

BS EN 60086-3:2016



BSI Standards Publication

# Primary batteries

Part 3: Watch batteries

### **National foreword**

This British Standard is the UK implementation of EN 60086-3:2016. It is identical to IEC 60086-3:2016. It supersedes BS EN 60086-3:2011 which will be withdrawn 29 June 2019.

The UK participation in its preparation was entrusted to Technical Committee CPL/35, Primary cells.

A list of organizations represented on this committee can be obtained on request to its secretary.

This publication does not purport to include all the necessary provisions of a contract. Users are responsible for its correct application.

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EUROPEAN STANDARD

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September 2016

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English Version

**Primary batteries - Part 3: Watch batteries  
(IEC 60086-3:2016)**Piles électriques - Partie 3 : Piles pour montres  
(IEC 60086-3:2016)Primärbatterien - Teil 3: Uhrenbatterien  
(IEC 60086-3:2016)

This European Standard was approved by CENELEC on 2016-06-29. CENELEC members are bound to comply with the CEN/CENELEC Internal Regulations which stipulate the conditions for giving this European Standard the status of a national standard without any alteration.

Up-to-date lists and bibliographical references concerning such national standards may be obtained on application to the CEN-CENELEC Management Centre or to any CENELEC member.

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European Committee for Electrotechnical Standardization  
Comité Européen de Normalisation Electrotechnique  
Europäisches Komitee für Elektrotechnische Normung

**CEN-CENELEC Management Centre: Avenue Marnix 17, B-1000 Brussels**

## **European foreword**

The text of document 35/1359/FDIS, future edition 4 of IEC 60086-3, prepared by IEC/TC 35 "Primary cells and batteries" and by ISO/TC 114 "Horology" was submitted to the IEC-CENELEC parallel vote and approved by CENELEC as EN 60086-3:2016.

The following dates are fixed:

- latest date by which the document has to be implemented at national level by publication of an identical national standard or by endorsement (dop) 2017-03-29
- latest date by which the national standards conflicting with the document have to be withdrawn (dow) 2019-06-29

This document supersedes EN 60086-3:2011.

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## **Endorsement notice**

The text of the International Standard IEC 60086-3:2016 was approved by CENELEC as a European Standard without any modification.

In the official version, for Bibliography, the following note has to be added for the standard indicated :

IEC 60068-2-78:2001      NOTE      Harmonized as EN 60068-2-78:2001 (not modified).

## Annex ZA (normative)

### Normative references to international publications with their corresponding European publications

The following documents, in whole or in part, are normatively referenced in this document and are indispensable for its application. For dated references, only the edition cited applies. For undated references, the latest edition of the referenced document (including any amendments) applies.

NOTE 1 When an International Publication has been modified by common modifications, indicated by (mod), the relevant EN/HD applies.

NOTE 2 Up-to-date information on the latest versions of the European Standards listed in this annex is available here: [www.cenelec.eu](http://www.cenelec.eu).

<u>Publication</u>	<u>Year</u>	<u>Title</u>	<u>EN/HD</u>	<u>Year</u>
IEC 60086-1	2015	Primary batteries - Part 1: General	EN 60086-1	2015
IEC 60086-2	2015	Primary batteries - Part 2: Physical and electrical specifications	EN 60086-2	2016
IEC 60086-4	2014	Primary batteries - Part 4: Safety of lithium batteries	EN 60086-4	2015
IEC 60086-5	2016	Primary batteries - Part 5: Safety of batteries with aqueous electrolyte	EN 60086-5	1)

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1) To be published.

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# INTERNATIONAL ELECTROTECHNICAL COMMISSION

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## **PRIMARY BATTERIES –**

### **Part 3: Watch batteries**

#### **FOREWORD**

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International Standard IEC 60086-3 has been prepared by IEC technical committee 35: Primary cells and batteries, and ISO technical committee 114: Horology.

This fourth edition cancels and replaces the third edition published in 2011. This edition constitutes a technical revision.

This edition includes the following significant technical changes with respect to the previous edition:

- a) A harmonization of the cell sizes and service output tests with IEC 60086-2;
- b) Clarifications of Clauses 6: Sampling and Quality Assurance, 7: Test methods, and 8: Visual examination and acceptance condition;
- c) Harmonization of temperature and humidity conditions with IEC 60086-1.



This publication is published as a double logo standard.

The text of this standard is based on the following documents:

FDIS	Report on voting
35/1359/FDIS	35/1362/RVD

Full information on the voting for the approval of this standard can be found in the report on voting indicated in the above table.

This publication has been drafted in accordance with the ISO/IEC Directives, Part 2.

A list of all parts in the IEC 60086 series, published under the general title *Primary batteries*, can be found on the IEC website.

The committee has decided that the contents of this publication will remain unchanged until the stability date indicated on the IEC website under "<http://webstore.iec.ch>" in the data related to the specific publication. At this date, the publication will be

- reconfirmed,
- withdrawn,
- replaced by a revised edition, or
- amended.

## INTRODUCTION

This part of IEC 60086 provides specific requirements and information for primary watch batteries. This part of IEC 60086 was prepared through joint work between the IEC and ISO to benefit primary battery users, watch designers and battery manufacturers by ensuring the best compatibility between batteries and watches.

This part of IEC 60086 will remain under continual scrutiny to ensure that the publication is kept up to date with the advances in both battery and watch technologies.

