

# INTERNATIONAL STANDARD

# ISO 14565

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## **Animal feeding stuffs — Determination of vitamin A content — Method using high- performance liquid chromatography**

*Aliments des animaux — Détermination de la teneur en vitamine A —  
Méthode par chromatographie liquide à haute performance*



Reference number  
ISO 14565:2000(E)

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## Foreword

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International Standards are drafted in accordance with the rules given in the ISO/IEC Directives, Part 3.

Draft International Standards adopted by the technical committees are circulated to the member bodies for voting. Publication as an International Standard requires approval by at least 75 % of the member bodies casting a vote.

Attention is drawn to the possibility that some of the elements of this International Standard may be the subject of patent rights. ISO shall not be held responsible for identifying any or all such patent rights.

International Standard ISO 14565 was prepared by Technical Committee ISO/TC 34, *Food products*, Subcommittee SC 10, *Animal feeding stuffs*.

Annex A of this International Standard is for information only.



# Animal feeding stuffs — Determination of vitamin A content — Method using high-performance liquid chromatography

## 1 Scope

This International Standard specifies a method for the determination of the total vitamin A (retinol) content of animal feeding stuffs and pet foods using high-performance liquid chromatography. The vitamin A content is the content of all-*trans*-retinyl alcohol and *cis*-isomers determined by the method described in this International Standard, and is expressed in International Units per kilogram (IU/kg).

## 2 Normative references

The following normative documents contain provisions which, through reference in this text, constitute provisions of this International Standard. For dated references, subsequent amendments to, or revisions of, any of these publications do not apply. However, parties to agreements based on this International Standard are encouraged to investigate the possibility of applying the most recent editions of the normative documents indicated below. For undated references, the latest edition of the normative documents referred to applies. Members of ISO and IEC maintain registers of currently valid International Standards.

ISO 3696:1987, *Water for analytical laboratory use — Specification and test methods*.

ISO 6498, *Animal feeding stuffs — Preparation of test samples*.

## 3 Term and definition

For the purposes of this International Standard, the following term and definition apply.

### 3.1

#### **vitamin A content**

content of all-*trans*-retinyl alcohol and *cis*-isomers determined in accordance with this International Standard

NOTE The vitamin A content is expressed in International Units per kilogram (IU/kg); 1 IU of vitamin A is equal to 0,300 µg of all-*trans*-retinol.

## 4 Principle

The sample is saponified with ethanolic potassium hydroxide solution and the vitamin A is extracted into light petroleum. The light petroleum is removed by evaporation and the residue is dissolved in 2-propanol. The vitamin A concentration in the 2-propanol extract is determined by reverse-phase liquid chromatography using conditions that give a single peak for all retinol isomers.

## 5 Reagents

Use only reagents of recognized analytical grade.

**5.1 Water**, complying with at least grade 3 in accordance with ISO 3696.

**5.2 Potassium hydroxide solution (KOH)**.

Dissolve 500 g of potassium hydroxide in water (4.1) and dilute to 1 litre.

**5.3 Ethanol**,  $w(\text{C}_2\text{H}_5\text{OH}) = 95\%$  (by volume), or equivalent industrial methylated spirit.

**5.4 2-Propanol** ( $\text{C}_3\text{H}_7\text{OH}$ ).

**5.5 Light petroleum**, boiling range 40 °C to 60 °C.

The residue on evaporation shall be less than 20 mg/l.

**5.6 Vitamin A standard substances**

**5.6.1 All-trans-retinyl acetate**, vitamin A acetate ( $\text{C}_{22}\text{H}_{32}\text{O}_2$ ), 328.5 g/mol, with a purity of at least 90 %.

**5.6.2 All-trans-retinol**, vitamin A alcohol ( $\text{C}_{20}\text{H}_{30}\text{O}$ ), 286,5 g/mol, with a purity of at least 90 %

**5.7 Methanol**, HPLC grade.

**5.8 Mobile phase for liquid chromatography**.

Mix together methanol (5.7) and water (4.1) in the proportions 770 + 30 (by volume).

