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Cosmetics — Analytical methods — GC/MS method for the identification and assay of 12 phthalates in cosmetic samples ready for analytical injection



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National foreword

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Cosmétiques - Méthodes analytiques - Méthode CG-SM pour l'identification et l'analyse de 12 phtalates dans des échantillons de produits cosmétiques prêts à être injectés dans un système analytique

Kosmetische Mittel - Analysenmethoden - GC/MS-Methode für die Identifizierung und die Quantifizierung von 12 Phthalaten in zur direkten Injektion geeigneten Proben kosmetischer Mittel

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BS EN 16521:2014 EN 16521:2014 (E)

Foreword

This document (EN 16521:2014) has been prepared by Technical Committee CEN/TC 392 "Cosmetics", the secretariat of which is held by AFNOR.

This European Standard shall be given the status of a national standard, either by publication of an identical text or by endorsement, at the latest by November 2014 and conflicting national standards shall be withdrawn at the latest by November 2014.

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Introduction

Phthalates are esters of phthalic acid (Figure 1). About 80 % of all phthalates manufactured are used as "plasticizers" to make plastics flexible without sacrificing strength or durability. These compounds are present in cosmetic products like perfumes and toiletries. Some phthalates, particularly those of low molecular weight, are introduced into cosmetics as ingredients, for examples DEP and DMP are used as solvents and perfume fixatives [1-3] or DEP can be used as alcohol denaturing [2, 4]. Their presence in such products may come from their use as ingredients during the manufacturing process or may come from the migration of phthalates from packaging when plastic is used. Their presence as contaminant could also be due to the manufacturing process or raw materials used. Some analytical methods are proposed in the literature for the determination of phthalates in cosmetic products [1, 4-12].

Figure 1 — Esters of phthalic acid

This standard proposes a GC/MS method for a simultaneous assay in cosmetic samples ready for analytical injection of 12 phthalates listed in Table 1. These chromatographic conditions are not suitable for the quantification of di-isononyl phthalate (DiNP) or di-isodecyl phthalate (DiDP). According to SCCP [13], the possible presence of DiNP or DiDP in cosmetics does not seem to be a problem for human health. A GC/MS method using positive chemical ionisation with ammonia as collision gas is proposed in literature for the determinations of those compounds in cosmetic products [14].

Analyses are carried out on a GC/MS system with electron impact ionization mode (Ei). The separation of phthalates is obtained on a cross-linked 5 %-phenyl/95 %-dimethylpolysiloxane capillary column 30 m \times 0,25 mm (i. d.) \times 0,25 µm film thickness using a temperature gradient. Phthalate quantification is performed by external calibration using an internal standard or by the standard addition. Cosmetic samples are analyzed directly or after a previous dilution in ethanol [15].

1 Scope

This European Standard describes a GC/MS method for the assay of 12 phthalates, amongst which the 8 phthalates regulated by the European cosmetic regulation 1223/2009 [16]. This method is given for the analysis of samples ready for analytical injection from cosmetic products or raw materials used in cosmetic products. Samples should be compatible with GC analysis possibly after dilution. This method does not include requirements for the preparation of samples in cosmetic matrices for which direct injection in GC is not feasible.

2 Reagents

If not otherwise specified, analytical-grade chemicals shall be used.

2.1 Phthalates considered

Table 1 — Phthalates considered

Phthalates	CAS	Manufacturer ^b	Quality
DBP ^a (dibutyl phthalate)	84–74–2	ALDRICH	97,0 %
DEHP ^a (diethylhexyl phthalate)	117–81–7	ALDRICH	99,8 %
BBP ^a (butylbenzyl phthalate)	85–68–7	ALDRICH	97,0 %
DMEP ^a (di(2-methoxyethyl) phthalate)	117–82–8	ALDRICH	97,0 %
DnPP ^a (di-n-pentyl phthalate)	131–18–0	CIL CLUZEAU	99,0 %
DiPP ^a (diisopentyl phthalate)	605–50–5	CIL CLUZEAU	95,0 %
DPP ^a (n-pentyl isopentyl phthalate)	84777–06–0	CIL CLUZEAU	95,0 %*
DiBP ^a (diisobutyl phthalate)	84–69–5	ACROS	98,0 %
DCHP (dicyclohexyl phthalate)	84–61–7	ALDRICH	98,0 %
DEP (diethyl phthalate)	84–66–2	ACROS	98,0 %
DMP (dimethyl phthalate)	131–11–3	ACROS	98,0 %
DnOP (di-n-octyl phthalate)	117–84–0	ALDRICH	98,0 %

a Regulated phthalates.

