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Glycerols for industrial use — Determination of water content — Karl Fischer method

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FOREWORD

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It was approved in March 1971 by the Member Bodies of the following countries:

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Glycerols for industrial use — Determination of water content — Karl Fischer method

1 SCOPE AND FIELD OF APPLICATION

This International Standard specifies the manner of application of the Karl Fischer method for the determination of the water content of glycerols for industrial use.

2 REFERENCES

ISO/R 760, Determination of water by the Karl Fischer method.

ISO/R 1615, Aliphatic polyhydric alcohols — Glycerine for industrial use — Determination of alkalinity or acidity — Volumetric method.

ISO 2096, Glycerols for industrial use — Methods of sampling.

3 PRINCIPLE

Application of one of the methods of titration described in ISO/R 760, taking into account, in the expression of the results, the possible alkalinity of the sample.

4 SAMPLING

Prepare the laboratory sample as described in ISO 2096.

5 PROCEDURE

5.1 Test portion

Place in a weighing vessel, of suitable capacity and capable of being hermetically closed, a quantity of the laboratory sample corresponding to a maximum consumption of Karl Fischer reagent of the order of 20 to 25 ml. Determine the exact mass used by weighing to the nearest 0.001 g before and after transfer to the reaction vessel.

5.2 Determination

Determine the water content of the test portion (5.1) by carrying out any of the procedures described in ISO/R 760.

6 EXPRESSION OF RESULTS

Express the results to one place of decimals.

6.1 Direct visual or electrometric titration

Water content is given, as a percentage by mass, by the formula:

$$\frac{V \times T}{m \times 10}$$
 - 0.018 A

where

V is the volume, in millilitres, of Karl Fischer reagent used for the titration;

T is the water equivalent of the Karl Fischer reagent, in milligrams of H_2O per millilitre;

m is the mass, in grams, of the test portion;

A is the alkalinity of the sample 1, in milliequivalents per 100 g.

NOTE — In most cases the correction factor 0,018 A is negligible.

6.2 Electrometric back titration

Water content is given, as a percentage by mass, by the formula :

$$\left[\left(V - V_1 \times \frac{20}{V_2}\right) \times \frac{T}{m \times 10}\right] - 0.018 A$$

where

V is the volume, in millilitres, of Karl Fischer reagent used in excess:

 V_1 is the volume, in millilitres, of the water/methanol standard solution used for the back titration;

 V_2 is the volume, in millilitres, of the water/methanol standard solution corresponding to 20 ml of Karl Fischer reagent;

T is the water equivalent of the Karl Fischer reagent, in milligrams of H₂O per millilitre;

m is the mass, in grams, of the test portion;

A is the alkalinity of the sample¹⁾, in milliequivalents per 100 g.

NOTE -- In most cases the correction factor 0.018 A is negligible.

¹⁾ Determined according to ISO/R 1615.

7 TEST REPORT

The test report shall include the following particulars:

- a) the reference of the method used;
- b) the results and the method of expression used;
- c) any unusual features noted during the determination;
- d) any operation not included in this International Standard, or the ISO documents to which reference is made, or regarded as optional.