If necessary, filter through a membrane filter (6.6).

**5.9 Sodium sulfate** ( $\text{Na}_2\text{SO}_4$ ), anhydrous.

**5.10 Sodium ascorbate solution**,  $\rho = 100$  g/l.

**5.11 Inert gas**, e.g. nitrogen.

## 6 Apparatus

Usual laboratory apparatus and, in particular, the following.

**6.1 High-performance liquid chromatograph**, consisting of the following.

**6.1.1 Pump**, set to deliver a constant eluent volume flow rate of 1 ml/min.

**6.1.2 HPLC injection device**.

**6.1.3 Column**, length 250 mm, internal diameter 4,6 mm, packed with a stationary phase consisting of octadecyl ( $\text{C}_{18}$ ) groups bonded to silica.

A column with at least 4 000 theoretical plates and a  $k'$  value of 0,6, both with respect to all-*trans*-retinol, has been found to be satisfactory. The particle size shall be not smaller than 5  $\mu\text{m}$  and not greater than 10  $\mu\text{m}$ . Other systems may be used provided that a satisfactory separation of vitamin A from other co-extractives is achieved.

**6.1.4 Detector**, allowing the measurement of ultraviolet radiation at 325 nm, and equipped with integrator/data-handling system.

**6.2 UV (or UV-Visible) spectrometer**, capable of measuring absorbance at the wavelengths defined in 9.6, equipped with quartz cells of 10 mm path length.

**6.3 Boiling water bath.**

**6.4 Rotary vacuum evaporator**, with water bath at 40 °C.

**6.5 Extraction apparatus** (see Figure 1) consisting of the following:

- a cylinder of 1 litre capacity fitted with a ground glass neck and stopper;
- a ground glass joint, fitting the cylinder and equipped with an adjustable tube passing through the centre; and
- a side-arm.

The adjustable tube should have a U-shaped lower end and a jet at the opposite end so that the upper liquid layer in the cylinder may be transferred to a separating funnel of 1 litre capacity.

Other extraction equipment such as conical flasks and separating funnels may be used in place of the apparatus shown in Figure 1, provided that satisfactory recoveries of vitamin A are achieved.

**6.6 Membrane filter**, 0,45 µm pore size, for filtration of mobile phase (5.8) and sample test solutions.

## 7 Sampling

It is important that the laboratory receive a sample which is truly representative and has not been damaged or changed during transport or storage.

Sampling is not part of the method specified in this International Standard. A recommended sampling method is given in ISO 6497 [4].