2.2 Ethanol

- 2.3 Internal standard, 4,4-Dibromodiphenyl from Fluka 1) 97,0 % was used as internal standard (ISTD).
- 2.4 Internal standard stock solution (SM-ISTD), $c = 1~000~\mu g/ml$.

b This is an example of a suitable product available commercially. This information is given for the convenience of users of this document and does not constitute an endorsement by CEN of this product.

^{*} Mixed isomers (DiPP, DnPP and DPP)

¹⁾ This is an example of a suitable product available commercially. This information is given for the convenience of users of this document and does not constitute an endorsement by CEN of this product.

Weigh approximately 10 mg of 4,4-Dibromodiphenyl (2.3) into a 10 ml volumetric flask. Firstly, dissolve in a small amount of ethanol (2.2) and then fill to the calibration mark with ethanol. This stock solution is daily prepared.

2.5 Phthalates stock solution, $c = 500 \mu g/ml$ (SM).

Weigh approximately 10 mg of each phthalate (2.1) into a 20 ml volumetric flask. Firstly, dissolve in a small amount of ethanol (2.2) and then fill to the calibration mark with ethanol. This stock solution has a shelf life of at least 2 weeks if stored in a refrigerator.

NOTE Due to the low amount of DIPP standard (sold in quantities of 10,0 mg), a stock solution of this phthalate is prepared independently from other phthalates. In this way, two intermediate stock solutions SM-1 and SM-2 at 1 000 μ g/ml are prepared. SM-1 is prepared weighting approximately 10 mg of each phthalate (2.1), except DiPP, into a 10,0 ml volumetric flask, whereas SM-2 is prepared weighting approximately 5,0 mg of DiPP (2.1) into a 5,0 ml volumetric flask. The final stock solution SM is obtained mixing equal volumes of both standard solutions ($c = 500 \mu$ g/ml). All these stock solutions have a shelf life of at least 2 weeks if stored in a refrigerator.

2.6 Calibration solutions (standard solutions)

1,0 ml of the phthalate stock solution SM (2.5) is transferred into a 10,0 ml volumetric flask and filled with ethanol (2.2) up to the calibration mark ($c = 50 \,\mu\text{g/ml}$). From this intermediate solution (S1), at least 5 calibration solutions are prepared by dilution in ethanol (2.2) after the addition of 100 μ l of the internal standard stock solution (SM-ISTD). Phthalates concentrations on these calibration solutions ranges from 0,25 μ g/ml to 5,0 μ g/ml with an ISTD concentration fixed at 10,0 μ g/ml. These calibration solutions are prepared extemporaneously and injected.

If cosmetic samples (perfume) are directly prepared in a 1,5 ml GC vial, according to 4.2, the preparation of calibration solutions have to be adapted: Calibration solutions, ranging from 0,25 μ g/ml to 5,0 μ g/ml, are prepared without internal standard and 10 μ l of the internal standard stock solution (SM-ISTD) are added to 1,0 ml of each calibration solution directly in a 1,5 ml GC vial. These calibration solutions are prepared extemporaneously. Vials are shaken and the solution injected.

3 Apparatus and equipment

- 3.1 Standard laboratory equipment.
- 3.2 Gas chromatography/mass spectrometry apparatus.
- 3.3 Gas chromatography/FID (for standard purity).
- 3.4 Analytical separation column.

The following parameters have proved useful:

GC column, low bleeding phase: 5 %-Phenyl-95 %-dimethylpolysiloxane, 30 m × 0,25 mm (i. d.) × 0,25 μ m or equivalent material.

4 Procedure

4.1 Standard purity

The purity of each standard and the internal standard and the respective percentages for geometrical isomers (n-pentyl isopentyl phthalate - DPP) shall be determined by GC-FID for further calculations.

4.2 Sample preparation

1,0 g of the cosmetic sample ready for analytical injection is transferred into a 10,0 ml volumetric flask and filled with ethanol (2.2) up to the calibration mark after the addition of 100 μ l of the internal standard stock solution (SM-ISTD). This solution is injected. In case of excessive concentration of phthalates, an appropriate previous dilution of the sample is performed. The limit of quantification using this sample preparation was set at 5 ppm.

