

BS EN 15290:2011



BSI Standards Publication

# Solid biofuels — Determination of major elements — Al, Ca, Fe, Mg, P, K, Si, Na and Ti

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**National foreword**

This British Standard is the UK implementation of EN 15290:2011. It supersedes DD CEN/TS 15290:2006 which is withdrawn.

The UK participation in its preparation was entrusted to Technical Committee PTI/17, Solid biofuels.

A list of organizations represented on this committee can be obtained on request to its secretary.

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English Version

**Solid biofuels - Determination of major elements - Al, Ca, Fe,  
Mg, P, K, Si, Na and Ti**Biocombustibles solides - Dosage des éléments majeurs -  
Al, Ca, Fe, Mg, P, K, Si, Na et TiFeste Biobrennstoffe - Bestimmung von Hauptelementen -  
Al, Ca, Fe, Mg, P, K, Si, Na und Ti

This European Standard was approved by CEN on 25 December 2010.

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## Foreword

This document (EN 15290:2011) has been prepared by Technical Committee CEN/TC 335 “Solid biofuels”, the secretariat of which is held by SIS.

This European Standard shall be given the status of a national standard, either by publication of an identical text or by endorsement, at the latest by August 2011, and conflicting national standards shall be withdrawn at the latest by August 2011.

This document supersedes CEN/TS 15290:2006.

In the pre-normative project BIONORM I&II a robustness test has been performed to find out if all critical parameters in the standard were addressed. Based on the results of that test it has been concluded that all critical parameters were covered. Only minor technical changes were necessary which have been implemented in the revised text. The revision also includes a change of deliverable from Technical Specification to European Standard and updated normative references.

Attention is drawn to the possibility that some of the elements of this document may be the subject of patent rights. CEN [and/or CENELEC] shall not be held responsible for identifying any or all such patent rights.

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## Introduction

The elements described as major elements of solid biofuels are in fact major elements of the fuel ashes more than of the fuels. The determination of these elements may be used to assess ash behaviour in a thermal conversion process or to assess utilisation of ashes. Moreover, fuel contamination or process additives are indicated by high values of certain elements. Contamination of fuel with sand or soil is indicated by high values of several elements.

In this European Standard, wet chemical methods are described. As an alternative, X-ray fluorescence (XRF) may be used when validated with suitable materials (biomass reference materials).

## 1 Scope

This European Standard specifies methods for the determination of major elements of solid biofuels respectively of their ashes, which are Al, Ca, Fe, Mg, P, K, Si, Na, Ti. The determination of other elements such as barium (Ba) and manganese (Mn) is also possible with the methods described in this European Standard.

The European Standard includes two parts: Part A describes the direct determination on the fuel, this method is also applicable for sulfur and minor elements, Part B gives a method of determination on a prepared 550 °C ash.

## 2 Normative references

The following referenced documents are indispensable for the application of this document. For dated references, only the edition cited applies. For undated references, the latest edition of the referenced document (including any amendments) applies.

EN 14588:2010, *Solid biofuels — Terminology, definitions and descriptions*

EN 14774-3, *Solid biofuels — Determination of moisture content — Oven dry method — Part 3: Moisture in general analysis sample*

EN 14775, *Solid biofuels — Determination of ash content*

FprEN 14780, *Solid biofuels — Sample preparation*

EN 15296, *Solid biofuels — Conversion of analytical results from one basis to another*

EN ISO 7980, *Water quality — Determination of calcium and magnesium — Atomic absorption spectrometric method (ISO 7980:1986)*

EN ISO 11885, *Water quality — Determination of selected elements by inductively coupled plasma optical emission spectrometry (ICP-OES) (ISO 11885:2007)*

EN ISO 17294-2, *Water quality — Application of inductively coupled plasma mass spectrometry (ICP-MS) — Part 2: Determination of 62 elements (ISO 17294-2:2003)*

ISO 9964-1, *Water quality — Determination of sodium and potassium — Part 1: Determination of sodium by atomic absorption spectrometry*

ISO 9964-2, *Water quality — Determination of sodium and potassium — Part 2: Determination of potassium by atomic absorption spectrometry*

ISO 9964-3, *Water quality — Determination of sodium and potassium — Part 3: Determination of sodium and potassium by flame emission spectrometry*

## 3 Terms and definitions

For the purposes of this document, the terms and definitions given in EN 14588:2010 and the following apply.