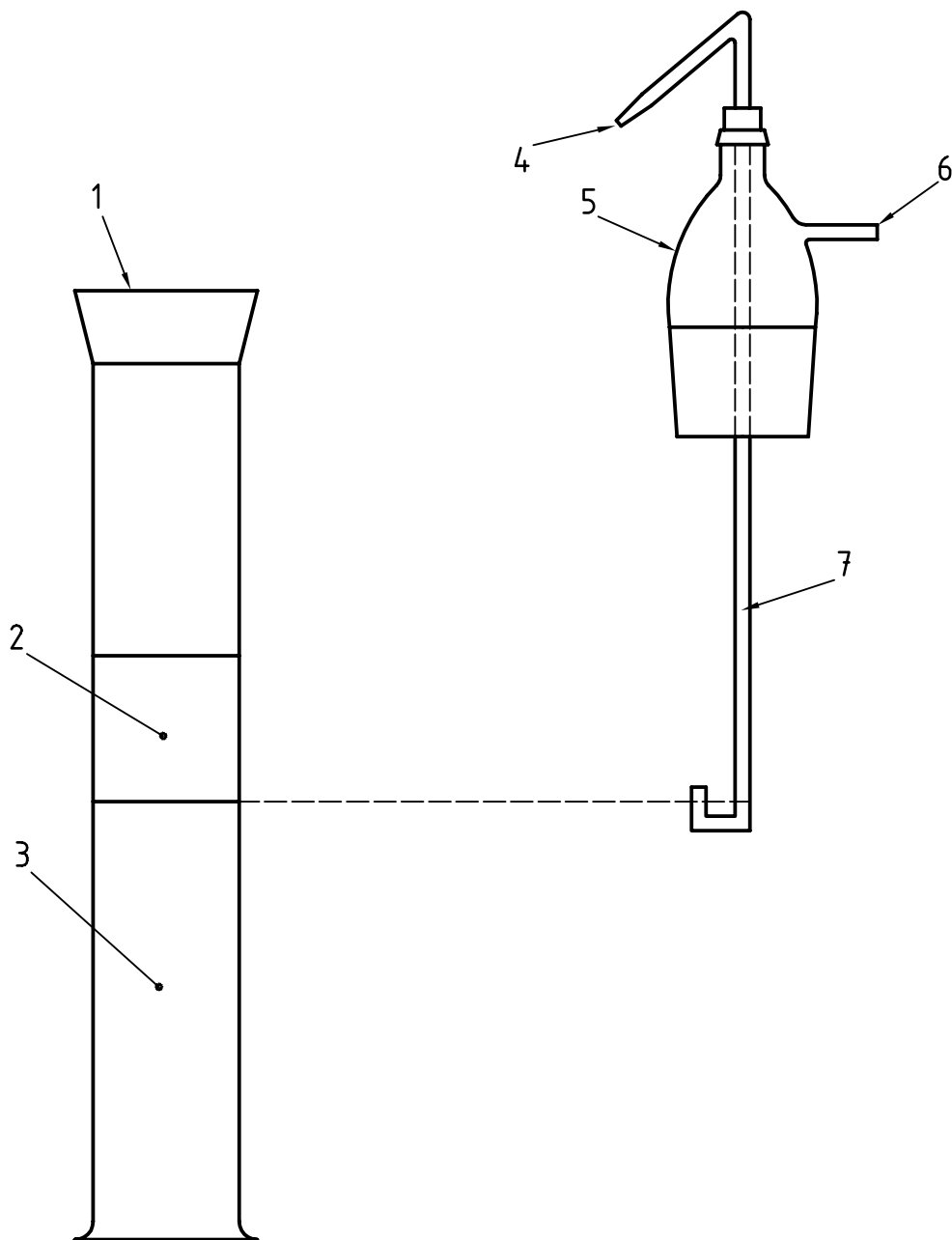
Store the sample in such a way that deterioration and change in its composition are prevented.

## 8 Preparation of test sample

Prepare the test sample in accordance with ISO 6498.

Just prior to starting the analysis, grind a portion of the well-mixed laboratory sample so that it passes through a sieve with 1 mm apertures. Mix thoroughly.

Homogenize canned pet foods. Mince semi-moist pet foods to pass through a plate with 4 mm apertures.



**Key**

- |   |   |   |  |
|---|---|---|--|
| 1 | Cylinder, of capacity 1 litre, with ground-glass neck | 5 | Bottle, of capacity 1 litre, with ground-glass joint |
| 2 | Light petroleum layer                                 | 6 | Side-arm   |
| 3 | Aqueous layer + saponified feed                       | 7 | Adjustable tub                                       |
| 4 | Jet   |   |  |

**Figure 1 — Example of extraction apparatus**



## 9 Procedure

### 9.1 General

Because of the sensitivity of vitamin A to UV radiation and air, perform all operations away from natural and strong fluorescent light and as rapidly as is consistent with accurate working. Use amber glassware where possible. Complete each assay within one working day. Carry out the saponification and extraction of the *all-trans*-retinyl acetate standard and the feeding stuff samples at the same time.

### 9.2 Saponification

Weigh, to the nearest 0,1 g, approximately 50 g of the prepared test sample (see clause 8) into a 1 litre conical flask.