An alternative sample preparation consists to prepare the cosmetic sample directly in a 1,5 ml GC vial. 10 μ l of the internal standard stock solution (SM-ISTD) are added to 1,0 ml of cosmetic samples (perfume). Vials are shaken and the solution injected. The limit of quantification obtained using this sample preparation was set at 0,5 ppm. Calibration solutions are prepared according to 2.6.

4.3 Gas chromatography (GC) measurement conditions

When using the apparatus (3.2) and column (3.4), the following conditions have shown to be useful:

Table 2 — GC/MS programme

Apparatus	Gas chromatography with mass selective detector (GC/MS) Autosampler GC column, low bleeding phase: 5 %-Phenyl-95 %- dimethylpolysiloxane, 30 m × 0,25 mm × 0,25 µm (or equivalent)				
Column					
Oven programme	Ramp Temperature Time				
· ·	-	100 °C	0 min		
	30 °C/min	200 °C	0 min		
	3 °C/min	260 °C	0 min		
	30 °C/min	320 °C	5 min		
Injector T	300 °C				
Interface T	250 °C				
Source T	230 °C 30 min He/1 ml/min				
Injection time					
Gas / flow rate					
Injection port		split/splitless			
Injection parameters :	1 μl/Constant pressure,				
	Split 1/20				
Detection mode		Quadrupole			
Ionisation mode		EI (70 eV)			
Mass detection	Identification: full-scan (m/z 40 to 350)				
(Full/SIM)	Quantification: SIM using 3 specific ions				
Internal standard	4,4-Dibromodiphenyl				
Calibration	0,25 μg/ml to 5,0 μg/ml				
Solvent used	olvent used Ethanol (injection)				

Examples of chromatogram, obtained using the GC/MS programme described in Table 2, are given in Annex A (Figure A.1).

4.4 Detection

4.4.1 General

The detection and quantitative determination can be performed by evaluating the mass traces of each phthalate. Relative Retention (RR), m/z ions used for quantification and ions ratios obtained on a standard solution for each phthalate are given in Table 3.

4.4.2 MS detection in Selected Ion Monitoring (SIM) mode

- 1) from 2,5 min lons 135;163; 194 (dwell time 50 ms/ion);
- 2) from 4,0 min lons 105; 149; 177 (dwell time 50 ms/ion);
- 3) from 5,0 min lons 104; 149; 223 (dwell time 50 ms/ion);
- 4) from 7,0 min lons 59; 104;149; 205; 223 (dwell time 30 ms/ion);
- 5) from 8,2 min lons 104; 149; 219, 237, 310, 312 (dwell time 25 ms/ion);
- 6) from 10,5 min lons 57; 71; 97 (dwell time 50 ms/ion);
- 7) from 13,0 min lons 91; 149; 206 (dwell time 50 ms/ion);
- from 16,5 min lons 149; 167; 225; 226, 249; 279, (dwell time 25 ms/ion);
- 9) from 20,0 min lons 149; 167; 279 (dwell time 50 ms/ion).

Retention times of each group should be determined after the injection of a standard solution.

NOTE 1 A dwell time at 50 msec per ion for a SIM window including three fragments is correct. If 6 ions are present, a 25 msec per ion dwell time can be applied. At least, a minimum of 10 points of acquisition for one peak allows an acceptable quantification.

Compound	Indicative RR	lon 1 (m/z)	lon 2 (m/z)	lon 3 (m/z)	lon ratio 2/1 % b	lon ratio 3/1 % b
DMP	0,43	163	194	135	6	6
DEP	0,52	149	177	105	25	10
DiBP a	0,78	149	104	223	5	2
DBP a	0,91	149	205	223	4	6
DMEP a	0,96	59	149	104	10	8
ISTD ^C	1,00	312	310	314	20	40
DiPP a	1,08	149	237	104	7	5
DPP a	1,15	149	237	219	5	2
DnPP a	1,23	149	237	104	6	4
BBP a	1,65	149	91	206	70	20
DCHP	2,04	149	167	249	30	5
DEHP a	2,11	149	167	279	50	30
DnOP	2,57	149	279	167	18	3

Table 3 — List of m/z ions used for quantification and ions ratio

The Relative Retention (RR) of each phthalate should be determined after the injection of a standard solution. The retention time of the ISTD (4,4 dibromodiphenyl) should be near 8,5 min. m/z ions used for quantification are bolded in the table, the 2 other m/z ions given in the table are qualifier ions used to confirm the identity of the component. Some phthalates show qualifier ions with very low intensity which leads to a weak specifity of the ion ratio (general issue for benzene derivatives such as phthalates).

