Designation: E582 – 07 (Reapproved 2013)ε1

Standard Test Method for Minimum Ignition Energy and Quenching Distance in Gaseous Mixtures[1]

This standard is issued under the fixed designation E582; the number immediately following the designation indicates the year of original adoption or, in the case of revision, the year of last revision. A number in parentheses indicates the year of last reapproval. A superscript epsilon (ε) indicates an editorial change since the last revision or reapproval.

ε1 NOTE—Warning notes were editorially updated throughout in October 2013.

1. Scope

1.1 This test method covers the determination of minimum energy for ignition (initiation of deflagration) and associated flat-plate ignition quenching distances.[2] The complete description is specific to alkane or alkene fuels admixed with air at normal ambient temperature and pressure. This method is applicable to mixtures of the specified fuels with air, varying from the most easily ignitable mixture to mixtures near to the limit-of-flammability compositions.

1.2 Extensions to other fuel-oxidizer combinations, and to other temperatures and pressures can be accomplished with all the accuracy inherent in this method if certain additional conditions are met: (*a*) mixture stability and compatibility with bomb, seal, and other materials is established through time tests described in Section 9; (*b*) the expected peak pressure from the test is within the pressure rating of the bomb (established as required by the particular research laboratory); (*c*) spark breakdown within the bomb is consistent with Paschen's law for the distance being tested; (*d*) the temperature, including that of the discharge electrodes, is uniform; and (*e*) if the temperature is other than ambient, the energy storage capacitance required is less than about 9 pF.

1.3 This method is one of several being developed by Committee E27 for determining the hazards of chemicals, including their vapors in air or other oxidant atmospheres. The measurements are useful in assessing fuel ignitability hazards due to static or other electrical sparks. However, the quenching distance data must be used with great prudence since they are primarily applicable to the ignition stage and therefore, represent values for initial pressure and not the smaller values existing at higher pressures.

1.4 The values stated in inch-pound units are to be regarded as standard. The values given in parentheses are mathematical conversions to SI units that are provided for information only and are not considered standard.

1.5 *This standard should be used to measure and describe the properties of materials, products, or assemblies in response to heat and flame under controlled laboratory conditions and should not be used to describe or appraise the fire hazard or fire risk of materials, products, or assemblies under actual fire conditions. However, results of this test may be used as elements of a fire risk assessment which takes into account all of the factors which are pertinent to an assessment of the fire hazard of a particular end use.*

1.6 *This standard does not purport to address all of the safety concerns, if any, associated with its use. It is the responsibility of the user of this standard to establish appropriate safety and health practices and determine the applicability of regulatory limitations prior to use.* Specific safety precautions are listed in Section 5.

2. Terminology

2.1 *Definitions:*

2.1.1 *ignition, n*—the initiation of combustion.

2.1.2 *minimum ignition energy (MIE), n*—electrical energy discharged from a capacitor, which is just sufficient to effect ignition of the most easily ignitable concentration of fuel in air under the specific test conditions.

2.2 *Definitions of Terms Specific to This Standard:*

2.2.1 *ignition quenching distance, n*—maximum spacing between eletrode flanges that will not permit spark ignition and flame propagation beyond the flanges, when tested under the specified test conditions.

3. Significance and Use

3.1 The minimum energies provide a basis for comparing the ease of ignition of gases. The flatplate ignition quenching distances provide an important verification of existing minimum ignition energy data and give approximate values of the propagation quenching distances of the various mixtures. It is emphasized that maximum safe experimental gaps, as from

[1] This test method is under the jurisdiction of ASTM Committee E27 on Hazard Potential of Chemicals and is the direct responsibility of Subcommittee E27.04 on Flammability and Ignitability of Chemicals.