NOTE Safety information is available in IEC 60086-4 and IEC 60086-5.

# PRIMARY BATTERIES –

## Part 3: Watch batteries

### 1 Scope

This part of IEC 60086 specifies dimensions, designation, methods of tests and requirements for primary batteries for watches. In several cases, a menu of test methods is given. When presenting battery electrical characteristics and/or performance data, the manufacturer specifies which test method was used.

### 2 Normative references

The following documents, in whole or in part, are normatively referenced in this document and are indispensable for its application. For dated references, only the edition cited applies. For undated references, the latest edition of the referenced document (including any amendments) applies.

IEC 60086-1:2015, *Primary batteries – Part 1: General*

IEC 60086-2:2015, *Primary batteries – Part 2: Physical and electrical specifications*

IEC 60086-4:2014, *Primary batteries – Part 4: Safety of lithium batteries*

IEC 60086-5:-1, *Primary batteries – Part 5: Safety of batteries with aqueous electrolyte*

### 3 Terms and definitions

For the purposes of this document, the terms and definitions given in IEC 60086-1 as well as the following terms and definitions apply.

#### 3.1

##### **capacitive reactance**

part of the internal resistance, that leads to a voltage drop during the first seconds under load

#### 3.2

##### **capacity**

electric charge (quantity of electricity) which a cell or battery can deliver under specified discharge conditions

Note 1 to entry: The SI unit for electric charge is the coulomb (1 C = 1 As) but, in practice, capacity is usually expressed in ampere hours (Ah).

#### 3.3

##### **fresh battery**

undischarged battery 60 days maximum after date of manufacture

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<sup>1</sup> To be published.

### 3.4

#### ohmic drop

part of the internal resistance that leads to a voltage drop immediately after switching the load on

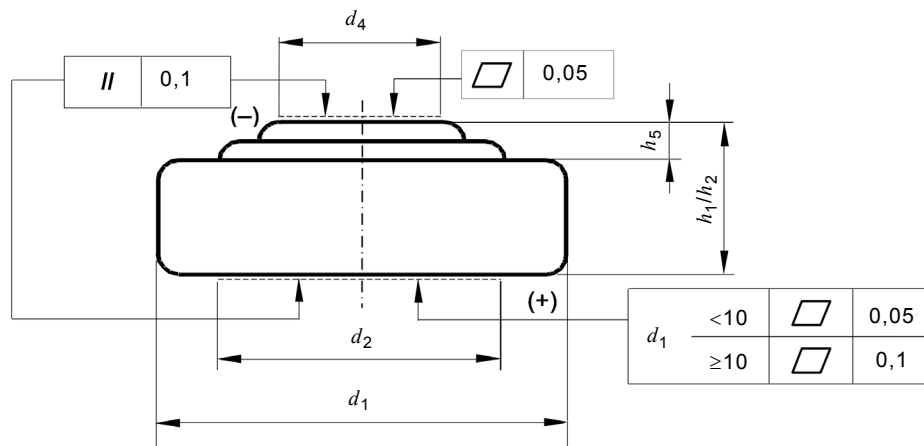
## 4 Physical requirements

### 4.1 Battery dimensions, symbols and size codes

Dimensions and tolerances of batteries for watches shall be in accordance with Figure 1, Table 1 and Table 2. The dimensions of the batteries shall be tested in accordance with 7.1.

The symbols used to denote the various dimensions in Figure 1 are in accordance with IEC 60086-2:2015, Clause 4.

*Dimensions in millimetres*



IEC

#### Key

$h_1$  maximum overall height of the battery

$h_2$  minimum distance between the flats of the positive and negative contacts

$h_5$  minimum projection of the flat negative contact

$d_1$  maximum and minimum diameter of the battery

$d_2$  minimum diameter of the flat positive contact

$d_4$  minimum diameter of the flat negative contact

NOTE This numbering follows the harmonization in the IEC 60086 series.

**Figure 1 – Dimensional drawing**

Table 1 – Dimensions and size codes

Dimensions in millimetres

Diameter		Height $h_1/h_2$														
		Code <sup>a</sup>														
Code <sup>a</sup>	$d_1$	Tolerance	Tolerance													
			10	12	14	16	20	21	25	26	27	30	31	32	36	42
4	4,8	$0$ -0,15	$0$ -0,10	$0$ -0,15	$0$ -0,15	$0$ -0,18	$0$ -0,20	$0$ -0,20	$0$ -0,20	$0$ -0,20	$0$ -0,20	$0$ -0,25	$0$ -0,25	$0$ -0,25	$0$ -0,25	$0$ -0,25
5	5,8	$0$ -0,15	1,05	1,25	1,45	1,65	1,65	1,65	2,15	2,15	2,15	2,70				
6	6,8	$0$ -0,15	1,05	1,25	1,45	1,65	1,65	1,65	2,15	2,15	2,60					
7	7,9	$0$ -0,15	1,05	1,25	1,45	1,65	1,65	1,65	2,10	2,10	2,60				3,10	5,40
9	9,5	$0$ -0,15	1,05	1,25	1,45	1,65	1,65	2,05	2,10	2,10	2,70					
10	10,0	$0$ -0,30									2,50					
11	11,6	$0$ -0,20	1,05	1,25	1,45	1,65	1,65	2,05	2,10	2,10	2,60			3,05	3,60	5,40
12	12,5	$0$ -0,25		1,20		1,60	2,00	2,00			2,50					

NOTE Open boxes in the above matrix are not necessarily available for standardisation due to the concept of overlapping tolerances.

