# INTERNATIONAL STANDARD

ISO 13338

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## **Determination of tissue corrosiveness of a gas or gas mixture**

Détermination de la corrosivité des gaz ou mélanges de gaz sur les tissus



Reference number ISO 13338:1995(E)

ISO 13338:1995(E)

#### **Foreword**

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Draft International Standards adopted by the technical committees are circulated to the member bodies for voting. Publication as an International Standard requires approval by at least 75 % of the member bodies casting a vote.

International Standard ISO 13338 was prepared by Technical Committee ISO/TC 58, Gas cylinders, Subcommittee SC 2, Cylinder fittings.

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ii

#### Introduction

The purpose of ISO 5145 is to establish practical criteria for the determination of outlet connections of gas cylinders of water capacity 150 litres or less. These criteria are based on certain physicochemical properties of the gases, in particular their corrosiveness.

One of the difficulties in the application of ISO 5145 resides in the absence of criteria for classifying the corrosiveness level of a gas or a gas mixture. In fact,

- in the case of pure gases, there are few data in the literature, but above all,
- in the case of gas mixtures, these data are very often nonexistent.

Investigation has shown that there is no correlation between the toxicity values defined by LC $_{50}$  (see ISO 10298) and corrosiveness.

### Determination of tissue corrosiveness of a gas or gas mixture

#### 1 Scope

This International Standard gives

- for pure gases: a complete list indicating their corrosiveness,
- for gas mixtures: a calculation method, in the absence of experimental data, relating to the corrosiveness of each of their components,

in order to determine the corrosiveness of gases and gas mixtures on tissue so that a suitable outlet connection can be assigned to each of them.

#### 2 Normative reference

The following standard contains provisions which, through reference in this text, constitute provisions of this International Standard. At the time of publication, the edition indicated was valid. All standards are subject to revision, and parties to agreements based on this International Standard are encouraged to investigate the possibility of applying the most recent edition of the standard indicated below. Members of IEC and ISO maintain registers of currently valid International Standards.

ISO 5145:1990, Cylinder valve outlets for gases and gas mixtures — Selection and dimensioning.

#### 3 Definitions and symbols

#### 3.1 Definitions

For the purposes of this International Standard, the following definitions apply.

**3.1.1 tissue corrosiveness of gases or gas mixtures:** Ability of a gas to damage or destroy living tissues (eyes, skin and mucous membranes).

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3.1.2 irritant gas: Gas which may cause a temporary reaction to the skin, eyes and mucous membranes.

NOTE 1 An irritant gas is regarded for the purposes of ISO 5145 as noncorrosive.

#### 3.2 Symbols

L limit

V volume

C indicates a corrosive component

C+ indicates a very corrosive component

i indicates an irritant component

nc indicates an noncorrosive component

#### 4 Classification

Two levels of corrosiveness are defined (C+: very corrosive, C: corrosive), to provide greater accuracy in the calculation method for the corrosiveness of mixtures.

In accordance with the above, gases are classified into the following categories:

C+: very corrosive
C: corrosive | corrosive in the sense of ISO 5145
i: irritant | noncorrosive, non-irritant | corrosive in the sense of ISO 5145

For a complete definition for purposes of the gas cylinder connection, the subdivisions of the FTSC code defined at the bottom of table 1 shall also be taken into account:

0: noncorrosive, non-irritant (nc)

1: forms nonhalogen acids (C+, C or i)

2: basic (C+, C or i)

3: forms halogen acids (C+, C or i)

#### 5 Categories of corrosiveness for pure gases

The corrosiveness category of each gas (C+, C, i or nc) corresponding to the classification defined in clause 4 is shown in table 1.