**3.1 Reference Material**  
**RM**  
material or substance, one or more of whose property values are sufficiently homogeneous and well established to be used for the calibration of an apparatus, the assessment of a measurement method, or for assigning values to materials

**3.2 Certified Reference Material**  
**CRM**  
reference material, accompanied by a certificate, one or more of whose property values are certified by a procedure which establishes traceability to an accurate realisation of the unit in which the property values are expressed, and for which each certified value is accompanied by an uncertainty at a stated level of confidence

**3.3 NIST Standard Reference Material®**  
**SRM**  
CRM issued by NIST that also meets additional NIST-specific certification criteria and is issued with a certificate or certificate of analysis that reports the results of its characterisations and provides information regarding the appropriate use(s) of the material

## 4 Symbols and abbreviations

### 4.1 Symbols

- Al Aluminium
- Ca Calcium
- Fe Iron
- Mg Magnesium
- P Phosphorus
- K Potassium
- Si Silicon
- Na Sodium
- Ti Titanium

### 4.2 Abbreviations

- |         |  |
|---------|--|
| CRM     | Certified Reference Material                               |
| ICP-OES | Inductively Coupled Plasma – Optical Emission Spectrometry |
| ICP-MS  | Inductively Coupled Plasma – Mass Spectrometry             |
| FAAS    | Flame Atomic Absorption Spectrometry                       |
| FES     | Flame Emission Spectrometry                                |
| SRM     | Standard Reference Material                                |



## 5 Principle

The sample is digested in a closed vessel by the help of reagents, temperature and pressure. The digestion is either carried out directly on the fuel (part A) or on a 550 °C prepared ash (part B).

The detection of the elements may be done by ICP-OES, ICP-MS, FAAS or FES.

## 6 Reagents

### 6.1 General

All reagents should be of analytical grade or better. If minor elements are also to be determined, the best qualities should be used.

### 6.2 Water

Water containing negligible amounts of major elements, i.e. amounts that do not contribute significant to the determinations. Deionised water will normally fulfil this requirement.

### 6.3 Nitric acid (HNO<sub>3</sub>)

≥ 65 % (w/w), ρ = 1,41 g/ml

### 6.4 Hydrogen peroxide (H<sub>2</sub>O<sub>2</sub>)

30 % (w/w), ρ = 1,11 g/ml

### 6.5 Hydrofluoric acid (HF)

40 % (w/w), ρ = 1,13 g/ml

**CAUTION — Hydrofluoric acid may lead to health hazards.**

### 6.6 Boric acid (H<sub>3</sub>BO<sub>3</sub>)

4 % (w/w)

### 6.7 Use of Certified Reference Materials (CRM or SRM)

Use certified reference materials, issued by an internationally recognized authority, to check if the accuracy of the calibration meets the required performance characteristics. Examples of certified reference materials are: NBS 1570 spinach leaves, NBS 1571 orchard leaves, NBS 1573 tomato leaves and NBS 1575 pine needles.

When, due to matrix effects or concentration range limitations, no good recoveries for the certified reference materials can be obtained, calibration with at least two CRM or SRM materials may solve these problems. In that case CRM or SRM materials other than used for the calibration shall be used for verification purposes.

**NOTE** A CRM or SRM is prepared and used for three main purposes: (1) to help develop accurate methods of analysis; (2) to calibrate measurement systems used to facilitate exchange of goods, institute quality control, determine performance characteristics, or measure a property at the state-of-the-art limit; and (3) to ensure the long-term adequacy and integrity of measurement quality assurance programs.

## 7 Apparatus

### 7.1 Heating oven or heating block suitable for the decomposition system in use.

Resistance heated oven or heating block that can be used at a temperature of at least 220 °C with an accuracy of  $\pm 10$  °C.

### 7.2 Microwave oven.

Intended for laboratory use and equipped with temperature control.

### 7.3 Sample digestion vessels.

Intended for the heating system used, normally made of a fluoro plastic.

### 7.4 Balance.

Part A: Balance with a resolution of at least 1 mg.