<sup>a</sup> See Annex A.

**Table 2 – Dimensions and size codes***Dimensions in millimetres*

Diameter			$d_4$	Height $h_1/h_2$					
Code <sup>a</sup>	$d_1$	Tolerance		Code <sup>a</sup>					
				12	16	20	25	30	32
				Tolerances					
0 -0,20	0 -0,20	0 -0,25	0 -0,30	0 -0,30	0 -0,30				
16	16	0 -0,25	5,00	1,20	1,60	2,00	2,50		3,20
20	20	0 -0,25	8,00	1,20	1,60	2,00	2,50		3,20
23	23	0 -0,30	8,00	1,20	1,60	2,00	2,50	3,00	
24	24,5	0 -0,30	8,00	1,20	1,60			3,00	

NOTE Open boxes in the above matrix are not necessarily available for standardisation due to the concept of overlapping tolerances.

<sup>a</sup> See Annex A.

## 4.2 Terminals

Negative contact (-): the negative contact (dimension  $d_4$ ) shall be in accordance with Tables 1 and 2. This is not applied to those batteries with a two-step negative contact.

Positive contact (+): the cylindrical surface is connected to the positive terminal. Positive contact should be made to the side of the battery but may be made to the base.

## 4.3 Projection of the negative terminal ( $h_5$ )

The dimension  $h_5$  shall be as follows:

$$h_5 \geq 0,02 \text{ for } h_1/h_2 \leq 1,65$$

$$h_5 \geq 0,06 \text{ for } 1,65 < h_1/h_2 < 2,5$$

$$h_5 \geq 0,08 \text{ for } h_1/h_2 \geq 2,5$$

The negative contact should be the highest point of the battery.

## 4.4 Shape of negative terminal

The space requirements shall be contained within an angle of 45° (see Figure 2).

The minimum values of  $l_1$ , for different heights of  $h_1/h_2$ , are given in Table 3.

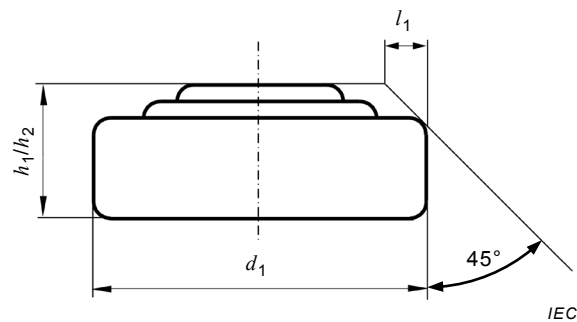


Figure 2 – Shape of negative terminal

Table 3 – Minimum values of  $l_1$ 

Dimensions in millimetres

$h_1/h_2$	$l_1 \text{ min}$
$1 < h_1/h_2 \leq 1,90$	0,20
$1,90 < h_1/h_2 \leq 3,10$	0,35
$3,60 \leq h_1/h_2 \leq 4,20$	0,70
$5,40 \leq h_1/h_2$	0,90

#### 4.5 Mechanical resistance to pressure

A force  $F$  (N), as specified in Table 4, applied for 10 s through a steel ball of 1 mm diameter, at the centre of each contact area, shall not cause any deformation prejudicial to the proper functioning of the battery, i.e. after this test, the battery shall pass the tests specified in Clause 7.

Table 4 – Applied force  $F$  by battery dimensions

Battery dimensions		Force
$d_1$	$h_1/h_2$	$F$
mm	mm	N
<7,9	<3,0	5
	$\geq 3,0$	10
$\geq 7,9$	<3,0	10
	$\geq 3,0$	10

#### 4.6 Deformation

The dimensions of batteries shall conform with the relevant specified dimensions at all times including discharge to the defined end-point voltage.

NOTE 1 A battery height increase up to 0,25 mm can occur, if discharged below this voltage.

NOTE 2 A battery height decrease can occur in B and C systems as discharge continues.

#### 4.7 Leakage

Undischarged batteries and, if required, batteries tested according to 7.2.6 shall be examined as stated in 7.3. The acceptable number of defects shall be agreed between the manufacturer and the purchaser.

## 4.8 Marking

### 4.8.1 General

The designation and the polarity shall be marked on the battery. Battery marking should not impede electrical contact. All other markings may be given on the packing instead of on the battery:

- a) designation according to normative Annex A, or common;
- b) expiration of a recommended usage period or year and month or week of manufacture;  
The year and month or week of manufacture may be in code. The code is composed by the last digit of the year and by a number indicating the month. October, November and December should be represented by the letters O, Y and Z respectively.

#### EXAMPLE

41: January 2014;

4Y: November 2014.

- c) polarity of the positive (+) terminal;
- d) nominal voltage;
- e) name or trade mark of the supplier;
- f) cautionary advice;
- g) caution for ingestion of batteries shall be given. Refer to IEC 60086-4:2014 (7.2 a) and 9.2) and IEC 60086-5:-<sup>1</sup> (7.1 l) and 9.2) for details.

NOTE Examples of the common designations can be found in Annex D of IEC 60086-2:2015.