Table 1 — Corrosiveness categories of pure gases

Group No.	Gas name	Chemical formula	Synonym	FTSC Code	Corrosiveness category
7	Ammonia	NH <sub>3</sub>	R717	0202	С
4	Antimony pentafluoride	SbF <sub>5</sub>	0303		C+
8	Arsine	AsH <sub>3</sub>		2300	nc
12	Bis-(trifluoromethyl)peroxide	(CF <sub>3</sub> ) <sub>2</sub> O <sub>2</sub>		4300	nc
4	Boron trichloride	BCl <sub>3</sub>	Boron chloride	0203	С
4	Boron trifluoride	BF <sub>3</sub>	Boron fluoride	0253	C+
12	Bromine pentafluoride	BrF <sub>5</sub>		4303	C+
12	Bromine trifluoride	BrF <sub>3</sub>		4303	C+
4	Bromoacetone	CH₃COCH₂Br		0303	С
13	Buta-1,3-diene (inhibited)	$CH_2 = CH - CH = CH_2$		5100	nc
8	Carbon monoxide	со		2250	nc
8	Carbonyl sulfide	cos	Carbonoxylsulfide	2301	nc
4	Carbonyl fluoride	CF₂O		0213	C+
12	Chlorine	Cl <sub>2</sub>		4203	C+
12	Chlorine pentafluoride	CIF <sub>5</sub>		4303	C+
12	Chlorine trifluoride	CIF <sub>3</sub>		4303	C+
8	Chloromethane	CH <sub>3</sub> CI	Methyl chloride R40	2200	nc
13	Chlorotrifluoroethylene	C <sub>2</sub> CIF <sub>3</sub>		5200	nc
8	Cyanogen	(CN) <sub>2</sub>		2300	i
4	Cyanogen chloride	CICN		0303	С
8	Cyclopropane	C <sub>3</sub> H <sub>6</sub>	Trimethylene	2200	nc
4	Deuterium chloride	DCI		0213	С
4	Deuterium fluoride	DF		0203	C+
8	Deuterium selenide	D₂Se		2301	i
8	Deuterium sulfide	D <sub>2</sub> S		2301	i
13	Diborane	B <sub>2</sub> H <sub>6</sub>		5350	nc
4	Dibromodifluoromethane	CBr <sub>2</sub> F <sub>2</sub>	R12B2	0200	nc
4	Dichloro(2-chlorovinyl)arsine	C <sub>2</sub> H <sub>2</sub> AsCl <sub>3</sub>	Lewisite	0303	C+
8	Dichlorosilane	SiH <sub>2</sub> Cl <sub>2</sub>		2203	С
9	Diethylzinc	(C <sub>2</sub> H <sub>5</sub> ) <sub>2</sub> Zn		3300	nc
7	Dimethylamine	(CH <sub>3</sub> ) <sub>2</sub> NH		2202	С
8	Dimethylsilane	(CH <sub>3</sub> ) <sub>2</sub> SiH <sub>2</sub>		2300	nc
4	Diphosgene	C <sub>2</sub> O <sub>2</sub> Cl <sub>4</sub>		0303	С
4	Ethyldichloroarsine	C <sub>2</sub> H <sub>5</sub> AsCl <sub>2</sub>		0303	С
13	Ethylene oxide	C <sub>2</sub> H <sub>4</sub> O	Oxirane	5200	i
12	Fluorine	F <sub>2</sub>		4343	C+
8	Fluoroethane	C <sub>2</sub> H <sub>5</sub> F	Ethyl fluoride	2300	nc
8	Germane	GeH₄		2300	nc