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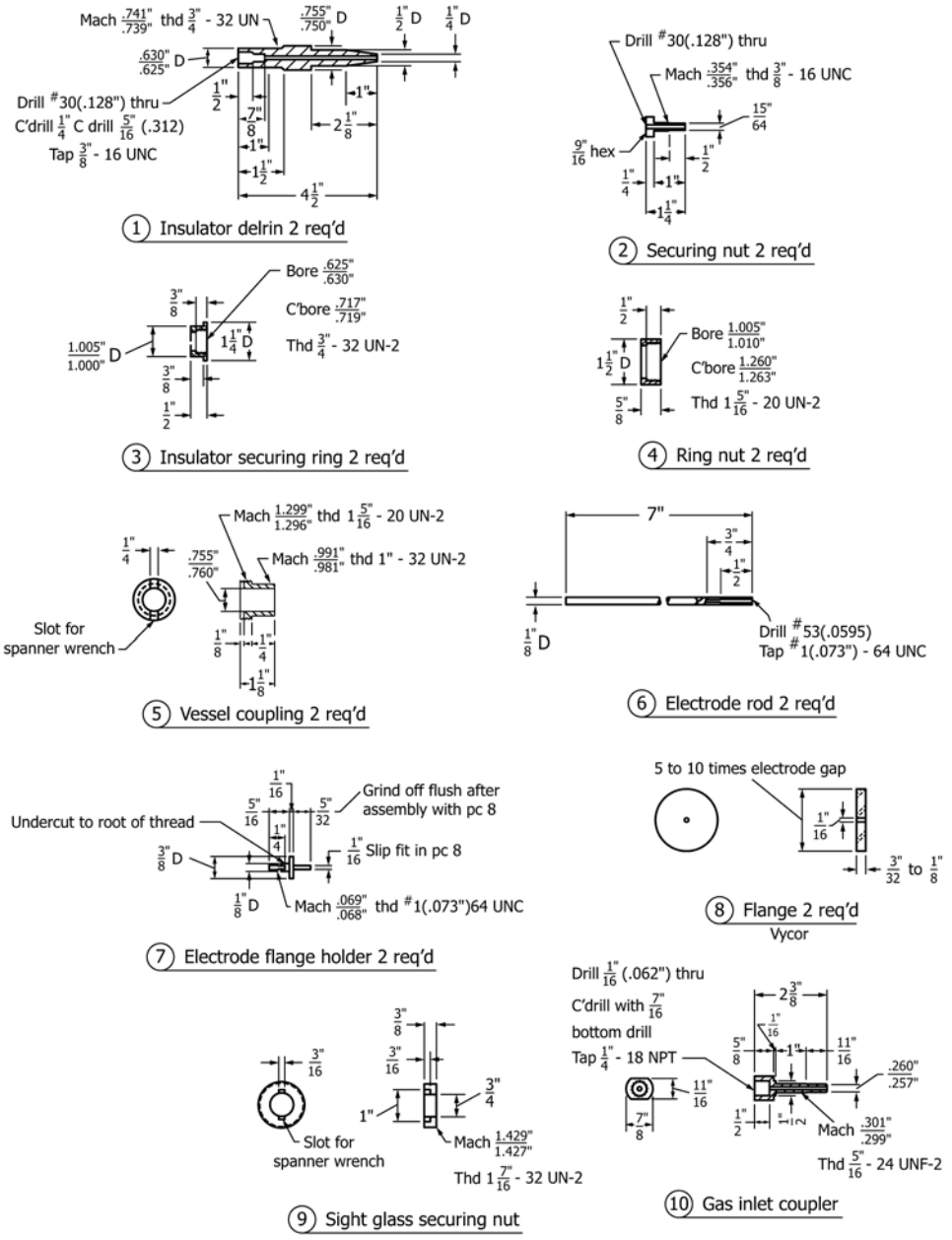
[2] Litchfield, E. L., Hay, M. H., Kubala, T. S., and Monroe, J. S., "Minimum Ignition Energy and Quenching Distance in Gaseous Mixtures," *BuMines*, R. L. 7009, August 1967, p. 11.

"flame-proof" or "explosion-proof" studies, are less than the flat-plate ignition quenching distances.