Part B: Balance with a resolution of at least 0,1 mg.

### 7.5 Plastic volumetric flasks.

## 8 Preparation of the test sample

The test sample is the general analysis test sample with a nominal top size of 1 mm or less, prepared in accordance with FprEN 14780.

The moisture content of the test sample shall be determined as described in EN 14774-3.

## 9 Procedure

### 9.1 Digestion

#### 9.1.1 Part A: Direct determination on the fuel

The decomposition shall be carried out in closed vessels. It can be done in a heating oven, a heating block or in a microwave oven.

- a) Mix 500 mg of ground and homogenised sample, weighed to the nearest 1 mg, with 3,0 ml H<sub>2</sub>O<sub>2</sub> (30 %), 8,0 ml HNO<sub>3</sub> (65 %) and 1,0 ml HF (40 %) in a closed digestion vessel. A reaction time of minimum 5 min shall be kept before closing the vessel. Closing the digestion vessel too early can result in a fast pressure build up, sometimes exceeding the maximum pressure limit of the vessel.

NOTE 1 If the sample is expected to have an ash content above 10 %, 2,0 ml HF (40 %) should be used.

- b) The heating of the vessel shall not be too fast. Heat the sample according to the following heating programmes for digestion:

- 1) Resistance heating<sup>1)</sup>: Step 1: Ramp to 220 °C over 1 h

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1) The stated temperature refers to heating device (e.g. oven).

Step 2: Hold for 1 h at 220 °C

2) Microwave heating<sup>2)</sup>: Step 1: Ramp to 190 °C over 15 min

Step 2: Hold for 20 min at 190 °C

NOTE 2 If the maximum pressure limit of the vessel is exceeded during the digestion and by that an opening of the relief valve has occurred, the digestion should be discarded due to possible loss of Si (in form of gaseous SiF<sub>4</sub>).

NOTE 3 Some available digestion bomb systems use fluoropolymer vessels, which cannot withstand temperatures above 170 °C. In such cases this lower temperature can be used, provided that the sample is held longer at this temperature and that comparable results can be obtained, e.g. by the use of equivalent biomass reference materials.

c) After cooling to room temperature, HF is neutralised by adding 10 ml H<sub>3</sub>BO<sub>3</sub> (4 %).

NOTE 4 If 2,0 ml HF (40 %) was used for the digestion 20 ml H<sub>3</sub>BO<sub>3</sub> (4 %) should be used for the neutralisation.

d) Reheat the sample according to the following heating programmes for neutralisation:

1) Resistance heating<sup>1)</sup>: Step 1: Heat rapidly to 180 °C

Step 2: Hold for 15 min at 180 °C

2) Microwave heating<sup>2)</sup>: Step 1: Heat rapidly to 150 °C

Step 2: Hold for 15 min at 150 °C

e) After cooling, transfer the digest to a volumetric flask. Rinse the digestion vessel carefully and transfer the rinse solution to the volumetric flask. Add deionised water to the digest to an appropriate volume, depending on the detection method to be used.

### 9.1.2 Part B: Determination on a prepared 550 °C ash

a) Heat the sample according to the procedure described in EN 14775 to obtain ash. Take care that the ashing procedure is performed exactly according to this procedure as deviations in ashing temperature, time and air refreshing rate will influence the results. In deviation of EN 14775 only crucibles made of platinum or graphite can be used for the preparation of the ash, but larger types of crucibles may be used. The use of the stated additives in EN 14775 to ensure complete combustion is not allowed in the preparation. Also a continuous ashing by refilling of the sample on the previous ash in the crucible is not allowed.

NOTE 1 To prepare a sufficient amount of ash for the digestion larger amounts of sample, compared to the procedure given in EN 14775, will often be necessary.

The ash percentage on dry basis obtained for the prepared ash thus shall be calculated and compared to obtained results for the ash content on dry basis determined exactly according to EN 14775.

NOTE 2 If the ash content for the prepared ash is known also the results for major elements determined for the prepared ash may be calculated to fuel basis.

b) Homogenise the prepared ash in an agate mortar and reignite the homogenised ash at 550 °C for 30 min.

The weighing of the test portion of the ash for the digestion has to be carried out immediately after the preparation.