Add 200 ml of ethanol (5.3). Swirl the flask contents to disperse the sample.

Add 2 ml of sodium ascorbate (5.10) and 50 ml of potassium hydroxide solution (5.2) and mix by swirling.

Fit a reflux condenser to the flask and immerse the flask in the boiling water bath (6.3).

Allow the contents of the flask to reflux for 60 min, swirling occasionally.

Remove and cool the flask to room temperature as rapidly as possible under a stream of cold water.

### 9.3 Extraction of vitamin A (retinol)

Transfer the contents of the flask to the extraction cylinder (see 6.5).

Rinse the flask with two 25 ml portions of ethanol or industrial methylated spirit (5.3) and transfer the rinsings to the cylinder.

Repeat the rinsing of the flask with two 125 ml portions of light petroleum (5.5) and one 250 ml portion of water (4.1), each time transferring the rinsings to the cylinder.

Stopper the cylinder and shake well for 1 min, taking care to release any pressure from time to time.

Cool the cylinder under a stream of cold water while waiting for the two liquid phases to separate, before removing the stopper.

When the layers have separated, remove the stopper, wash the sides of the stopper with a few millilitres of light petroleum (5.5) and insert the adjustable tube (see 6.5), positioning the lower open end so that it is just above the level of the interface.

By application of a slight pressure of inert gas (5.11) to the side-arm tube, transfer the upper, light petroleum layer to a 1 litre separating funnel (see 6.5).

Add 125 ml of light petroleum (5.5) to the cylinder, stopper and shake well for 1 min.

Allow the layers to separate and transfer the upper layer to the separating funnel using the adjustable tube (see 6.5) as before.

Again, add 125 ml of light petroleum (5.5) to the cylinder, stopper and shake well for 1 min.

Again, allow the layers to separate and transfer the upper layer to the separating funnel using the adjustable tube as before.

Wash the combined light petroleum extracts with four 100 ml portions of water using at first only gentle inversion then only gentle shaking in order to keep emulsion formation to a minimum.

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Transfer the washed extract through a medium/fast filter paper containing 60 g of anhydrous sodium sulfate (5.9), into a flask suitable for vacuum evaporation.

Rinse the separating funnel with two 20 ml portions of light petroleum (5.5) and add the rinsings, through the filter, to the evaporation flask.

Wash the filter further with two 25 ml portions of light petroleum (5.5) and collect the washings in the evaporation flask.

Evaporate the light petroleum extract to dryness under vacuum at a temperature not exceeding 40 °C.

Restore atmospheric pressure by admitting inert gas (5.11).

### 9.4 High-performance liquid chromatography

**9.4.1** Dissolve the residue in the minimum volume of 2-propanol (5.4) and transfer quantitatively to a 20 ml volumetric flask.

Rinse the evaporation flask with three small portions of 2-propanol (5.4), transferring the rinsings to the volumetric flask. Dilute to volume with 2-propanol (5.4) and mix. For material containing more than 100 000 IU/kg, further dilution may be necessary.

NOTE 1 IU = International Unit; 1 IU of vitamin A is equal to 0,300 µg of all-*trans*-retinol.

If necessary, filter the sample extract through a membrane filter (6.6).

**9.4.2** If delay is unavoidable, store the extract under inert gas (5.11) in a refrigerator at a temperature of 4 °C and then allow it to return to room temperature in the dark.

**9.4.3** Inject 10 µl of the sample extract onto the column of the liquid chromatograph (6.1) and measure the area of the retinol peak.

**9.4.4** Calculate the mean peak area from replicate injections of the sample extract and determine the retinol concentration of the extract by reference to the mean peak area found from replicate injections of a retinol standard of similar concentration. The retention time of retinol is about 5 min. Make alternate injections of sample extract and standard solution.

### 9.5 Hydrolysis of all-*trans*-retinyl acetate for calibration

Prepare a solution of all-*trans*-retinyl acetate (5.6) in ethanol (5.3) so that 1 ml contains approximately 15 000 IU of vitamin A.