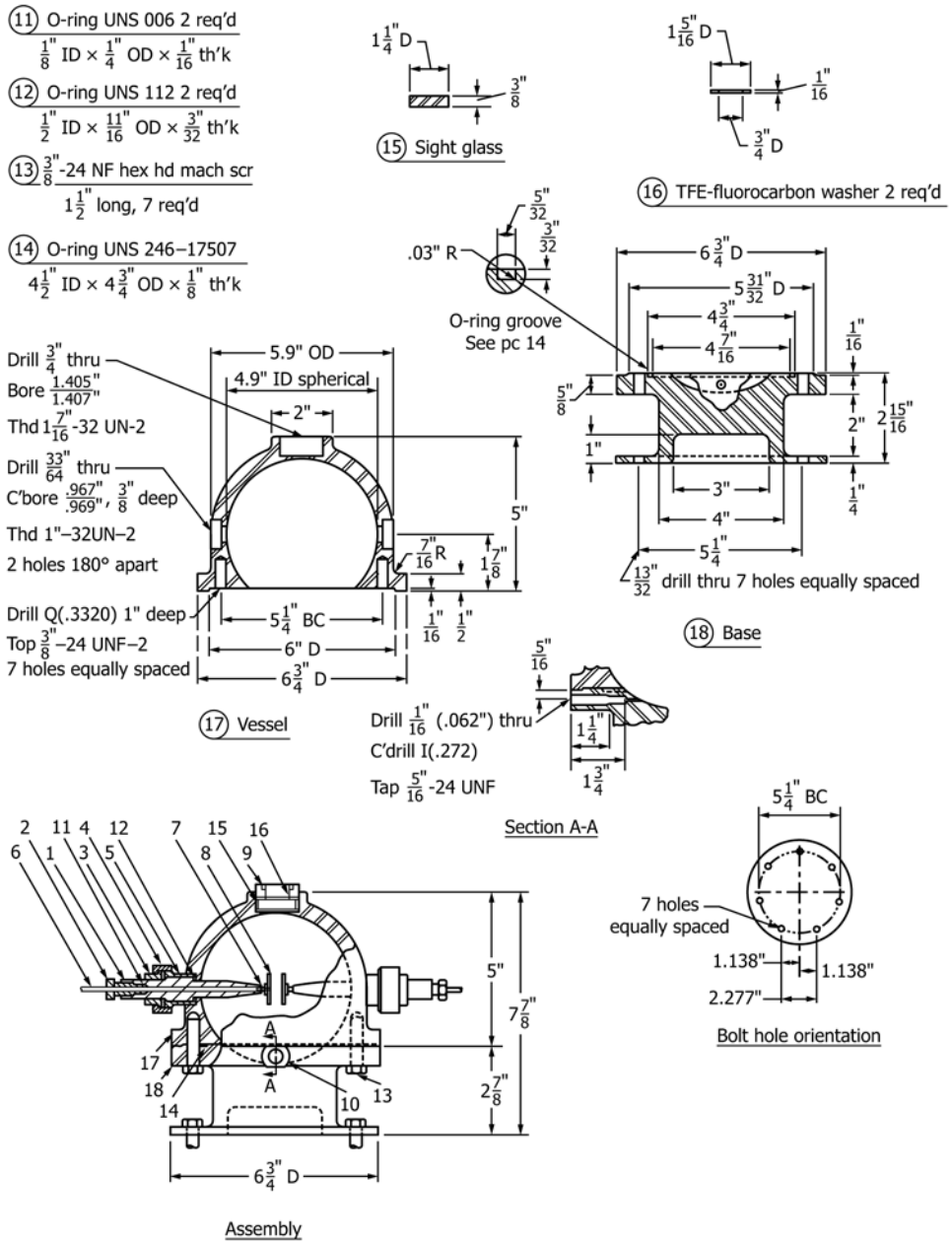
4. Apparatus

4.1 *Reaction Vessel*—The recommended reaction vessel is manufactured according to the specifications of Fig. 1 and Fig. 2. This is a spherical vessel, manufactured of Type 304 stainless steel, and passivated after machining. The spherical geometry maximizes the useable spark-gap length for a given vessel volume. The reaction vessel provides for opposed mounting of the spark electrodes which permits rapid and convenient variation of the gap length without the necessity for opening the vessel. The input orifice (Fig. 2, Section *A-A*) is located so that the gases are introduced approximately tangentially to the vessel walls, thus providing a turbulent swirling motion that facilitates mixing. A sight glass permits direct observation of flame initiation and propagation throughout the reaction volume.



NOTE 1—Tolerance is ±0.010 in., unless noted.
NOTE 2—Break all sharp edges.
NOTE 3—Material is Type 304 stainless steel.
NOTE 4—Thread depth is 75 to 80 %.
NOTE 5—1 in. = 25.4 mm.

FIG. 1 Electrode Assembly (I)



NOTE 1—Tolerance is ±0.010 in., unless noted.
NOTE 2—Break all sharp edges.
NOTE 3—Material is Type 304 stainless steel.
NOTE 4—Thread depth is 75 to 80 %.
NOTE 5—1 in. = 25.4 mm.

FIG. 2 Electrode Assembly (II).