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2) The stated temperature refers to digest solution.

For the digestion of the ash similar working steps, as for the digestion of the fuel, are evident:

c) Mix 50 mg of ground and homogenised ash, weighed to the nearest 0,1 mg, with 2,0 ml H<sub>2</sub>O<sub>2</sub> (30 %), 3,0 ml HNO<sub>3</sub> (65 %) and 2,0 ml HF (40 %) in a closed decomposition vessel. A reaction time of minimum 5 min shall be kept before closing the vessel.

d) Digest the sample following one of the heating programmes described in 9.1.1 for digestion.

NOTE 3 If the maximum pressure limit of the vessel is exceeded during the digestion and by that an opening of the relief valve has occurred, the digestion should be discarded due to possible loss of Si (in form of gaseous SiF<sub>4</sub>).

e) After cooling to room temperature, the HF is neutralized by adding 20 ml H<sub>3</sub>BO<sub>3</sub> (4 %) and 10 ml deionised water.

NOTE 4 The water is necessary to keep K in solution for bio-ashes with high KCl content.

f) Reheat the sample according to the heating programmes for neutralisation described in 9.1.1.

g) After cooling, transfer the digest to a volumetric flask. Rinse the digestion vessel carefully and transfer the rinse solution to the volumetric flask. Add deionised water to the digest to an appropriate volume, depending on the detection method to be used.

## 9.2 Detection methods

For the detection of the concentrations of Al, Ca, Fe, Mg, P, K, Si, Na, Ti in the digests, the following methods can be used:

- ICP-OES according to the principles of EN ISO 11885;
- ICP-MS according to the principles of EN ISO 17294-2;
- AAS according to the principles of EN ISO 7980, ISO 9964-1 and ISO 9964-2;
- FES according to the principles of ISO 9964-3.

## 9.3 Calibration of the apparatus

When the analytical system is evaluated for the first time for this application, establish a calibration function for the measurement in accordance with the manufacturers' instructions. Adjust the established calibration function during the analysis if necessary. Check the performance of the instrument using the accepted standard procedures like replicate analysis, use of SRM and/or CRM, control samples and control charts. The calibration and quality control scheme shall be organized and maintained in such a way that the required uncertainty of measurement can be obtained. The results of the validation study of BioNorm2 (Annex B) demonstrates what is achievable with commercial instruments that are used by experienced laboratories.

## 9.4 Analysis of digests

Analyse test portions of the digests in accordance with the manufacturer's instructions.

## 9.5 Blank test

Carry out a blank test, using the same procedure and methods as described in 9.1.1, 9.1.2, 9.2, 9.3 and 9.4 but omitting the test portion. This assesses both the contents of the elements in the reagents and any contamination from equipment and the laboratory atmosphere. This contribution shall not be quantitatively significant.

NOTE A content of the elements in the digests of the blank experiment at 20 % or less of the content of the elements in the digests can be considered as not quantitatively significant.

## 10 Calculations

The content of an element in the sample on dry basis,  $w_i$ , expressed in mg/kg, is calculated from the mean of duplicate determinations using Equation (1):

$$w_i = \frac{(c_i - c_{i,0}) \times V}{m} \times \frac{100}{(100 - M_{ad})} \quad (1)$$

where

$w_i$  is the concentration of the element in the sample, on a dry basis, in mg/kg;

$c_i$  is the concentration of the element, in the diluted sample digest, in mg/l;

$c_{i,0}$  is the concentration of the element, in the solution of the blank experiment, in mg/l;

$V$  is the volume of the diluted sample digest solution, in ml;

$m$  is the mass of the test portion used, in g;

$M_{ad}$  is the moisture content in the analysis test sample in % m/m.

The results may be calculated to other bases e.g. to as received basis according to EN 15296.

If the determination has been carried out on a prepared ash (Part B) the results may be calculated to the fuel basis using Equation (2):

$$w_{i,fuel} = w_{i,ash} \times \frac{A_d}{100} \quad (2)$$

where

$A_d$  is the obtained ash content, concerning the prepared ash used for the digestion, in % m/m, dry basis;

$w_{i,fuel}$  is the concentration of the element in the fuel sample, on a dry basis, in mg/kg;

$w_{i,ash}$  is the concentration of the element in the prepared ashed sample, on a dry basis, in mg/kg.

