the jar, stack at least five film specimens so that the emulsion surface of one specimen is against the back surface of the adjacent specimen. Place the stack under a uniform pressure of 35 kPa. This can be accomplished by placing a weight on the film stack, the dimensions of the weight being greater than those of the film specimens. The jar containing the weighted stack shall be put back into the forced-air circulating oven for 3 days at 40 °C \pm 2 °C.

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⁴⁾ The relative humidity is based on the normal vapour pressure of the salt solution but the relative humidity tolerance cannot be specified.

⁵) This can be most easily accomplished by placing the specimens in the 96 % relative humidity jar in the morning and in the

^{11 %} relative humidity jar in the evening.

Alternatively, the temperature and humidity conditions can be achieved by means of air-conditioned cabinets, or airconditioned rooms.

Remove the film stack from the oven and allow it to cool. Individually remove the film specimens from the stack and observe each for evidence of film blocking (sticking) (see 7.3).

9.9 Image stability: Microfilm-reader test

9.9.1 Sample preparation

Specimens of film shall be exposed to the specified densities or images as indicated in Table 2. Appropriate specimen sizes are 105 mm \times 148 mm for microfiche, 50 mm \times 50 mm for sheet film products, or 50 mm \times the film width for roll film products. Two specimens of low and high density areas shall be tested in a microfilm reader equipped with a 150 W projection lamp⁶⁾ kept in a conditioned room at 23 °C ± 2 °C and (50 ± 2) % relative humidity.

9.9.2 Conditioning

The film specimens shall be conditioned at 23 °C \pm 2 °C and (50 \pm 2) % relative humidity for at least 8 h. This can be accomplished by means of an air-conditioned room or an air-conditioned cabinet. The specimens shall be supported in such a way as to permit free circulation of air around the film, and the linear air velocity shall be at least 150 mm/s.

9.9.3 Film-gate temperature measurements

Operate the microfilm reader for at least 1 h at an ambient temperature of 23 °C ± 2 °C prior to performing any tests. Place a test strip of processed silver-gelatin microfilm, having a minimum density of 1,50, in the microfilm reader. The film-gate temperature at the surface of a silver-gelatin microfilm test strip shall read 70 ℃ as described in ISO 7565. The temperature shall be controlled at ± 2 °C.

9.9.4 Procedure

Insert the TPS film specimens containing patches of low density and high density in the microfilm reader for 1 h with the film-gate temperature maintained at 70 ∞C. Measure the density changes.

9.9.5 Measurement

TPS film specimens shall be measured for Status A blue diffuse transmission density prior to the microfilm-reader test (original) and after 1 h in the microfilm reader (final). Density patches shall be measured on a densitometer having spectral conformance to ISO 5-3, and geometric conformance to ISO 5-2. The sample shall meet the density criteria given in 8.2.

9.10 Image stability: Dark-ageing test

9.10.1 Specimen preparation

Four specimens of TPS film shall be exposed and processed to the D_{low} and D_{high} densities indicated in Table 3. Appropriate specimen sizes are 105 mm \times 148 mm (microfiche), 50 mm \times 50 mm (sheet film products), and 50 mm \times width of film (roll film products). The four specimens shall receive an additional ten exposures each in a vesicular or diazo duplicator. These exposures shall be equivalent to standard exposures for film duplication. The vesicular and diazo duplicators shall be equipped with a 1 200 W to 2 000 W mercury vapour lamp in the exposure station. **Conditioning**
 Complished by means of an air-conditioned arc 23 °C ± 2 °C and (50

be accomplished by means of an air-conditioned room or an a

150 mm/s.
 9.9.3 Film-gate temperature measurements
 Operate the microf

In addition to the four specimens, at least 14 processed TPS films or fiche of the same kind shall be prepared to be used as fillers for heat-sealed bag incubations. The density readings of these fillers are not critical.

-

⁶) Generally the microfilm readers are equipped with tungsten or tungsten-halogen lamp sources.