4.2 *Electrode Assembly:*

4.2.1 The electrodes (Fig. 1) have metal tips flanged with glass plates. The tips screw into ⅛-in. stainless steel rods which extend through inserts in the bomb walls to permit external electrical connections. Gas seals are provided between the reaction vessel and the inserts and between the inserts and the ⅛-in. rods by O-ring seals (see Fig. 2, Assembly). The glass flange material should be either borosilicate or high silica and the flanges should be fastened to the stainless steel tips with a thin layer of epoxy cement. The facing surfaces should be planar and coplanar to 0.001 in. (0.025 mm) or 1 % of the intended test gap, whichever is larger.

4.2.2 Two inserts are required to carry the ⅛-in. rods through the walls of the reaction vessel. At least one of these inserts must be made of high-electrical resistivity insulating material. Hard rubber, phenolic plastic, poly(methyl methacrylate) (PMMA), and many other materials are suitable for use with the alkane and alkene fuels. In the excepted cases (other similarly energetic fuels), the insulating material must not react with or absorb the fuel being tested.

4.2.3 Where the test arrangement is optimized through the use of a "double-ended'' power supply, (see Fig. 3(b)) two insulating inserts are required. Otherwise, one of the inserts may be machined from Type 304 stainless steel.

4.2.4 Insulation between the two electrodes should exceed 10^{12} Ω as discussed in 4.3.3.

4.2.5 Measurement of the gap width is made by available techniques and implements most suitable for the gap distance being measured. Calibrated leaf gages, inside micrometers, or vernier calipers are suitable, depending upon the gap distance. The measurements should be made with a repeatability of ±0.001 in. (0.025 mm) or 1 %, whichever is most conservative. To facilitate such measurements, it is helpful to have leaf gages of known thicknesses for frequently used gap distances. High-quality machinist's micrometers will generally provide adequate accuracy.

4.3 *Power Supply and Electrical Circuit:*

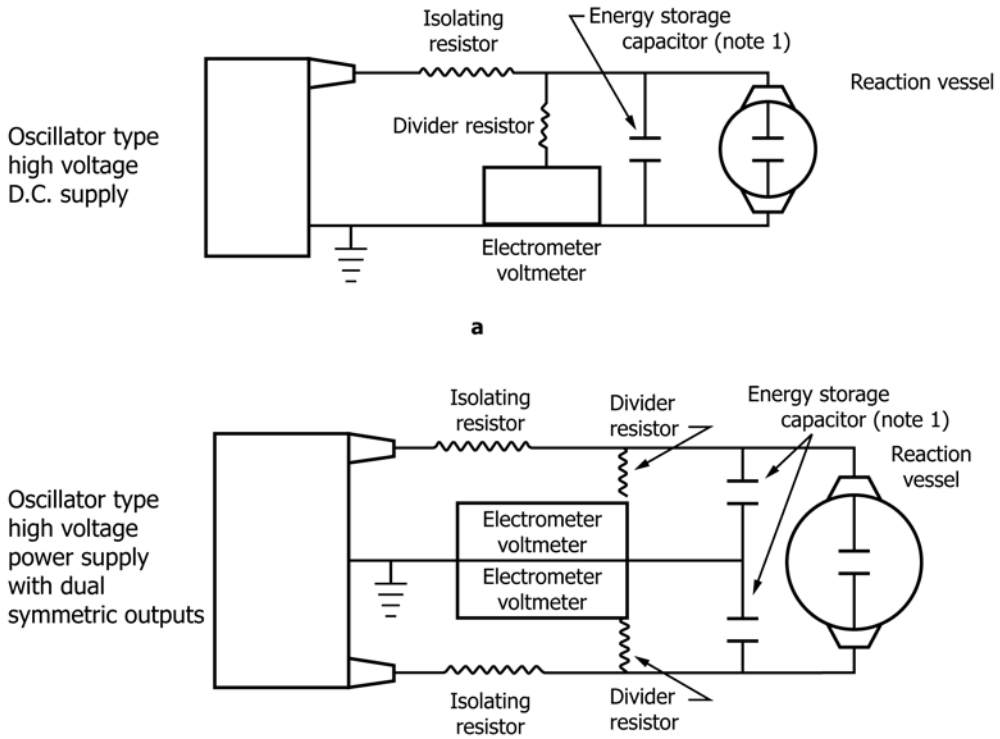
4.3.1 The power supply should be of the oscillator type, so that its filter condensers will be electrically small. The maximum output current should be about 1 mA. (**Warning—**With such a power supply, the probability of lethal shock to the operator from the high-voltage circuits becomes negligible. However, all usual and normal hazards to personnel will exist on the 60-Hz supply, main-side of the power supply.)