### 4.8.2 Disposal

Marking of batteries with respect to the method of disposal shall be in accordance with local legal requirements.

## 5 Electrical requirements

### 5.1 Electrochemical system, nominal voltage, end-point voltage and open-circuit voltage

The requirements concerning the electrochemical system, the nominal voltage, the end-point voltage and the open-circuit voltage are given in Table 5.

**Table 5 – Standardised electrochemical systems**

Letter	Negative electrode	Electrolyte	Positive electrode	Nominal voltage ( $V_n$ ) V	End-point voltage (EV) V	Open-circuit voltage ( $U_{OC}$ or OCV) V	
						Max.	Min.
B	Lithium (Li)	Organic electrolyte	Carbon monofluoride (CF) <sub>x</sub>	3,0	2,0	3,70	3,00
C	Lithium (Li)	Organic electrolyte	Manganese dioxide (MnO <sub>2</sub> )	3,0	2,0	3,70	3,00
L	Zinc (Zn)	Alkali metal hydroxide	Manganese dioxide (MnO <sub>2</sub> )	1,5	1,0	1,68	1,50
S	Zinc (Zn)	Alkali metal hydroxide	Silver oxide (Ag <sub>2</sub> O)	1,55	1,2	1,63	1,57



## 5.2 Closed circuit voltage $U_{CC}$ (CCV), internal resistance and impedance

Closed circuit voltage and internal resistance shall be measured according to 7.2.

AC impedance should be measured with an LCR meter.

Limit values shall be agreed between the manufacturer and the purchaser.

## 5.3 Capacity

The capacity shall be agreed between the manufacturer and the purchaser on the basis of a continuous discharge test lasting approximately 30 days, according to 7.2.6.

## 5.4 Capacity retention

The capacity retention is the ratio between the capacities under the given discharge conditions measured on fresh batteries and a sample of the same lot stored during 365 days at  $(20 \pm 2)$  °C and a relative humidity between  $55 \pm 20$  %.

The ratio of capacity retention shall be agreed between the manufacturer and the purchaser. The minimum value should be at least 90 % for a period of 12 months. The capacity measurement is carried out according to 7.2.6.

For the purpose of verifying compliance with this standard, conditional acceptance may be given after completion of the initial capacity tests.

## 6 Sampling and quality assurance

The use of sampling plans or product quality indices should be agreed between manufacturer and purchaser.

Where no agreement is specified, refer to ISO 2859 and ISO 21747 for sampling and quality compliance assessment advice.

## 7 Test methods

### 7.1 Shape and dimensions

#### 7.1.1 Shape requirement

The shape of the negative contact is checked preferably by optical projection or by an open gauge according to Figure 3.

The measurement method shall be agreed between the manufacturer and the purchaser.

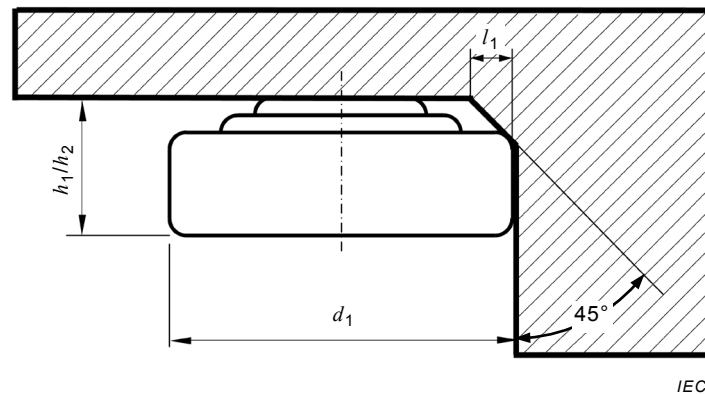


Figure 3 – Shape requirement

## 7.2 Electrical characteristics

### 7.2.1 Environmental conditions

Unless otherwise specified, the sample batteries shall be tested at a temperature of  $(20 \pm 2) ^\circ\text{C}$  and a relative humidity between  $55 + 20 / - 40 \%$ .

During use, batteries may be exposed to low temperatures; it is therefore recommended to carry out complementary tests at  $(0 \pm 2) ^\circ\text{C}$  and at  $(-10 \pm 2) ^\circ\text{C}$ .

### 7.2.2 Equivalent circuit – effective internal resistance – DC method

Resistance of any electrical component determined by calculating the ratio between the voltage drop  $\Delta U$  across this component and the range of current  $\Delta i$  passing through this component and causing the voltage drop  $R = \Delta U / \Delta i$ .

NOTE As an analogy, the internal d.c. resistance  $R_i$  of any electrochemical cell is defined by the following relation:

$$R_i (\Omega) = \frac{\Delta U (\text{V})}{\Delta i (\text{A})} \quad (1)$$

The internal d.c. resistance is illustrated by the schematic voltage transient as given below in Figure 4.