NOTE 1 IU of vitamin A is equal to 0,344 µg of all-*trans*-retinyl acetate.

Using a 5 ml burette with 0,02 ml subdivisions, transfer 2,5 ml ± 0,02 ml of this solution to a 150 ml flask.

Add 20 ml of ethanol (5.3), 1 ml of potassium hydroxide solution (5.2) and 5 ml of sodium ascorbate solution (5.10).

Fit a condenser to the flask. Immerse the flask in the boiling water bath and allow to reflux for 60 min.

Cool the flask to room temperature under a stream of cold water and transfer the contents to a separating funnel (see 6.5).

Rinse the flask with 50 ml of water (4.1), followed by 25 ml of ethanol (5.3), adding the rinsings to the separating funnel.

Extract the aqueous phase with one 80 ml portion of light petroleum (5.5) and then with two 50 ml portions of light petroleum (5.5).

Combine the light petroleum extracts, then wash with two 50 ml portions of water. Add 2 g of anhydrous sodium sulfate (5.9).

Transfer the light petroleum extract quantitatively to a 250 ml volumetric flask and dilute to volume. The retinol concentration of this solution (solution 1) is approximately 150 IU/ml.

## 9.6 Standardization of retinol solution for calibration

Pipette  $5 \text{ ml} \pm 0,03 \text{ ml}$  of solution 1 (9.5) into a 50 ml volumetric flask and remove the solvent, at ambient temperature, with a stream of inert gas (5.11).

Dissolve the residue in 2-propanol (5.4), and then dilute to volume with 2-propanol.

Measure the absorbance ( $A$ ) of the solution, using 2-propanol as reference, at wavelengths of 310 nm, 325 nm and 334 nm. The absorbance values will be approximately 0,7 to 0,8. If necessary, an intermediate dilution may be used.

Using the following equation, calculate the corrected absorbance at 325 nm:

$$A_{325,\text{corr}} = 6,815 \times A_{325} - 2,555 \times A_{310} - 4,26 \times A_{334}$$

If  $A_{325,\text{corr}}/A_{325}$  is less than 0,97, use the value of  $A_{325,\text{corr}}$  for the standardization; otherwise use  $A_{325}$ .

The retinol concentration of solution 1 is given by:

$$\text{concentration (IU/ml)} = A_{325} \times 183 \text{ IU/ml, or}$$

$$\text{concentration (IU/ml)} = A_{325,\text{corr}} \times 183 \text{ IU/ml.}$$

## 9.7 Preparation of a retinol standard for chromatography

Prepare a retinol solution in 2-propanol (5.4) the concentration of which is approximately the same as that expected in the sample extract (9.4.1). For each 1000 IU of vitamin A per kilogram of sample, a retinol concentration of 2,5 IU/ml is expected in the extract.

Evaporate an aliquot volume of solution 1 (9.5) to dryness at ambient temperature with a stream of inert gas (5.11). Dissolve the residue in the appropriate volume of 2-propanol to provide the required retinol concentration and mix it.

If necessary, filter the standard solution through a membrane filter (6.6).

Alternatively for calibration purposes, a standard solution of vitamin A (retinol) in 2-propanol (5.4) prepared by diluting a stock standard solution of all-*trans*-retinol made by dissolving an appropriate quantity of all-*trans*-retinol standard substance (5.6.2) directly in 2-propanol (5.4) may be used.

In this case, check the vitamin A standard by measuring the absorbance of the standard solution in quartz cells (6.2) at wavelengths of 300 nm, 325 nm, 350 nm and 370 nm against 2-propanol as reference. Determine the  $A/A_{325}$  ratio at each of the wavelengths for all-*trans*-retinol. If the ratio does not exceed 0,602, 0,432 and 0,093 at 300 nm, 350 nm and 370 nm respectively, the standard substance is suitable for use (see references [5], [6]).

