



# Standard Test Method for Hydrogen Sulfide in Liquefied Petroleum (LP) Gases (Lead Acetate Method)<sup>1</sup>

This standard is issued under the fixed designation D2420; the number immediately following the designation indicates the year of original adoption or, in the case of revision, the year of last revision. A number in parentheses indicates the year of last reapproval. A superscript epsilon ( $\epsilon$ ) indicates an editorial change since the last revision or reapproval.

## 1. Scope\*

1.1 This test method<sup>2</sup> covers the detection of hydrogen sulfide in liquefied petroleum (LP) gases. The sensitivity of the test is about 4 mg/m<sup>3</sup> (0.15 to 0.2 grain of hydrogen sulfide per 100 ft<sup>3</sup>) of gas.

1.2 The values stated in SI units are to be regarded as standard. The values given in parentheses are for information only.

1.3 *This standard does not purport to address all of the safety concerns, if any, associated with its use. It is the responsibility of the user of this standard to establish appropriate safety and health practices and determine the applicability of regulatory limitations prior to use.*

## 2. Referenced Documents

2.1 *ASTM Standards:*<sup>3</sup>

[D1193 Specification for Reagent Water](#)

[D1265 Practice for Sampling Liquefied Petroleum \(LP\) Gases, Manual Method](#)

[D1835 Specification for Liquefied Petroleum \(LP\) Gases](#)

2.2 *GPA Standard:*<sup>4</sup>

[GPA 2140 Liquefied Petroleum Gas Specifications and Test Methods](#)

## 3. Summary of Test Method

3.1 Vaporized LP gas is passed over moist lead acetate paper under controlled conditions. Hydrogen sulfide reacts with lead acetate to form lead sulfide which produces a coloration on the

paper varying from yellow to black, depending upon the amount of hydrogen sulfide present.

## 4. Significance and Use

4.1 Liquefied petroleum gases and their products of combustion must not be unduly corrosive to the materials with which they come in contact. The potential personnel exposure hazards of H<sub>2</sub>S also make the detection and measurement of hydrogen sulfide important, even in low concentrations. In addition, in some cases the odor of the gases shall not be objectionable. (See Specification [D1835](#) and GPA 2140.)

## 5. Interferences

5.1 Methyl mercaptan, if present, produces a transitory yellow stain on the lead acetate paper that will fade completely in less than 5 min.

5.2 Other sulfur compounds present in liquefied petroleum gas do not interfere with the test.

## 6. Apparatus and Materials

6.1 *Apparatus for Detecting Hydrogen Sulfide in Liquefied Petroleum Gas*, as shown in [Fig. 1](#) or commercially available apparatus.

6.2 *Lead Acetate Test Paper*—Commercially available test paper is satisfactory. Alternatively, the test paper may be prepared by dipping strips of *smooth* filter paper in 5 % aqueous lead acetate solution and removing the excess solution with a clean blotter. The strips of paper shall be approximately 50 mm long and 10 mm wide and contain a 3 mm hole near one end so they will hang freely vertical in the test apparatus.

6.3 *Gas Flow Indicator*—Capable of measuring flow rates in the range of 2 to 3 L of gas per minute. A wet test meter rotameter and mass flow controller have been found to be suitable.

6.4 *Water*—Unless otherwise indicated, reference to water shall be understood to mean reagent water that meets the requirements of Type II or III of Specification [D1193](#), or distilled water.

6.5 *Tubing*—High-pressure connecting tubing to connect the test apparatus to the sample collection point. Internally clean, non-reactive and non-absorptive stainless steel material

<sup>1</sup> This test method is under the jurisdiction of ASTM Committee [D02](#) on Petroleum Products and Lubricants and is the direct responsibility of Subcommittee [D02.H0](#) on Liquefied Petroleum Gas.

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<sup>2</sup> This test method is based on Edwards, J. D., and McBride, R. S., "Lead Acetate Test for Hydrogen Sulphide in Gas," *Technologic Papers T41*, National Institute for Standards and Technology, Aug. 9, 1914.

<sup>3</sup> For referenced ASTM standards, visit the ASTM website, [www.astm.org](http://www.astm.org), or contact ASTM Customer Service at [service@astm.org](mailto:service@astm.org). For *Annual Book of ASTM Standards* volume information, refer to the standard's Document Summary page on the ASTM website.

<sup>4</sup> Available from Gas Processors Association (GPA), 6526 E. 60th St., Tulsa, OK 74145, [www.gpaglobal.org](http://www.gpaglobal.org).