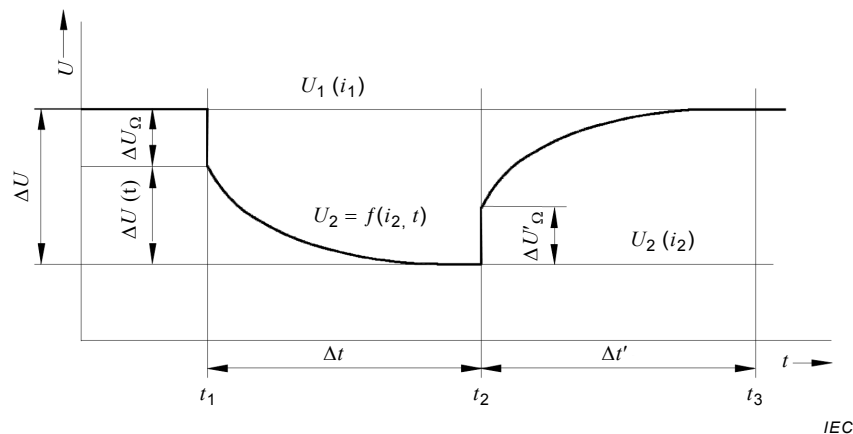


Figure 4 – Schematic voltage transient

As can be seen from the diagram in Figure 4, the voltage drop  $\Delta U$  of the two components differs in nature, as shown in the following relation:

$$\Delta U = \Delta U_{\Omega} + \Delta U(t) \quad (2)$$

The first component  $\Delta U_{\Omega}$  for ( $t = t_1$ ) is independent of time (ohmic drop), and results from the increase in current  $\Delta i$  according to the relation:

$$\Delta U_{\Omega} = \Delta i \times R_{\Omega} \quad (3)$$

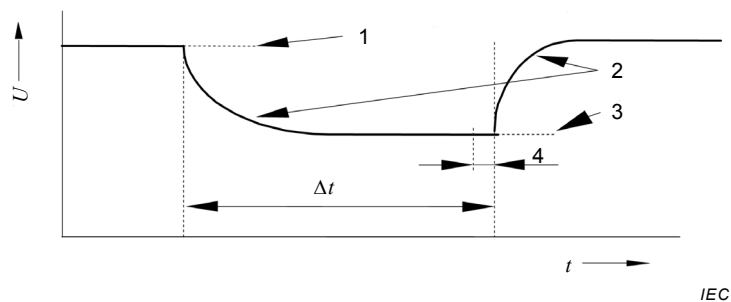
In this relation,  $R_{\Omega}$  is a pure ohmic resistance. The second component  $\Delta U(t)$  is time dependent and is of electrochemical origin (capacitive reactance).

### 7.2.3 Equipment

The equipment used for the voltage measurements shall have the following specifications:

- accuracy:  $\leq 0,25 \%$ ;
- precision:  $\leq 50 \%$  of last digit;
- internal resistance:  $\geq 1 \text{ M}\Omega$ ;
- measurement time: in the tests proposed in the following subclauses, it is important to make sure that the measurement is taken during the flat period of the voltage transient (see Figure 5). Otherwise, a measurement error due to the capacitive reactance may occur (lower internal resistance).

The time  $\Delta t'$  necessary for the measurement shall be brief in comparison to  $\Delta t$ , and the measurement equipment compatible with these criteria.



#### Key

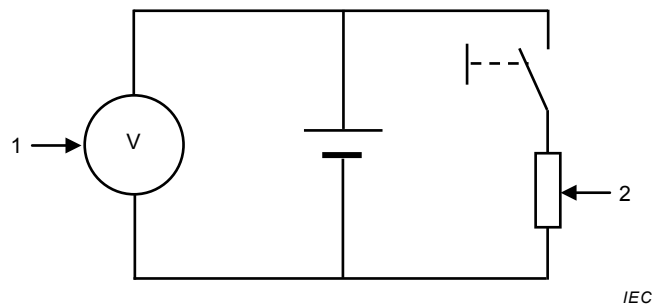
- 1 open-circuit voltage  $U_{oc}$  (OCV)
- 2 effect of capacitive reactance
- 3 closed circuit voltage  $U_{cc}$  (CCV)
- 4  $\Delta t'$  (measurement  $U_{cc}$ )

Figure 5 – Curve:  $U = f(t)$

### 7.2.4 Measurement of open-circuit voltage $U_{oc}$ (OCV) and closed circuit voltage $U_{cc}$ (CCV)

Refer to Figure 6:

- First measurement  $U_{oc}$ : The switch is left open while this measurement is being carried out.
- Next measurement  $U_{cc}$ : The battery being tested shall be connected to the load  $R_m$ . The switch shall be left closed during the duration  $\Delta t$  according to Table 6.

**Key**

- 1 reading  $U_{cc} / U_{oc}$   
 2  $R_m$  resistance of measurement

**Figure 6 – Circuitry principle****Table 6 – Test method for  $U_{cc}$  (CCV) measurement**

Test method	Battery with KOH electrolyte <sup>a</sup>		All other batteries	
	$R_m$ $\Omega$	$\Delta t$ s	$R_m$ $\Omega$	$\Delta t$ ms
A <sup>b</sup>	$150 \pm 0,5 \%$	$1 \pm 5 \%$	$1\,500 \pm 0,5 \%$	$10 \pm 5 \%$
B <sup>c</sup>	$150 \pm 0,5 \%$	0,5 – 2	$470 \pm 0,5 \%$	500 – 2 000
C <sup>d</sup>	$200 \pm 0,5 \%$	$5 \pm 5 \%$	$2\,000 \pm 0,5 \%$	$7,8 \pm 5 \%$

$R_m$  should take into consideration the resistance of the connection lines of the battery being tested and the contact resistance of the switch.