Depending on the instrument, ion ratios may be determined independently from the GC/MS software according to the formula below:

Ion ratio $(m/z_{i,j}) = [(Area (m/z_i)/Area (m/z_i)] \cdot 100$

NOTE 2 During assay inject regularly solvent to check the absence of carry over.

5 Evaluation

5.1 Identification and quantitative determination

Phthalates are identified by comparing the retention times and the mass spectrum of the sample with those of calibration substances. The quantitative determination of the analyte is performed based on a calibration function or the standard addition. The calibration solutions are chromatographed in accordance with the conditions given in 4.2. Phthalate concentrations are calculated from the calibration by linear regression on the basis of the obtained peak areas. For DPP, the calibration curve is plotted for the major isomer. For DiPP and DnPP, the DPP isomers shall be considered.

a Regulated Phthalates.

b The maximum permitted tolerances for relative ion intensities are in agreement with document [17].

C ISTD: 4,4 dibromodiphenyl (CAS 92–86–4).

5.2 Calculation

The phthalate content, w, with respect to the sample, is calculated using Formula (1):

$$w = \frac{c \cdot V \cdot F}{m \cdot 10\ 000} \tag{1}$$

where

- w is the phthalate content, in g/100 g;
- c is the phthalate concentration in the sample solution, in μg/ml, determined from the calibration function;
- V is the volume of the sample solution, in ml;
- *F* is the dilution factor, if required;
- *m* is the initial weight of the sample, in g.

NOTE For information, validation data using this standard protocol were obtained using the analytical approach described in ISO 12787 and published in the Journal of Chromatography A, 1253 (2012) 144–153 [15]

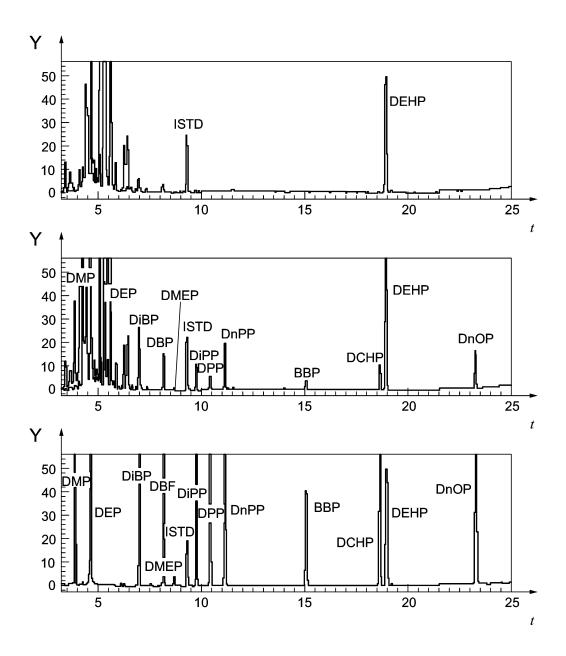
6 Test report

The test report shall contain the following data:

- a) information necessary for the identification of the sample (type, origin and designation of the sample);
- b) a reference to this European Standard;
- c) the date and type of sampling procedure (if known);
- d) the date of receipt and date of analysis;
- e) the date of test;
- f) the test results and the units in which they have been expressed;
- g) justification of any deviations from this official method;
- h) operations not specified in the method or regarded as optional, which might have affected the results;
- i) identification of laboratory that performed the test;
- j) identification, date and signature of the person responsible for the report.

Annex A (informative)

Example of Chromatograms



Key

Higher chromatogram: unspiked sample
Middle chromatogram: spiked sample

Lower chromatogram: 5,0 mg/l standard solution

Y abundance (u. a.) t time in min

Figure A.1 — Example of Chromatograms obtained using the GC/MS programme described in Table 2

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