Group No.	Gas name	Chemical formula	Synonym	FTSC Code	Corrosiveness category
8	Heptafluorobutyronitrile	C₃F <sub>7</sub> N		2300	nc
4	Hexafluoroacetone	C₃F <sub>6</sub> O	Perfluoroacetone	0203	С
8	Hexafluorocyclobutene	C <sub>4</sub> F <sub>6</sub>		2300	nc
4	Hydrogen bromide	HBr	Hydrobromic acid (anhydrous)	0203	С
4	Hydrogen chloride	HCI	Hydrochloric acid (anhydrous)	0213	С
13	Hydrogen cyanide	HCN	Hydrocyanic acid (anhydrous)	5301	i
4	Hydrogen fluoride	HF	Hydrofluoric acid (anhydrous)	0203	C+
4	Hydrogen iodide	НІ	Hydroiodic acid (anhydrous)	0203	С
8	Hydrogen selenide	H₂Se		2301	i
8	Hydrogen sulfide	H₂S		2301	i
12	lodine pentafluoride	IF <sub>5</sub>		4303	C+
4	lodotrifluoromethane	CF <sub>3</sub> I	Trifluoromethyl iodide	0200	nc
4	Methyl bromide	CH <sub>3</sub> Br	Bromomethane	0300	i
8	Methyl mercaptan	CH₃SH	Methanethiol	2201	i
13	Methyl vinyl ether (inhibited)	C₃H <sub>6</sub> O	Methoxyethylene	5200	nc
4	Methyldichloroarsine	CH <sub>3</sub> AsCl <sub>2</sub>		0303	C+
8	Methylsilane	CH₃SiH₃		2300	nc
7	Monoethylamine	C <sub>2</sub> H <sub>5</sub> NH <sub>2</sub>	Ethylamine R631	2202	С
7	Monomethylamine	CH <sub>3</sub> NH <sub>2</sub>	Methylamine R630	2202	С
4	Mustard gas	C₄H <sub>8</sub> Cl₂S		0303	C+
8	Nickel carbonyl	Ni(CO)₄	Nickel tetracarbonyl	2300	nc
12	Nitric oxide	NO	Nitrogen oxide	4351	С
12	Nitrogen dioxide	NO <sub>2</sub>	Nitrogen(IV) oxide	4301	С
12	Nitrogen trifluoride	NF <sub>3</sub>		4153	i
12	Nitrogen trioxide	N <sub>2</sub> O <sub>3</sub>	Nitrogen sesquioxide	4301	С
4	Nitrosyl chloride	NOCI		0203	C+
12	Oxygen difluoride	F <sub>2</sub> O		4343	C+
12	Ozone	O <sub>3</sub>		4330	i
9	Pentaborane	B <sub>5</sub> H <sub>10</sub>		3300	nc
8	Pentafluoropropionitrile	C <sub>3</sub> F <sub>5</sub> N		2300	nc
4	Perfluorobut-2-ene	C <sub>4</sub> F <sub>8</sub>		0200	nc
4	Phenylcarbylamine chloride	C <sub>6</sub> H <sub>5</sub> NCCl <sub>2</sub>		0303	С
4	Phosgene	COCl <sub>2</sub>	Carbonyl chloride	0303	С
9	Phosphine	PH <sub>3</sub>		3310	nc
4	Phosphorus pentafluoride	PF <sub>5</sub>		0203	C+
4	Phosphorus trifluoride	PF <sub>3</sub>		0203	C+
13	Propylene oxide	C₃H₅O	Methyl oxirane	5200	i