<sup>a</sup> Application with high peak current.  
<sup>b</sup> Method A (recommended test): requires specialised test equipment.  
<sup>c</sup> Method B: to be used in the absence of method A test equipment.  
<sup>d</sup> Method C: to be used only by agreement between the manufacturer and the purchaser.

**7.2.5 Calculation of the internal resistance  $R_i$** 

The internal resistance may be determined by the following calculation:

$$R_i = \frac{U_{oc} - U_{cc}}{U_{cc} / R_m}$$

NOTE The relation  $U_{cc} / R_m$  corresponds to the current delivered through the discharge resistance  $R_m$  (see 7.2.4).

**7.2.6 Measurement of the capacity****7.2.6.1 General**

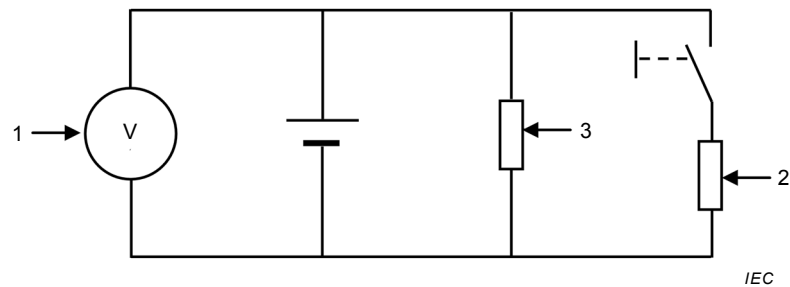
There are two methods for measuring capacity:

- the recommended method is method A, which is more indicative of watch requirements;
- method B is a more general method and is already specified in IEC 60086-1 and IEC 60086-2.

When presenting capacity data, the manufacturer shall specify which test method was used.

### 7.2.6.2 Method A

a) Circuitry principle (see Figure 7)



#### Key

- 1 reading  $U_{cc} / U'_{oc}$
- 2  $R_m$  resistance of measurement
- 3  $R_d$  resistance of continuous discharge

**Figure 7 – Circuitry principle for method A**

b) Procedure

The duration of the discharge test at the resistor  $R_d$  approximates to 30 days.

Value of the resistance  $R_d$ : the value of the resistive load (specified in Tables 7 and 8) shall include all parts of the external circuit and shall be accurate to within  $\pm 0,5 \%$ .

c) Determination of the capacity

The measurements of the open-circuit voltage  $U'_{oc}$  and that of the closed circuit voltage  $U_{cc}$  are carried out at least once a day on the battery permanently connected to  $R_d$ , until the first passage of the  $U_{cc}$  under the end-point voltage defined in Table 5 is obtained.

1) First measurement  $U'_{oc}$ : the resistance  $R_d$  being much higher than  $R_m$ ,  $U'_{oc}$  approximates to  $U_{oc}$ .

The switch is left open while the measurement is being carried out.

2) Next measurement  $U_{cc}$ : the battery being tested is connected to  $R_m$ . The switch is left closed during the duration  $\Delta t$  according to Table 7.

**Table 7 – Test method A for  $U_{cc}$  (CCV) measurement**

Batteries with KOH electrolyte		All other batteries	
$R_m$	$\Delta t$	$R_m$	$\Delta t$
$\Omega$	s	$\Omega$	ms
$150 \pm 0,5 \%$	$1 \pm 5 \%$	$1\ 500 \pm 0,5 \%$	$10 \pm 5 \%$

3) Calculation of the capacity  $C$ : the capacity of the battery is obtained by adding the partial capacity amounts  $C_p$ , calculated after each measurement with the following formula:

$$C_p = \frac{U'_{oc} \times t_i}{R_d}$$

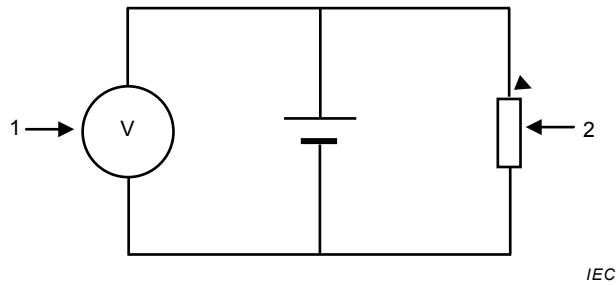
where  $t_i$  is the time between two measurements

$$C = \Sigma C_p$$

4) Near the end of discharge, it is recommended to carry out several measurements a day in order to obtain sufficient accuracy.

### 7.2.6.3 Method B

a) Circuitry principle (see Figure 8)



#### Key

- 1 reading  $U_{cc}$   
 2  $R_d$  resistance of continuous discharge

**Figure 8 – Circuitry principle for method B**

b) See procedure in 7.2.6.2 b).

c) Determination of the capacity: when the on-load voltage of the battery under test drops for the first time below the specified end point as specified in Table 5, the time  $t$  is calculated and defined as service life.

The capacity is calculated by the following formula:

$$C = \frac{U_{cc} \text{ (average)}}{R_d} t$$

where

$C$  is the capacity;

$U_{cc} \text{ (average)}$  is the average voltage value of  $U_{cc}$  during discharge duration time  $(0-t)$ ;

$t$  is the service life.