## 11 Performance characteristics

The achievable performance of the method is given in Annex B showing the results obtained by a European intercomparison study carried out for a sample of wood chips and a sample of an exhausted olive residue. These two samples represent the extremity of the method. The wood chip sample represents samples with low contents of most of the elements and the olive residue samples with high amounts of most of the elements.

## 12 Test report

The test report shall contain at least the following information:

- a) identification of the laboratory performing the test and the date of the test;
- b) identification of product (sample) tested;
- c) reference to this European Standard (EN 15290);
- d) method used for determination;
- e) results of the test including the basis in which they are expressed, as indicated in Clause 10;
- f) any unusual features noted during the test procedure;
- g) any operation not included in this European Standard, or regarded as optional.

## Annex A (informative)

### List of conversion factors

The following list gives conversion factors for the calculation on the composition on an oxide basis in the case of determination on a prepared 550 °C ash.

Al → Al<sub>2</sub>O<sub>3</sub> 1,89

Ca → CaO 1,40

Fe → Fe<sub>2</sub>O<sub>3</sub> 1,43

Mg → MgO 1,66

P → P<sub>2</sub>O<sub>5</sub> 2,29

K → K<sub>2</sub>O 1,20

Si → SiO<sub>2</sub> 2,14

Na → Na<sub>2</sub>O 1,35

Ti → TiO<sub>2</sub> 1,67

## Annex B (informative)

### Performance data

The round robin was carried out by laboratories in Austria, Belgium, Denmark, Finland, Germany, Ireland, Italy, the Netherlands, Spain, Sweden and the United Kingdom. The variety of instruments and other analytical conditions were used in accordance with the quality parameters specified in the method.

The tests were carried out using two samples, wood chips and exhausted olive residues produced in the EU-project BioNorm in 2008 in accordance with CEN/TS 14780. The sample "wood chips" was made of German coniferous wood chips; the chips were dried and milled to 1 mm by means of cutting mill. The sample "exhausted olive residues" was obtained from olive oil industry in Spain from a typical outdoor storage facility. In the original sample stones and other natural impurities were present. These impurities and stones were removed manually and the sample was prepared from the residues in two steps using a coarse cutting mill equipped with a 10 mm sieve and a laboratory cutting mill equipped with WC cutting tools and a 1 mm sieve.

All data is reported on dry basis.

The performance data according to ISO 5725-2 [1] are presented in Tables B.1 to B.9.

NOTE 1 See Table B.1 for definition of the symbols used in Tables B.1 to B.9.

NOTE 2 A guideline can be found in EN 15296 on how to use these validation parameters.

**Table B.1 — Performance data for Aluminium (Al)**

Sample	$n$	$l$	$o$	$x$	$s_R$	$CV_R$	$s_r$	$CV_r$
			%	mg/kg	mg/kg	%	mg/kg	%
wood chips	12	57	5	47	8	18	2	4,1
exhausted olive residues	11	55	0	2 360	170	7,2	110	4,7
Definition symbols								
$n$	is the number of laboratories after outlier elimination							
$l$	is the number of outlier free individual analytical values							
$o$	is the percentage of outlying values from replicate determination							
$x$	is the overall mean							
$s_R$	is the reproducibility standard deviation							
$CV_R$	is the coefficient of the variation of the reproducibility							
$s_r$	is the repeatability standard deviation							
$CV_r$	is the coefficient of the variation of the repeatability							



Table B.2 — Performance data for Calcium (Ca)

Sample	<i>n</i>	<i>l</i>	<i>o</i>	<i>x</i>	<i>s<sub>R</sub></i>	<i>CV<sub>R</sub></i>	<i>s<sub>r</sub></i>	<i>CV<sub>r</sub></i>
			%	mg/kg	mg/kg	%	mg/kg	%
wood chips	11	54	1,8	1 500	100	6,6	24	1,6
exhausted olive residues	13	65	0	14 200	1 040	7,3	607	4,3

Table B.3 — Performance data for Iron (Fe)

