

Ammonium nitrate —

Part 9: Method for determination of sulphate content

NOTE It is recommended that this Part of BS 4267 be read in conjunction with the information in the “General introduction”, published separately as BS 4267-0.

WARNING. Ammonium nitrate is a strong oxidizing agent. If necessary, break the test sample up by crushing rather than grinding.

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The Committees responsible for this British Standard are shown in Part 0.

The following BSI references relate to the work on this standard:
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Foreword

This Part of BS 4267 has been prepared under the direction of the Chemicals Standards Committee. It supersedes clause 12 of BS 4267:1968, to which it is technically equivalent and which has been deleted by amendment.

This Part of BS 4267 is related to ISO 3329 “*Ammonium nitrate for industrial use — Determination of sulphur compounds — Method by reduction and titrimetry*” published by the International Organization for Standardization (ISO).

This standard describes a method of test only, and should not be used or quoted as a specification defining limits of purity. Reference to this Part should indicate that the method of test used is in accordance with BS 4267-9:1987.

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Summary of pages

This document comprises a front cover, an inside front cover, pages i and ii, pages 1 and 2, an inside back cover and a back cover.

This standard has been updated (see copyright date) and may have had amendments incorporated. This will be indicated in the amendment table on the inside front cover.

1 Scope

This Part of BS 4267 describes a method for determination of sulphate content of ammonium nitrate for industrial use. The method is applicable to products with a sulphate content of between 10 mg/kg and 200 mg/kg.

NOTE The titles of the publications referred to in this standard are listed on the inside back cover.

2 Principle

A test portion is volatilized after the addition of sodium carbonate. The soluble residue is dissolved in hot water, the solution filtered and the sulphate precipitated by the addition of barium chloride solution. The sulphate content is determined by a visual turbidimetric method, using standard solutions of sulphuric acid for reference.

3 Reagents

3.1 General. During the analysis, use only reagents of recognized analytical grade and water complying with BS 3978.

3.2 Sodium carbonate, free from sulphate (see 5.3).

3.3 Hydrochloric acid solution, $c(\text{HCl}) = 2 \text{ mol/L}$ approximately.

3.4 Sulphuric acid solution, $c(\frac{1}{2}\text{H}_2\text{SO}_4) = 0.002 \text{ mol/L}$.

3.5 Sulphate seeding reagent, freshly prepared.

Dissolve 2.0 g of barium chloride dihydrate, $\text{BaCl}_2 \cdot 2\text{H}_2\text{O}$, in 75 mL of water, add 20 mL of 95 % (V/V) ethanol and 5 mL of the sulphuric acid solution (3.4).

4 Apparatus

4.1 Six matched Nessler cylinders, complying with BS 612.

4.2 Burette, class A, 25 mL, graduated in 0.05 mL, complying with BS 846.

4.3 One-mark volumetric flask, 50 mL, complying with BS 1792.

4.4 Porcelain dish, about 100 mm in diameter.

4.5 Furnace, capable of being maintained at $600 \pm 10 \text{ }^\circ\text{C}$.

5 Procedure

5.1 Test portion

Weigh, to the nearest 0.1 g, approximately 50 g of the test sample.

5.2 Standard solutions

Prepare the range of standard solutions at the same time as the determination. Using one Nessler cylinder (4.1) for each standard solution, add successively 2.0, 4.0, 6.0, 8.0 and 10.0 mL of the sulphuric acid solution (3.4). Make up each cylinder to 20 mL with water, add 1 mL of the sulphate seeding reagent (3.5), mix well and stand for 5 min. These standard solutions contain 196 μg to 980 μg sulphate, as H_2SO_4 , i.e. approximately 200 μg to 1 000 μg .

5.3 Blank test

Weigh 0.02 g of the sodium carbonate (3.2) and dissolve it in 20 mL of water in a Nessler cylinder (4.1). Add 0.4 mL of the hydrochloric acid solution (3.3) and mix. Add 1 mL of the sulphate seeding reagent (3.5) and stand for 5 min.

If turbidity is observed discard the sodium carbonate. Repeat this procedure with sodium carbonate from different samples, until no turbidity is observed.

5.4 Determination

5.4.1 Preparation of the test solution. Transfer the test portion (5.1) into the porcelain dish (4.4) and dissolve in about 40 mL of water. Add 0.05 g of the sodium carbonate (3.2) and place the dish on a sand bath at a temperature of approximately $120 \text{ }^\circ\text{C}$ in a fume cupboard. When the water has evaporated, heat over a low flame until white fumes have ceased to be evolved. Heat the dish at $600 \pm 10 \text{ }^\circ\text{C}$ in the furnace (4.5) for 15 min. Allow to cool.

WARNING. If the ammonium nitrate is heated too rapidly, a violent decomposition may result.

Dissolve the residue in 10 mL of water, add 1 mL of the hydrochloric acid solution (3.3) and filter into the volumetric flask (4.3). Wash the filter paper with several portions of water and dilute to the mark.

5.4.2 Turbidity test. If the sulphate content is expected to be between 10 mg/kg and 50 mg/kg transfer 20.0 mL of the test solution (5.4.1) to one of the Nessler cylinders (4.1).

If the sulphate content is expected to be between 40 mg/kg and 200 mg/kg transfer 5.0 mL of the test solution (5.4.1) to one of the Nessler cylinders (4.1) and add 15.0 mL of water.

Add 1 mL of the sulphate seeding reagent, mix well and stand for 5 min.

Compare the turbidity produced with that of each of the standard solutions (5.2) and note the sulphate content (in μg) of the sulphate standard which it most nearly matches.

6 Expression of results

The sulphate content, expressed as sulphuric acid (H_2SO_4), in mg/kg, is given by the following expression:

$$\frac{M}{V}$$

where

- M is the mass of H_2SO_4 (in μg), in the sulphate standard, the turbidity of which most nearly matches that produced from the test solution (5.4.2);
- V is the volume of the test solution (5.4.1) taken (in mL).

7 Test report

The test report shall include the following information:

- an identification of the sample;
- a reference to this British Standard method, i.e. BS 4267-9:1987;
- the results expressed in accordance with clause 6;
- any unusual features noted during the determination;
- any operation not included in this Part of BS 4267 or regarded as optional.

Publications referred to

BS 612, *Specification for Nessler cylinders.*

BS 846, *Specification for burettes.*

BS 1792, *Specification for one-mark volumetric flasks.*

BS 3978, *Specification for water for laboratory use.*

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