### 7.2.7 Calculation of the internal resistance $R_i$ during discharge in case of method A (optional)

After each measurement of  $U'_{oc}$  and  $U_{cc}$  is carried out according to the procedure described in 7.2.6, it is possible to calculate the internal resistance  $R_i$  of the battery using the following formula:

$$R_i = \frac{U'_{oc} - U_{cc}}{U_{cc} I R_m}$$

**Table 8 – Discharge resistance (values)**

Code number according to the dimensions	Letter for electrochemical systems		Code number according to the dimensions	Letter for electrochemical systems	
	L	S		C	B
	Discharge resistance kΩ			Discharge resistance kΩ	
416			1025	68	
421			1212		
510			1216	62	
512			1220	62	
514			1225		30
516		82	1612		
521		68	1616	30	
527		56	1620	47	
610			1625		
612			1632		
614		120	2012	30	
616		100	2016	30	30
621		68	2020	30	
626		47	2025	15	
710			2032	15	
712		100	2312		
714		68	2316		
716		68	2320	15	15
721		47	2325		15
726		33	2412		
731		27	2416		
736	22	22	2330	15	
754		15	2430	15	
910					
912					
914					
916		47			
920		33			
921		33			
927		22			
936		15			
1110					
1112					
1114					
1116		39			
1120		22			
1121	22	22			
1126		15			
1130	15	15			
1136		15			
1142	10	10			
1154	6,8	6,8			

NOTE Blank values under consideration.

### 7.3 Test methods for determining the resistance to leakage

#### 7.3.1 Preconditioning and initial visual examination

Before carrying out the tests specified in 7.3.2 and 7.3.3, the batteries shall be submitted to a visual examination according to the requirements stated in Clause 8.

For tests in 7.3.2.1 and 7.3.2.2, batteries shall be pre-stored at the specified temperature (40 °C and 45 °C respectively) for 2 h. Batteries shall be moved from the preconditioning (alternative pre-stored) chamber (or oven) into the high temperature and humidity test chamber within minutes in order to avoid cooling of the battery and the risk of condensation at elevated humidity.

#### 7.3.2 High temperature and humidity test

##### 7.3.2.1 Recommended test

The battery shall be stored under the conditions specified in Table 9.

**Table 9 – Storage conditions for the recommended test**

Temperature °C	Relative humidity %	Test time days
40 ± 2	90 to 95	30 or 90

The test time of 30 days may be used for an accelerated routine quality control test, whereas the test time of 90 days applies to qualification testing of new batteries.

##### 7.3.2.2 Optional test

After agreement between the manufacturer and purchaser, the following testing conditions may be chosen (see Table 10).

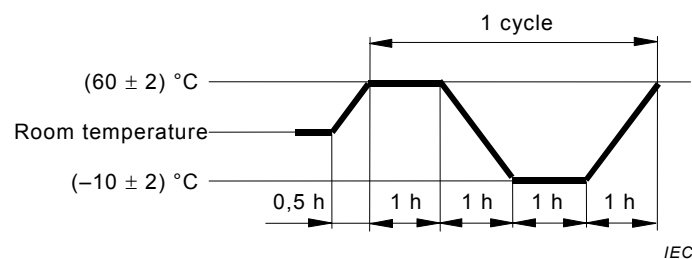
**Table 10 – Storage conditions for optional test**

Temperature °C	Relative humidity %	Test time days
45 ± 2	90 to 95	20 or 60

The test time of 20 days may be used for an accelerated routine quality control test, whereas the test time of 60 days applies to qualification testing of new batteries.

#### 7.3.3 Test by temperature cycles

The battery shall be submitted to 150 temperature cycles according to the schedule in Figure 9:



**Figure 9 – Test by temperature cycles**



## **8 Visual examination and acceptance conditions**

### **8.1 Preconditioning**

Before carrying out the initial visual examination or after the tests specified in Clause 7, the batteries shall be stored for at least 24 h at room temperature and at a relative humidity between  $55 \pm 20$  %.

The leakage should be observed after crystallisation of the electrolyte. The time of the storage of 24 h can be prolonged if necessary. This examination may be applied to new or used batteries, or to batteries which have been submitted to different tests.

### **8.2 Magnification**

The visual examination shall be carried out at a magnification of x15.

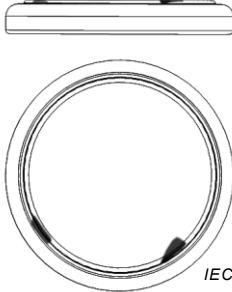
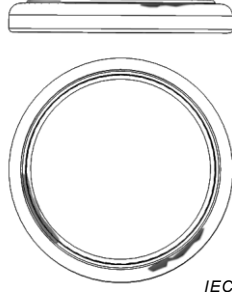
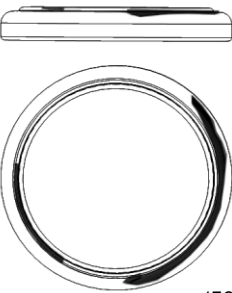

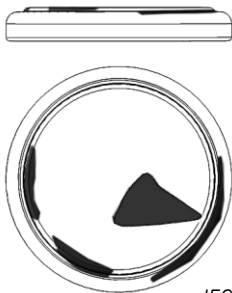
### **8.3 Lighting**

The visual examination shall be carried out under a diffuse white light of 900 lx to 1 100 lx at the surface of the battery to be inspected.