9.10.2 Specimen incubation

Prior to the incubation, condition the four specimens and the filler TPS film samples at 23 °C \pm 2 °C and (50 \pm 2) % relative humidity for at least 8 h as outlined in 9.9.2. Two film specimens for each of the two incubation conditions shall be sandwiched between at least seven film samples of the same kind and then be heat-sealed in a moistureproof metallic-foil envelope after the air has been squeezed out. Placing the specimens in two bags is recommended to reduce any effect of pinholes.

9.10.3 Incubation conditions

Two heat-sealed bags containing the film specimens shall be incubated at each of the temperatures and times specified in Table 3. Temperature shall be controlled to ± 2 °C. A description of how these incubation conditions were chosen and the time-temperature relationship for dark ageing ([7], [8], [9], [14]) are discussed in annex D.

9.10.4 Measurement

Status A blue diffuse transmission density of the TPS specimens shall be measured both before and after incubation. Densities shall be measured on a densitometer having spectral conformance to ISO 5-3, and geometric conformance to ISO 5-2. The specimens shall meet the density criteria given in 8.3.

10 Storage of films

Everyone concerned with the preservation of records on photographic film should realize that specifying the chemical and physical characteristics of the material does not, by itself, ensure that the records will not deteriorate. It is also essential to provide the correct storage temperature and humidity, and protection from the hazards of fire, water, fungus, and certain atmospheric pollutants. Films shall be stored under the conditions specified in pertinent International Standards, for example, ISO 5466 [19] and ISO 10214 [21]. recommended to reduce any enerot of pimoles.

9.10.3 Incubation conditions for Standardization For Standardization Provided by INS under the Standardization Provided by IHS under the Standardization Provident International

Annex A

(normative)

Preparation of standard solution of tetrabutylammonium hydroxide

Polyester bases inherently have carboxyl groups incorporated in their molecular structure. In chemical terminology, this is referred to as "free acids" and is expressed as the volume of a standard solution of sodium hydroxide required to neutralize 1 g of polyester base. The intent of the test that is specified in 6.2 is to ensure that the base is chemically stable and also that it will not be adversely affected by the image-bearing layer or other layers. Thus, the entire film shall be heated in accordance with 9.2 to simulate long-time ageing. However, it is necessary to remove all the layers from the base prior to determining the acidity to simplify the analytical procedure.

Use a primary 0,1 mol/l standard solution of benzoic acid to standardize a tetrabutylammonium hydroxide (TBAH) titrant. Then titrate the TBAH against the polyester base to determine the amount of free acid. Prepare the nominal 0,1 mol/l primary standard solution of benzoic acid by weighing (1.22 ± 0.01) g of benzoic acid into a 100 ml volumetric flask. Make up to the mark with a 70/30 (m/m) mixture of purified o-cresol/chloroform (DANGER: $\langle B \rangle \langle C \rangle$ $\langle S \rangle$).

WARNING — Chloroform is harmful if inhaled. Avoid breathing vapour, mist, or gas. Use with adequate ventilation. If inhaled, move to fresh air. Contact should be avoided between chloroform and the eyes, skin or clothing. If contact occurs, obtain medical attention immediately.

^o-cresol is toxic if swallowed. Contact should be avoided between o-cresol and the eyes, skin or clothing. Wash after handling. In case of contact, flush eyes and skin thoroughly with water and obtain medical attention immediately.

The use of chloroform and o-cresol shall conform to all applicable national and local regulations.

Take care to dispose of chloroform and o-cresol in accordance with national and local regulations for hazardous waste disposal.

The concentration, c_p , in moles per litre, of the primary benzoic acid standard is calculated from:

$$
c_{\rm p} = \frac{0.01 m_{\rm B}}{1.22}
$$

where m_B is the mass, in grams, of benzoic acid used.

Then use the primary standard to determine the molarity of the TBAH base titrant.