\*A Summary of Changes section appears at the end of this standard

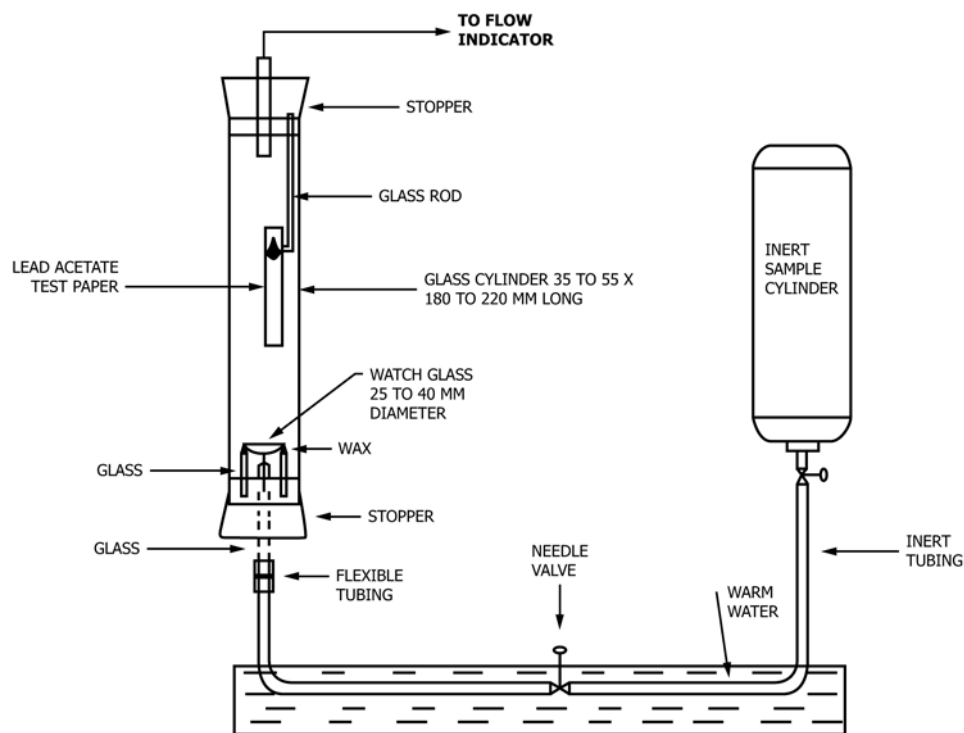


FIG. 1 Typical Apparatus for Detecting Hydrogen Sulfide in Liquefied Petroleum Gas

is recommended, but other materials that safely contain the pressure, resist being cut or torn, and have been tested to be inert relative to the stream being tested may be utilized.

6.5.1 Internal surfaces of sample containers, lines and fittings may be surface coated with an inert material to reduce bare metal surfaces which could react with, or absorb, trace hydrogen sulfide.<sup>5</sup>

6.6 *High-Pressure Sample Container, optional*—Designed for containing liquefied petroleum gas. An inert surface coating on the inner surface of the container may be used, and is recommended.

## 7. Sampling

7.1 Collect a representative sample by Practice D1265 using a clean, high-pressure sample container which is inert to hydrogen sulfide.

7.1.1 As an alternative sampling procedure, connect the test apparatus directly to the sample source and perform the test in the field. This method of direct field measurement eliminates any inaccuracy introduced by the sampling procedure or sample container.

7.2 (**Warning**—Experience has demonstrated that the chemical activity and physical absorptive properties of hydrogen sulfide can cause the concentration of H<sub>2</sub>S in a liquefied petroleum gas sample to be depleted before analysis, even when the sample is contained in scrupulously clean, stainless steel sample cylinders. Conversely, an increase in sulfide concentration has been found to occur when samples of liquefied petroleum gas containing no hydrogen sulfide are

transferred in sample cylinders that have not been properly cleaned and which have been in use with samples containing hydrogen sulfide. This demonstrates that data obtained on samples tested for hydrogen sulfide in the laboratory might not reliably represent the hydrogen sulfide concentration at the sample source.)

## 8. Procedure

8.1 Connect the test apparatus to the sample source by use of a minimum length of high-pressure tubing (see 6.5).

8.2 Flush the line and apparatus for about 1 min.

8.3 Fill the water bath with warm water (50 °C to 70 °C). Tap water is satisfactory.

8.4 By use of the needle valve, adjust the rate of gas flow to 2.3 ± 0.2 L/min.

8.5 Immediately, place a single strip of water-moistened lead acetate paper on the hook in the test cylinder so that the paper is held midway between the watchglass and the bottom of the upper stopper.

8.6 Expose the moist paper for exactly 2 min to the gas flow of 2.3 ± 0.2 L/min. Remove the test paper and proceed as described in Section 10.

## 9. Interpretation of Results

9.1 Compare the exposed test paper with a moist test paper that has not been exposed.

9.1.1 In the absence of distinct coloration, interpret the result as “negative” (“pass”).

9.1.2 If there is a distinct coloration, wait 5 min before interpreting the test results.

<sup>5</sup> Internal surface coatings and coated cylinders are available commercially.

9.1.2.1 If, after 5 min, the paper has no distinct coloration, interpret the result as “negative” (“pass”).

NOTE 1—The presence of methyl mercaptan in liquefied petroleum gas gives a distinct yellow stain to the test paper which, however, fades completely in less than 5 min.

9.1.2.2 If, after 5 min, the paper does have a distinct coloration, interpret the result as “positive” (“fail”).

NOTE 2— Concentrations of H<sub>2</sub>S in LPG of 4 mg/m<sup>3</sup> and higher will give a positive result.

## 10. Report

10.1 Report the result, with reference to this test method, as either “negative” (“pass”) or “positive” (“fail”).

## 11. Precision and Bias

11.1 In the case of pass–fail data or results from other qualitative tests no generally accepted method for determining precision or bias is currently available.

11.2 *Bias*—The procedure in Test Method D2420 for measuring hydrogen sulfide has no bias because the value of hydrogen sulfide is defined only in terms of this test method.

## 12. Keywords

12.1 hydrogen sulfide; lead acetate; liquefied petroleum (LP) gases

## SUMMARY OF CHANGES

Subcommittee D02.H0 has identified the location of selected changes to this standard since the last issue (D2420 – 12a) that may impact the use of this standard. (Approved June 15, 2013.)

(1) Revised 6.4 to allow use of distilled water.

(2) Revised 8.3 and 8.5 to clarify water requirements.

Subcommittee D02.H0 has identified the location of selected changes to this standard since the last issue (D2420 – 12) that may impact the use of this standard. (Approved Nov. 1, 2012.)

(1) Revised 6.3, 6.4, Section 7, 8.1, Section 9, and Fig. 1.

(3) Reorganized 9.1.2 to create 9.1.2.1 and 9.1.2.2.

(2) Added Specification D1193 and Practice D1265 to Referenced Documents, Section 10, 6.5.1, and 6.6.

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