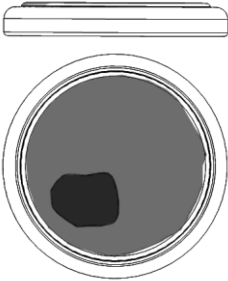
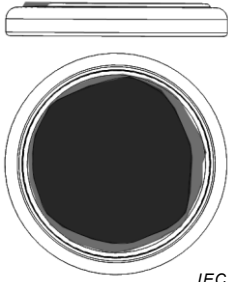
### **8.4 Leakage levels and classification**

The visual examination shall be carried out under a diffuse white light of 900 lx to 1 100 lx at the surface of the battery to be inspected. (See Table 11).

**Table 11 – Leakage levels and classification (1 of 2)**

Leakage levels		Diagram	Definition
Classification	Grade		
Salting	S1		<p>Little salting found near the gasket, affecting less than 10 % of the perimeter of the gasket, detected while observing at a magnification of x15. The leak is not detectable with the naked eye</p>
	S2		<p>Traces of salting near gasket can be detected with the naked eye. At a magnification of x15, it may be noted that these salts affect more than 10 % of the perimeter of the gasket</p>
	S3		<p>Salt spreads on both sides of the gasket can be detected with the naked eye, but do not reach the flat of the negative contact</p>
Clouds	C1		<p>Leaks spread in clouds on both sides of the gasket, do reach the flat of the negative contact but do not reach the central part of the flat negative contact</p>
	C2		<p>Leaks spread in clouds, which reach the central part of the flat negative contact</p>

**Table 11 (2 of 2)**

Leakage levels		Diagram	Definition
Classification	Grade		
Leaks	L1	 <p style="text-align: right;"><i>IEC</i></p>	The accumulation of crystallised liquid coming from the electrolyte swells up on part of the cloud spread, which covers the entire surface of the flat negative contact
	L2	 <p style="text-align: right;"><i>IEC</i></p>	The accumulation of crystallised liquid coming from the electrolyte swells up on the entire cloud spread, which covers the entire surface of the flat negative contact

### 8.5 Acceptance conditions

The acceptable level, as well as the proportion of defective pieces, shall be agreed between the manufacturer and the purchaser.

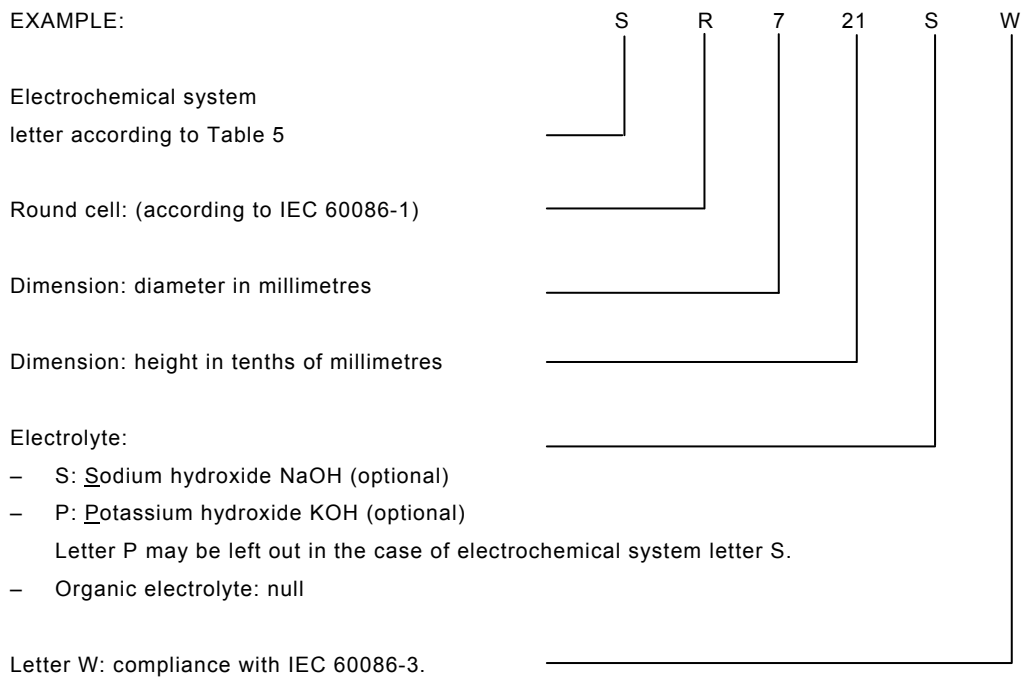
Fresh batteries, with a level of leakage exceeding S1, shall not be submitted for qualification. The acceptance criteria may be less restrictive for batteries which have been tested according to 7.3.2. If necessary, photographic references may be established.

## Annex A (normative)

### Designation

Watch batteries manufactured with the express purpose of complying with this standard should be designated by a system of coded letters and numbers as shown below. However, the letter W is used to indicate compliance with IEC 60086-3.

EXAMPLE:



## Bibliography

IEC 60068-2-78:2001, *Environmental testing – Part 2-78: Tests – Test Cab: Damp heat, steady state*

ISO 8601:2004, *Data elements and interchange formats – Information interchange – Representation of dates and times*

ISO 2859, *Sampling procedures for inspection by attributes package*

ISO 21747, *Statistical methods – Process performance and capability statistics for measured quality characteristics*

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