Group No.	Gas name	Chemical formula	Synonym	FTSC Code	Corrosiveness category
9	Silane	SiH₄	Silicon tetrahydride	3150	nc
4	Silicon tetrafluoride	SiF <sub>4</sub>	Tetrafluorosilane R764	0253	C+
4	Silicon tetrachloride	SiCl <sub>4</sub>		0203	С
13	Stibine	SbH <sub>3</sub>	Antimony hydride	5300	nc
4	Sulfur dioxide	SO <sub>2</sub>		0201	С
4	Sulfur tetrafluoride	SF <sub>4</sub>		0203	C+
4	Sulfuryl fluoride	SO <sub>2</sub> F <sub>2</sub>		0300	nc
8	Tetraethyllead	(C <sub>2</sub> H <sub>5</sub> ) <sub>4</sub> Pb		2300	nc
12	Tetrafluorohydrazine	N <sub>2</sub> F <sub>4</sub>		4343	C+
8	Tetramethyllead	(CH <sub>3</sub> ) <sub>4</sub> Pb		2300	nc
9	Triethylaluminium	(C <sub>2</sub> H <sub>5</sub> ) <sub>3</sub> AI		3300	nc
9	Triethylborane	(C <sub>2</sub> H <sub>5</sub> ) <sub>3</sub> B		3300	nc
8	Trifluoroacetonitrile	C <sub>2</sub> F <sub>3</sub> N		2300	i
8	Trifluoroethylene	C₂HF₃		2200	nc
7	Trimethylamine	(CH₃)₃N		2202	С
8	Trimethylsilane	(CH <sub>3</sub> ) <sub>3</sub> SiH		2300	nc
9	Trimethylstibine	(CH <sub>3</sub> ) <sub>3</sub> Sb		3300	nc
4	Tungsten hexafluoride	WF <sub>6</sub>		0303	С
4	Uranium hexafluoride	UF <sub>6</sub>		0303	С
13	Vinyl bromide (inhibited)	C₂H₃Br		5200	nc
13	Vinyl chloride (inhibited)	C <sub>2</sub> H <sub>3</sub> Cl	Chloroethylene R1140	5200	nc
13	Vinyl fluoride (inhibited)	C <sub>2</sub> H <sub>3</sub> F	Fluoroethylene R1141	5100	nc

#### **NOTES**

- 1 Description of each group:
- Group 4: nonflammable, toxic and corrosive or corrosive by hydrolysis;
- Group 7: basic, flammable and corrosive;
- Group 8: flammable, toxic and corrosive (acid) or noncorrosive;
- Group 9: spontaneously flammable;
- Group 12: oxidizing, toxic and corrosive;
- Group 13: flammable, subject to decomposition.
- 2 Key FTSC (ISO 5145)
- 0 = noncorrosive
- 1 = forms nonhalogenated acids
- 2 = basic
- 3 = forms halogenated acids

#### 6 Corrosiveness of gas mixtures — Calculation method

For each of the corrosiveness categories of a component, the lower concentration limits (as a volume percentage) corresponding to each of these corrosiveness categories for the mixture obtained are defined as shown in table 2 below.

Table 2 — Lower concentration limits for each category of corrosiveness of a component

Limits in volume percentage

Corrosiveness category of the component		Very corrosive	Corrosive (C)	Irritant (i)
	L <sub>C+</sub>	1	-	
Lower concentration limits	$L_{C}$	0,2	5	<del></del>
	$L_{i}$	0,02	0,5	5

#### 6.1 Gas mixtures containing one very corrosive, corrosive or irritant component

The diagram in table 3, which illustrates table 2, can be used in the following way:

The point corresponding to the percentage concentration of the component is placed on this diagram in the column corresponding to its corrosiveness category. The area in which it is located determines the corrosiveness of the mixture.

#### **EXAMPLE**

Mixture containing: 6 % NH<sub>3</sub> + 94 % N<sub>2</sub>

For the example chosen, ammonia is classified in clause 5, table 1, in category C (corrosive). By looking at table 2 or the diagram in table 3, it can be seen that for corrosive components the lower concentration limit for the "corrosive" category is  $L_{\rm C}=5$  % and the lower concentration limit for the "irritant" category is  $L_{\rm i}=0.5$  %.

The mixture in the example contains 6 % NH<sub>3</sub>, which is a concentration greater than 5 %, so the mixture is categorized as corrosive.

#### 6.2 Gas mixtures containing several very corrosive, corrosive or irritant components

The mixture shall be checked first to see if it is very corrosive (in accordance with 6.2.1). If it is not, then secondly if it is corrosive (in accordance with 6.2.2) and, if it is not, finally if it is irritant (in accordance with 6.2.3).