	<i>n</i>	<i>l</i>	<i>o</i>	<i>x</i>	<i>s<sub>R</sub></i>	<i>CV<sub>R</sub></i>	<i>s<sub>r</sub></i>	<i>CV<sub>r</sub></i>
			%	mg/kg	mg/kg	%	mg/kg	%
wood chips	15	69	5,5	54	6	11,7	4	6,6
exhausted olive residues	15	73	1,4	1 600	165	10,3	81	5,1

Table B.4 — Performance data for Magnesium (Mg)

Sample	<i>n</i>	<i>l</i>	<i>o</i>	<i>x</i>	<i>s<sub>R</sub></i>	<i>CV<sub>R</sub></i>	<i>s<sub>r</sub></i>	<i>CV<sub>r</sub></i>
			%	mg/kg	mg/kg	%	mg/kg	%
wood chips	12	60	0	194	14	7,2	3	1,7
exhausted olive residues	13	65	0	3 140	243	7,7	149	4,7

Table B.5 — Performance data for Phosphorus (P)

Sample	<i>n</i>	<i>l</i>	<i>o</i>	<i>x</i>	<i>s<sub>R</sub></i>	<i>CV<sub>R</sub></i>	<i>s<sub>r</sub></i>	<i>CV<sub>r</sub></i>
			%	mg/kg	mg/kg	%	mg/kg	%
wood chips	11	53	3,6	74	5	6,7	2	3,4
exhausted olive residues	13	65	0	1 490	127	8,5	58	3,9

Table B.6 — Performance data for Potassium (K)

Sample	<i>n</i>	<i>l</i>	<i>o</i>	<i>x</i>	<i>s<sub>R</sub></i>	<i>CV<sub>R</sub></i>	<i>s<sub>r</sub></i>	<i>CV<sub>r</sub></i>
			%	mg/kg	mg/kg	%	mg/kg	%
wood chips	13	63	3,1	691	77	11,1	12	1,8
exhausted olive residues	11	52	5,5	24 500	1 560	6,4	468	1,9

**Table B.7 — Performance data for Silicium (Si)**

Sample	<i>n</i>	<i>l</i>	<i>o</i>	<i>x</i>	<i>s<sub>R</sub></i>	<i>CV<sub>R</sub></i>	<i>s<sub>r</sub></i>	<i>CV<sub>r</sub></i>
			%	mg/kg	mg/kg	%	mg/kg	%
wood chips	8	39	2,5	320	106	33	36	11,2
exhausted olive residues	10	49	4,0	10 040	1 230	12,2	769	7,7

**Table B.8 — Performance data for Sodium (Na)**

Sample	<i>n</i>	<i>l</i>	<i>o</i>	<i>x</i>	<i>s<sub>R</sub></i>	<i>CV<sub>R</sub></i>	<i>s<sub>r</sub></i>	<i>CV<sub>r</sub></i>
			%	mg/kg	mg/kg	%	mg/kg	%
wood chips	9	44	0	13	6	48	2	14
exhausted olive residues	10	49	0	171	40	23	9	5,4

**Table B.9 — Performance data for Titanium (Ti)**

Sample	<i>n</i>	<i>l</i>	<i>o</i>	<i>x</i>	<i>s<sub>R</sub></i>	<i>CV<sub>R</sub></i>	<i>s<sub>r</sub></i>	<i>CV<sub>r</sub></i>
			%	mg/kg	mg/kg	%	mg/kg	%
wood chips	9	43	4,4	5,5	0,40	7,4	0,32	5,9
exhausted olive residues	11	54	1,8	136	11	7,9	6	4,4

## Bibliography

- [1] ISO 5725-2, *Accuracy (trueness and precision) of measurement methods and results — Part 2: Basic method for the determination of repeatability and reproducibility of a standard measurement method*
- [2] ISO/TS 21748, *Guidance for the use of repeatability, reproducibility and trueness estimates in measurement uncertainty estimation*
- [3] NIST definitions: <http://ts.nist.gov/MeasurementServices/ReferenceMaterials/DEFINITIONS.cfm>
- [4] NIST Technical note 1297:1994, *Guidelines for Evaluating and Expressing the Uncertainty of NIST Measurement Results*, <http://www.nist.gov/physlab/pubs/tn1297/index.cfm>





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