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Adipate esters for industrial use — Determination of acidity to phenolphthalein — Volumetric method

Esters de l'acide adipique à usage industriel — Détermination de l'acidité à la phénolphthaléine — Méthode volumétrique

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FOREWORD

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It has been approved by the Member Bodies of the following countries :

Austria	Ireland	Switzerland
Belgium	Israel	Thailand
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France	Poland	U.S.A.
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No Member Body expressed disapproval of the document.

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Adipate esters for industrial use – Determination of acidity to phenolphthalein – Volumetric method

1 SCOPE AND FIELD OF APPLICATION

This International Standard specifies a volumetric method for the determination of acidity to phenolphthalein of adipate esters for industrial use.

The method is applicable to simple and mixed alcohol esters that are liquid at ambient temperature.

2 PRINCIPLE

Titration of the acidity using a standard volumetric solution of sodium hydroxide in the presence of phenolphthalein.

3 REAGENTS

Distilled water, or water of equivalent purity, shall be used in the test.

3.1 Sodium hydroxide, 0,1 N standard volumetric solution.

3.2 Ethanol, 95 % (V/V).

3.3 Phenolphthalein, 10 g/l ethanolic solution.

Dissolve 1,0 g of phenolphthalein in 100 ml of the ethanol (3.2) and make faintly pink by the addition of dilute sodium hydroxide solution.

4 APPARATUS

Ordinary laboratory apparatus and

4.1 Conical flask, of borosilicate glass, capacity 250 ml, fitted with a stopper carrying a soda-lime tube.

4.2 Burette, graduated in 0,02 ml or smaller divisions.

5 PROCEDURE

5.1 Test portion

Weigh, to the nearest 0,5 g, about 50 g of the laboratory sample.

5.2 Determination

Place 50 ml of the ethanol (3.2) in the conical flask (4.1), add 0,5 ml of the phenolphthalein solution (3.3) and make faintly pink by the addition of the sodium hydroxide solution (3.1). Add the test portion (5.1) and titrate the mixture with the sodium hydroxide solution (3.1) until the pink colour remains for 5 s.

6 EXPRESSION OF RESULTS

Acidity, expressed as a percentage by mass of adipic acid $[\text{COOH}(\text{CH}_2)_4\text{COOH}]$, is given by the formula

$$\frac{0,73 V}{m} = A$$

or, expressed as milli-equivalents per kilogram, by the formula

$$\frac{100 V}{m} = B$$

where

V is the volume, in millilitres, of the sodium hydroxide solution (3.1) used for the titration;

m is the mass, in grams, of the test portion.

7 TEST REPORT

The test report shall include the following particulars :

- the reference of the method used;
- the results and the method of expression used;
- any unusual features noted during the determination;
- any operation not included in this International Standard or regarded as optional.