Prepare a 25 % (m/m) solution of TBAH by dissolving 25 g of TBAH in 75 ml of water. Prepare a nominal 0,1 mol/l TBAH titrant solution by adding 90 ml of isopropanol to 10 ml of the 25 % (m/m) aqueous TBAH solution. Standardize this against the primary benzoic acid solution by pipetting exactly 1 ml of the 0,1 mol/l benzoic acid solution into a 50 ml beaker and adding 30 ml of a 70/30 (m/m) mixture of o-cresol/chloroform ($\langle B \rangle$ $\langle C \rangle$ $\langle S \rangle$). Titrate this solution with TBAH base titrant using an automatic-recording titrimeter in the specified manner. Carry out a similar titration with a blank 30 ml sample of the o-cresol/chloroform mixture. Make both titrations in duplicate.

The concentration, c_T in moles per litre, of the TBAH base titrant, is calculated from:

$$
c_{\mathsf{T}} = \frac{V_{\mathsf{p}} \cdot c_{\mathsf{p}}}{V_{\mathsf{S}} - V_{\mathsf{B}}}
$$

where

 V_p is the volume, in millilitres, of the 0,1 mol/l benzoic acid solution; $V_S - V_B$

Where V_p is the volume, in millilitres, of the 0,1 mol/l benzoic acid solution;

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- c_p is the concentration, in moles per litre, of the 0,1 mol/l benzoic acid solution;
- V_S is the volume, in millilitres, of TBAH titrant used in the titration of the primary standard;
- *V*_B is the volume, in millilitres, of TBAH titrant used in the titration of the blank.

EXAMPLE

If 1,00 ml of 0,102 mol/l primary standard is titrated with 1,147 ml of TBAH base titrant and 30 ml of blank is titrated with 0,032 ml of TBAH base titrant, the concentration of TBAH is

 $c_{\text{T}} = \frac{1,00 \times 0,102}{1,147 - 0,032} = 0,091 \text{ mol/II}$

Annex B

(informative)

Numbering system for related International Standards

The current numbering system for TC 42 documents dealing with the physical properties and stability of imaging materials is confusing since the five digit numbers that are used are not in any consecutive order. To facilitate remembering the numbers, ISO has set aside a block of numbers from 18900 to 18999 and all revisions and new International Standards will be given a number within this block. The last three digits will be identical to the current ANSI/PIMA numbers of published documents. This will be advantageous to the technical experts from Germany, Japan, United Kingdom and the USA who have prepared the standard and who are familiar with the ANSI/PIMA numbers.

As the current International Standards are revised and published, their new numbers will be as given in Table B.1.

Table B.1 — New ISO numbers

Annex C

(informative)

Effect of residual compounds on the thermally processed silver image

The thermal processing of TPS films does not include removal of substances that can induce fog (non-image silver) during subsequent use and storage. However, fog formation requires thermal activation. Moreover, the degradation potential of the fog-inducing compounds diminishes slowly under ambient storage conditions, because they sublime and diffuse out of the image layer. Evidence of such diffusion and the influence of storage conditions was provided by the differences in the amount of fog-inducing activators detected by chemical analysis of samples incubated in tightly packed, heat-sealed bags and those incubated in a free-hanging mode. The former contained higher residual amounts of activator compounds than the latter.

In practice, TPS films usually are stored in tightly wound rolls or in enclosed stacks of microfiche. Therefore, diffusion will be restricted and require considerable time. This condition is simulated more closely by the sealed-bag than the free-hanging incubation test method. It is for this reason that the sealed-bag method is specified in this International Standard ([7], [8], [9]).

Another requirement of the accelerated ageing test of TPS films is the multiple exposure of the test specimens in a commercial duplicator prior to incubation. This is intended to simulate the effects of duplication of TPS films for at least ten times during their life span, with a vesicular or diazo duplicator and a high-intensity mercury vapour lamp.

Finally, the microfilm reader test specified in 8.2 has been included to simulate exposures of TPS films left inadvertently in a reader. The specified 1 h exposure in a microfilm reader at a film gate temperature of 70 \degree C is intended to simulate the effects of inadvertent, simultaneous exposure to excessive light and heat to TPS images.