#### 6.2.1 Very corrosive gas mixture

A mixture of very corrosive gases will be classified as a "very corrosive gas mixture" if

$$\sum \left(\frac{V_{C+}}{L_{C+}}\right) \geqslant 1$$

where

 $V_{C+}$  is the percentage, by volume, of each very corrosive component;

L<sub>C+</sub> is the percentage limit, by volume, for very corrosive gas mixtures (this limit is equal to 1 % for each very corrosive component, see table 2).

Any corrosive or irritant gases that may be present in the mixture do not enter into the calculation.

6

**EXAMPLE** 

Mixture containing

in which

HF is very corrosive (C+), F2 is very corrosive (C+)

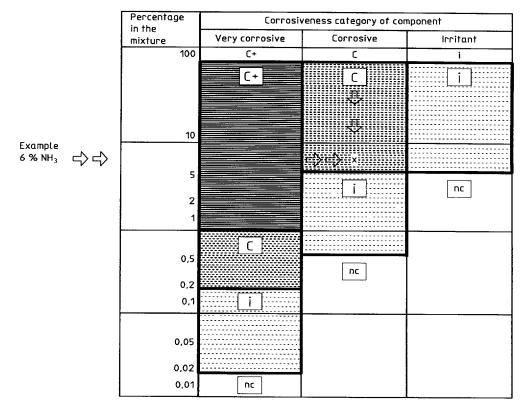
$$\sum \left(\frac{V_{C+}}{L_{C+}}\right) = \frac{0.5}{1} + \frac{0.6}{1} = 1.1 > 1$$

This gas mixture is thus very corrosive.

Table 3 — Diagram for a single very corrosive, corrosive or irritant component







#### 6.2.2 Corrosive gas mixtures

A mixture of very corrosive and/or corrosive gases will be classified as "corrosive" if

$$\sum \left(\frac{V_{\rm C}}{L_{\rm C}}\right) \ge 1$$

where

 $V_{\rm C}$  is the percentage, by volume, of each very corrosive or corrosive component;

L<sub>C</sub> is the limit percentage, by volume, for corrosive gas mixtures (this limit is equal to 0,2 % for each very corrosive gas component and to 5 % for each corrosive gas component, see table 2).

Any irritant gases present in the mixture do not enter into the calculation.

**EXAMPLE** 

Mixture containing:

in which

HF is very corrosive (C+), Cl<sub>2</sub> is very corrosive (C+), COCl<sub>2</sub> is corrosive (C)

$$\sum \left(\frac{V_{\rm C}}{L_{\rm C}}\right) = \frac{0.1}{0.2} + \frac{0.1}{0.2} + \frac{2}{5} = 1 + \frac{2}{5} = \frac{7}{5} > 1$$

This gas mixture is thus corrosive.

#### 6.2.3 Irritant gas mixtures

A mixture of very corrosive gases and/or corrosive gases and/or irritant gases will be classified as "irritant" if

$$\sum \left(\frac{V_{\rm i}}{L_{\rm i}}\right) \geqslant 1$$

where

 $V_i$  is the percentage, by volume, of each very corrosive, corrosive or irritant component;

L<sub>i</sub> is the limit percentage, by volume, for irritant gas mixtures (this limit is equal to 0,02 % for each very corrosive component, 0,5 % for each corrosive component and 5 % for each irritant component, see table 2).

**EXAMPLE** 

Mixture containing:

in which

 $\mathrm{Cl}_2$  is very corrosive (C+),  $\mathrm{COCl}_2$  is corrosive (C), HCN is irritant (i)

$$\sum \left(\frac{V_1}{L_1}\right) = \frac{0.01}{0.02} + \frac{0.4}{0.5} + \frac{3}{5} = \frac{19}{10} > 1$$

This gas mixture is thus irritant.

### Annex A (informative)

#### **Bibliography**

[1] ISO 10298:—11, Determination of the toxicity of a gas or gas mixture.

<sup>1)</sup> To be published.

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