4.3.2 The power supply can be single-side with one high-voltage output terminal and one low-voltage, neutral, or ground terminal (see Fig. 3(a)). Alternatively, the power supply may be double-ended with two high-voltage output terminals, one negative and one positive, together with a center-tapped grounding or neutral connection (see Fig. 3(b)). For maximum testing flexibility, the power supply should deliver variable or adjustable output voltage differences between 1 and 30 kV.

NOTE 1—The double-ended power supply should be used only in conjunction with two insulating inserts. The metal bomb structure must then be connected to the power supply center point and connected to system ground. The double-ended power supply gives somewhat higher gap breakdown voltages at larger spark gaps and, thus, somewhat lower ignition energies. This consideration should be of importance only if the very highest quality data are required.

4.3.3 The output filter capacitors of the power supply must be isolated from the discharge energy storage capacitance by an isolating resistor. The resistive-capacitive time constant of the charging circuit containing the energy storage capacitance should be several seconds; 10^{12} Ω is a desirable value for the most easily ignitable mixture (energy storage capacitance of 8 to 12 pF) with the value reduced inversely as the energy storage capacitance is increased for less easily ignitable mixtures. Two resistors should be used in series, four with the double-ended supply. One resistor shall be immediately at the power supply terminal, the other at the bomb energy storage capacitance. Supply-line electrical insulation needs to be greater than 10^{14} Ω to be consistent with 10^{12} Ω series resistance. Such resistance is most easily achieved through air insulation with appropriately rounded corners to reduce corona losses.

4.4 *Measurement of Energy Storage Capacitance*—The energy storage capacitance may be measured with a high-quality capacitance meter capable of accurate measurement in the range of 5 pF, or greater, capacity. Lower frequency instruments are generally preferred, since problems of lead length



NOTE 1—Distributed capacitance must be considered as part of the energy storage capacitance.

NOTE 2—See 4.3 for component value guidelines.

FIG. 3 Connections of Single and Double-Sided Power Supply in Circuit.

and spurious readings are minimized. If the capacitance meter is nulled with the test probe at some distance from the bomb, proximity effects will be observed as the test probe is brought closer to the bomb and energy-storage capacitance. These proximity effects must be nulled out to achieve accurate energy storage capacitance determinations.

4.5 *Measurement of Energy Storage Voltage*—Energy storage voltage cannot ordinarily be measured with an electrostatic voltmeter for the most easily ignitable mixtures, since the meter capacitance will probably exceed the desired energy storage capacitance. In these instances, energy storage voltage must be measured with an electrometer voltmeter used in conjunction with a voltage divider network. Total resistance of the divider network should be at least 10^{14} Ω.

4.6 *Gas-Handling System*—Plumbing for the gas-handling system should be of stainless steel. A vacuum gage in the system is necessary and access to ambient temperature and barometric data is desirable. Gas shut-off valves should be of good quality and suitable to the service.

5. Safety Precautions

5.1 If the recommendations of 4.3 are followed, there are no high-voltage hazards. The normal 60-Hz supply line hazards exist as always when equipment is operated from such lines. In the interest of stability, it is desirable to have the high-voltage power supply and electrometer voltmeter turned-on continuously during a normal working period. In this case, the voltage output control of the power supply should be turned to its minimum value except during the actual test. The two spark electrodes should be shorted together by two clips and a short piece of flexible braid. If two insulating inserts are used (4.2), the center of the braid should be connected to the metal of the reaction vessel.

5.2 Normal pressure vessel problems exist, therefore, the sight glass must be in good condition and adequately seated; and the bolts holding the two parts of the bomb together and the insert-retaining rings must be in place and tightened. Care must be given to the preparation of the gas mixture. The two components must not be permitted to mix in the supply lines, both for reasons of personnel safety and to ensure a mixture of known composition.

6. Preparation of Apparatus

6.1 Clean the test vessel to remove residues from previous tests or if required by 6.2.

6.2 Verify that the insulation resistance of high-voltage electrode(s) to the reaction vessel walls is at least 10^{12} Ω . Disassemble and clean if required.

6.3 Set up the desired test gap in the reaction vessel.

6.4 Determine the desired energy storage capacitance (see Section 8) and set up this value with calibrated electrical capacitors rated for the anticipated test voltage. The smaller values of capacitance should be trimmed with lengths of wire or metal rods.