Annex D

(informative)

Accelerated image stability test for thermally processed silver microfilms

D.1 Introduction

The experimental method used to derive the predicted lifetimes of TPS films at ambient conditions of storage is based on the data that appear in [7], [8] and [9]. In order to extrapolate the high temperature incubation results to normal keeping conditions, the data were treated by classical chemical kinetics as described in references [1], [2] and [6]. It is recognized that the rate-controlling step in image degradation of photographic films may be much more complex than a first order approximation. Nevertheless, Arrhenius plots provided a reasonable means of extrapolation to predict lifetimes of several imaging media at room temperature conditions ([15], [16], [17]).

D.2 Specimen preparation, density measurements, and incubation conditions

A special target containing standard alphanumeric characters, resolution charts, and densitometric step tablets was prepared ([18], [20]). This target was used to duplicate the information onto TPS films from two different commercial manufacturers by exposing in a sensitometric unit and processing according to the manufacturer's recommended temperatures. The step tablets on the TPS film specimens provided the means of measuring macro-densities. These TPS specimens were given four additional exposures in a vesicular duplicator containing high intensity mercury-vapour lamps.

Duplicate TPS film specimens were sandwiched between seven sheets of the same film after equilibration to 50 % RH at 21 ℃. This sandwich was wrapped with aluminium foil and placed inside a laminated (aluminium foilpolyethylene lining) heat-sealable envelope to prevent escape of moisture during high temperature incubation. In order to protect these envelopes from puncture during incubation, the sealed envelope was placed inside another sealed envelope. This technique simulates a confined atmosphere during incubation, which is the most severe situation for image degradation (see annex B). All the volatiles from the film and the thermal degradation products of the binder, developer, developer accelerator and undeveloped silver salts and the residual solvents are contained, possibly accelerating image degradation. The film samples were incubated at seven temperatures between 32 ∞C and 70 ∞C at 50 % RH for up to 700 days and were periodically withdrawn for density measurements. D.1 Introduction
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A densitometer was used in the visual density and Status A blue diffuse transmission density regions to monitor the density changes before and after incubation as a function of time. Microdensity in the visual and Status A blue regions was also measured on selected alphanumeric characters to further study the effects of incubation. Changes in *D_{min}* and *D_{max}* character density (i.e. around 1,8 to 2,0 Status A blue diffuse transmission density) and contrast (character density minus D_{min}) were determined as a function of time and incubation conditions. Visual density changes affect the readability of the information in a microfilm reader and the Status A blue diffuse transmission density changes affect duplication onto vesicular or diazo microfilms. The time required for a 0,1 density increase was determined at each incubation condition.

Selected incubated specimens were duplicated on vesicular microfilms to evaluate the effects of the reduced contrast on the image quality of the duplicates. In addition, non-incubated and incubated specimens were analysed chemically, and by transmission electron microscopy, to determine the presence of residual developer accelerator and the spectral absorption characteristics.

D.3 Results — Arrhenius estimations

All estimated life values in years, under 21 \degree C and 50 % RH storage conditions, are based on a modified Arrhenius plot in which the logarithm of the time required for a 0,1 Status A blue diffuse transmission density change at each incubation temperature is plotted against the reciprocal of the absolute temperature. Typical results obtained with two TPS films of two manufacturers are shown in Figure D.1.

The Arrhenius plots in Figure D.1 showed the product similarities between the two manufactures of TPS films. These results were further verified by round-robin studies at three independent laboratories on TPS film specimens after ten post-UV exposures, in which incubations were carried out at 50 % RH and temperatures of 45 ∞C and 55 ∞C up to 100 days, which confirmed their responses.

Extrapolation of the high temperature data to a room temperature of 21 ∞C in Figure D.1 indicated that the time for a 0,1 density gain at 50 % relative humidity exceeded 300 years for the Sample-A film and over 1 000 years for the Sample-B film. Although these data extrapolated to a higher life expectancy, a conservative LE-100 rating was assigned to TPS films.

The incubation conditions specified in Table 3 for LE-100 film were obtained by constructing a line parallel to the Arrhenius plots but starting at 100 years at 21 ∞C. The times for incubation temperatures of 40 ∞C and 50 ∞C fall on this line.