**NOTE** The use of a multiple point calibration may be used to facilitate analysis in cases where, for example, the expected concentration of vitamin A in the sample extract is unknown.

## 10 Expression of results

Calculate the numerical value of the vitamin A content of the test sample by the equation:

$$w_A = 20\,000 \times \frac{c}{m}$$

where

$w_A$  is the numerical value of the vitamin A content of the test sample, in International Units per kilogram (IU/kg);

$c$  is the numerical value of the retinol concentration of the extract, in International Units per millilitre (IU/ml);

$m$  is the numerical value of the mass of the test sample, in grams.

## 11 Precision

### 11.1 Interlaboratory tests

Details of interlaboratory tests on the precision of the method are given in annex A. The values derived from these tests may not be applicable to concentration ranges and matrices other than those given.

**Table 1 — Repeatability limit ( $r$ ) and reproducibility limit ( $R$ )**

Sample	Vitamin A content	$r$	$R$
	IU/kg	IU/kg	IU/kg
Pelleted poultry feed	9 888	860	2 068
Cattle ration	10 812	2 099	3 549
Fish food	15 879	2 800	4 672
Semi-moist pet food	33 250	1 536	2 446
Pig concentrate ration	40 661	4 049	7 941

### 11.2 Repeatability

The absolute difference between two independent single test results, obtained using the same method on identical test material in the same laboratory by the same operator using the same equipment within a short interval of time, will in not more than 5 % of cases exceed the repeatability limit ( $r$ ) mentioned in or derived from Table 1.

### 11.3 Reproducibility

The absolute difference between two single test results obtained using the same method on identical test material in different laboratories by different operators using different equipment, will in not more than 5 % of cases exceed the reproducibility limit ( $R$ ) mentioned in or derived from Table 1.

## 12 Test report

The test report shall specify:

- all information necessary for the complete identification of the sample;
- the sampling method used, if known;
- the test method used, with reference to this International Standard;
- all operating details not specified in this International Standard, or regarded as optional, together with details of any incidents which may have influenced the test result(s);
- the test result obtained, or the two test results obtained if the repeatability has been checked.

## Annex A (informative)

### Results of interlaboratory tests

The precision of the method was established by interlaboratory tests carried out in accordance with ISO 5725 [1]<sup>1)</sup>. The results of these tests have been published (see reference [7]). In the tests 12 to 14 laboratories participated and samples were investigated of semi-moist pet food, pelleted poultry feed, cattle ration, pig concentrate ration and fish food.

**Table A.1 — Statistical results of interlaboratory tests**

Parameter	Sample <sup>a</sup>				
	1	2	3	4	5
Number of laboratories	14	12	13	12	13
Number of individual results	56	24	26	24	26
Number of accepted results	56	24	26	24	26
Mean vitamin A content, IU/kg	33 250	9 888	10 812	40 661	15 879
Repeatability standard deviation ( $s_r$ ), IU/kg	668	304	742	1431	989
Repeatability coefficient of variation, %	2,0	3,1	6,9	3,5	6,2
Repeatability limit ( $r$ ) [ $r = 2,8 \times s_r$ ], IU/kg	1 536	860	2 099	4 049	2 800
Reproducibility standard deviation ( $s_R$ ), IU/kg	864	731	1 254	2 806	1 651
Reproducibility coefficient of variation, %	2,6	7,4	11,6	6,9	10,4
Reproducibility limit ( $R$ ) [ $R = 2,8 \times s_R$ ], IU/kg	2 446	2 068	3 549	7 941	4 672
<sup>a</sup> 1: semi-moist pet food; 2: pelleted poultry feed; 3: cattle ration; 4: pig concentrate ration; 5: fish food.					

1) ISO 5725:1986 (now withdrawn) was used to obtain the precision data.

## Bibliography

- [1] ISO 5725:1986, *Precision of test methods — Determination of repeatability and reproducibility for a standard test method by inter-laboratory tests.*
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