7. Procedure

7.1 Evacuate the reaction vessel to absolute pressures below 10 mmHg (13.33 Pa).

7.2 Load the desired gas mixture, as determined from the calculations, observing the necessary safety considerations of Section 5. First, add the minor constituent (fuel) to the desired partial pressure of fuel. Close the main valve to the reaction vessel and evacuate the supply lines. Introduce the major component (air) into the supply lines to a pressure greater than that of the fuel in the reaction vessel. Open the main valve and fill the reaction vessel to the desired total pressure, then close the main valve.

NOTE 2—This method of mixing, in conjunction with the mixing orifice in the reaction vessel of Section *A-A* of Fig. 2, offers maximum uniformity and accuracy in the prepared gas mixture. When used to prepare vapor-oxidizer mixtures, this method minimizes the possibility of vapor condensation.

NOTE 3—Experimenters desiring to use other techniques of mixture preparation and sample injection should bear the responsibility of establishing the adequacy of those techniques.

7.3 Remove the electrode shunting connections.

7.4 Gradually increase the applied voltage until a spark occurs between the spark gap electrodes. Immediately reduce the applied voltage. If no ignition is observed, repeat this for about five sparks. If ignition is not observed then proceed as follows:

7.4.1 If breakdown voltage is in accordance with expectations, either increase the energy storage capacitance or the spark gap length, depending upon the purpose of the test, and then repeat 7.4.

7.4.2 If breakdown voltage is not in accordance with expectations, discard the gas mixture, check the test gap, and rework the electrode surfaces, if necessary. Then repeat from 7.2.

7.5 When starting with fuel of unknown ignition characteristics, make initial trials with gaps believed to be in excess of the ignition quenching distance and with storage energies believed to be about ten times the ignition energy of the mixture. Reduction of the gap in successive tests will then establish approximately the ignition quenching gap. Ignitions should be obtained with ease at gaps in excess of the ignition quenching gap and disappear abruptly (usually over a distance of 0.001 to 0.002 in. (0.02 to 0.05 mm)) when the quenching gap is reached. Successive tests, at a gap slightly in excess of the ignition gap will then approximate the minimum ignition energy of the mixture.

8. Calculations

8.1 Calculate the partial pressure of the fuel, F, as follows:

8.1.1 For a 1-atm absolute mixture of fuel and air, the partial pressure of the fuel is as follows:

U.S. Customary Units:

$$F = B/100 \times 760 \quad (1)$$

SI Equivalents:

$$F = B/100 \times 101.3 \quad (2)$$

where:

F = partial pressure of fuel, mmHg or kPa, and
B = volume % of fuel.

8.2 Calculate the time constant, T, of the resistive-capacitive charging circuit as follows:

$$T = RC \quad (3)$$

where:

T = time, s,
R = resistance, Ω, and
C = capacitance, F.

8.3 Calculate the ignition energy, E, stored on the capacitor by an impressed voltage as follows:

$$E = ½\ CV^2 \quad (4)$$

where:

E = ignition energy, J,
C = capacitance, F, and
V = voltage, V.

9. Additional Considerations

9.1 Wherever the fuel partial pressure in the mixture exceeds about 80 % of the vapor pressure at ambient temperature, or where the equally energetic other fuels (1.2) are being investigated, confirmatory tests are extremely important. The corroborative tests are designed to ensure that the minimum ignition energy and flat-plate ignition quenching distances are independent of the time the mixture is held within the reaction vessel before the test is actually conducted. Maximum holding times for these corroborative tests should be about 2 h when the concern is possible fuel condensation; and about ten times the normal preparation and test interval when the intent is to demonstrate that there is no degradation of the test mixture by ambient temperature oxidation reactions.

10. Report

10.1 The report should specify the following:

10.1.1 Concentration or stoichiometry of the gas or gas mixture,

10.1.2 Test temperature,

10.1.3 Minimum ignition energy stored on the capacitor,

10.1.4 Minimum ignition energy calculated from the measured voltage and current at the spark gap, if available,

10.1.5 Ignition quenching distance, and

10.1.6 Any changes from standard procedures.

11. Precision and Bias

11.1 Available indications are that reproducibility (and presumably accuracy) of ±10 % in minimum ignition energy and ±2 % in ignition quenching distance are readily achieved in those instances where results are relatively independent of fuel-oxidizer concentrations. For mixture compositions approaching the limits of flammability, the minimum ignition energy varies so rapidly with mixture composition that reproducibility of the test mixture becomes the limiting and controlling parameter (it is possible that a 0.1 % variation in fuel concentration could change the ignition energies by factors of 100 to 1000). The bias of this method has not been established.