Annex E

(informative)

Bibliography

- [1] ADELSTEIN, P.Z. and McCREA, J.L. Stability of processed polyester base photographic films. Journal of Applied Photographic Engineering, **7**(6), December 1981, pp. 160-167.
- [2] ADELSTEIN, P.Z. and MCCREA, J.L. Permanence of processed Estar polyester base photographic films. Photographic Science and Engineering, September 1965, pp. 305-313.
- [3] CENTA, J.M. Performance characteristics for Cronar polyester photographic film base. Photogrammetric Engineering, **21**, September 1955, pp. 539-542.
- [4] WHITE, D.R., CASS, C.J., MESCHTER, E and HOLM, W.R. Polyester photographic film base. Journal of the Society of Motion Picture and Television Engineers, **64**, December 1955, pp. 674-678.
- [5] BROWN, D.W., LOWRY, R.E. and SMITH, L.E. Prediction of the Long Term Stability of Polyester Recording Media. NBSIR 86-3474, June 1986, US National Institute of Standards and Technology, Gaithersbrrg, MD 20899, USA.
- [6] RAM, A.T. Archival preservation of photographic films A perspective. Polymer Degradation and Stability, **8**, January 1990, pp. 3-29.
- [7] RAM, A.T., McCREA, J.L. and Snell, R. Image Stability of Processed Kodak Dacomatic DL Film. Journal of Imaging Technology, **15** (4), August 1989, pp. 169-177.
- [8] KURTTILA, K.R. Dry Silver Film Stability. Journal of Micrographics, **10**, January 1977, pp. 113-117.
- [9] SHEPARD, J.W. Early Dry Silver Technology at 3M. Journal of Applied Photographic Engineering, **8**, October 1982, pp. 210-212.
- [10] ANDERSON, D.H. and WOODALL, N.B. Infrared identification of materials in the fractional milligram range. Analytical Chemistry, **25**, December 1953, pp. 1906-1909.
- [11] Infrared Spectra of Plastics and Resins, P.B. 111438, 1954. US Department of Commerce, Office of Technical Services, Washington, DC, USA.
- [12] WEXLER, A. and HASEGAWA, S. Relative humidity-temperature relationships of some saturated salt solutions in the temperature range 0 to 50 ℃. Research Paper RP2512. Journal of Research of the National Bureau of Standards, **53** (1), July 1954, pp. 19-26. Copyright International Organization For Standardization for Standardization For Standardization For Standardization Providence and Standardization For Standardization Providence and Standardization For Standardization Pr
	- [13] CARR, D.S. and HARRIS, B.L. Solutions for maintaining constant relative humidity. Journal of Industrial Engineering Chemistry, **41**, September 1949, pp. 2014-2015.
	- [14] Thermally-Processed Silver Microfilm, NMA-1981. Technical Report of National Micrographics Association. Association for Information and Image Management, 1100 Wayne Avenue, Silver Spring, MD 20910, USA.
	- [15] KOPPERL, D.F. and HUTTEMANN, T.J. Effect of residual thiosulfate ion on the image stability of microfilms. Journal of Imaging Technology, **12**, August 1986. pp. 173-180.
	- [16] KOPPERL, D.F., LARSON, G.W., HUTCHINS, B.A. and BARD, C.C. A method to predict the effect of residual thiosulfate content on the long-term image stability characteristics of radiographic films. Journal of Applied Photographic Engineering, **8**, April 1982, pp. 83-89.
- [17] KOPPERL, D.F., GOETTING, R.R., HUTTEMANN, T.J. and SCHICKLING, M.L. Use of Arrhenius testing to evaluate the thiosulfate tolerance of black-and-white aerial film. Journal of Imaging Science and Technology, **36**, January/February 1992, pp. 42-45.
- [18] Practice for operational procedures/inspection and quality control of first-generation silver-gelatin microfilm of documents, MS23-1991. Association for Information and Image Management, 1100 Wayrie Avenue, Silver Spring, MD 20910, USA.
- [19] ISO 5466:1996, Photography Processed safety photographic films Storage practices.
- [20] ISO 8514-2:1992, Micrographics Alphanumeric computer output microforms Quality control Part 2: Method.
- [21] ISO 10214:1991, Photography Processed photographic materials Filing enclosures for storage.

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